

Wastewater and Treatment

Gretchen Onstad

ENV H 111

11/17/11

Water Security – US EPA

- To assess and reduce consequences, threats, and vulnerabilities to potential **terrorist attacks**
- To plan for and practice response to **natural disasters**, emergencies and incidents
- To develop new security technologies to
 - detect and monitor contaminants
 - prevent security breaches

2002 Water Security Legislation

- Public Health Security and Bioterrorism Preparedness and Response Act (Bioterrorism Act), Title IV requires...
 - US EPA to provide:
 - Information on potential threats to water systems
 - Strategies for responding to potential incidents
 - Information protection protocols for vulnerability assessments in its possession
 - Research studies in areas relevant to water security
 - Community water system serving more than 3,300 persons to:
 - Conduct a vulnerability assessment
 - Submit a certified copy to the EPA Administrator
 - Respond by preparing an emergency response plan
 - Send copy to EPA Administrator within 6 months of completion
 - Full implementation by 2004

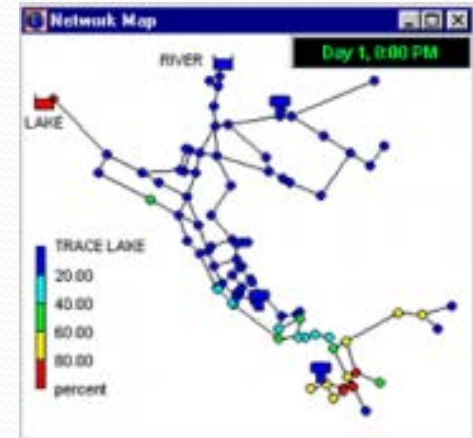
Response to Bioterrorism Act

- Water/Wastewater Agency Response Network
 - State-level organization for mutual aid and assistance to other utilities
- All-hazards approach prepares for bioterrorism and
 - Natural disasters: hurricanes, floods, tornadoes, earthquakes, icestorms, wildfires
 - Supports utility needs in the event of terrorist attack
 - Protection
 - Detection
 - Response
 - Recovery



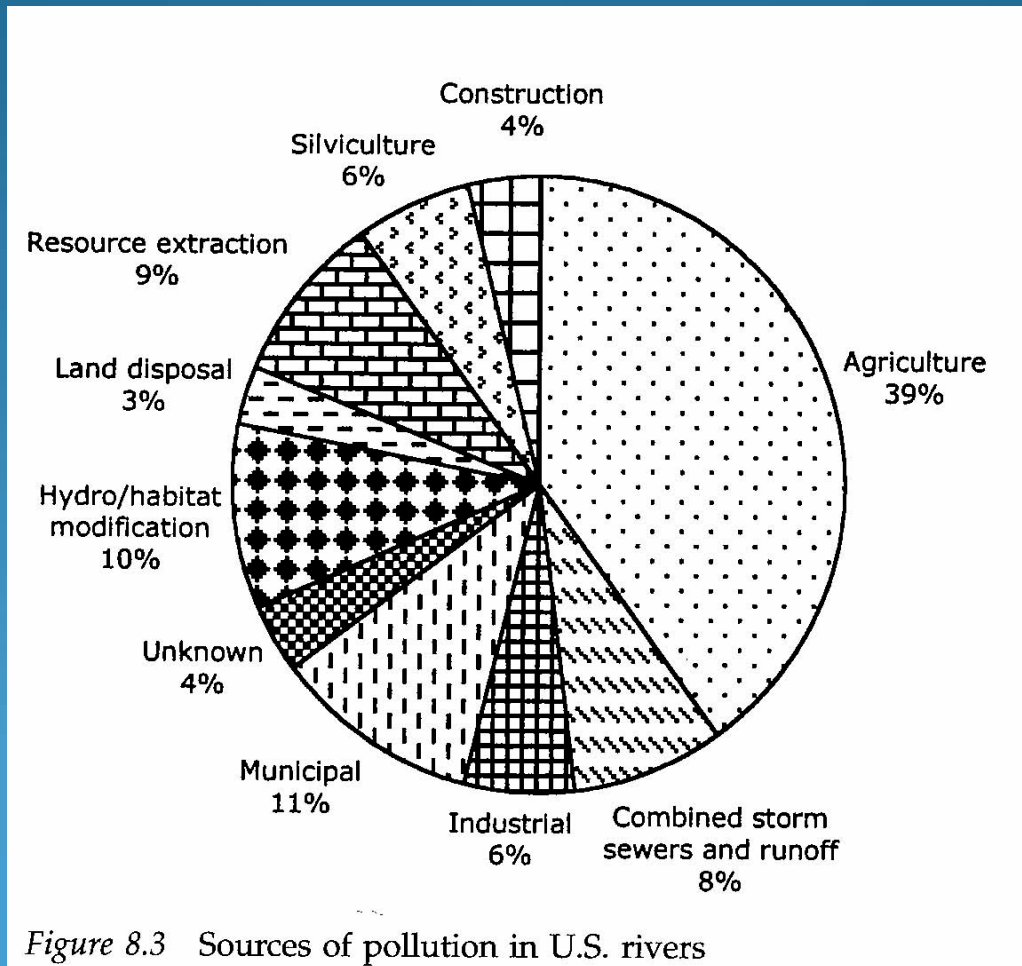
Integrating Water Security and Water Quality Research

- Current Security Research
 1. Online Monitoring
 2. Software
 - Hydraulic & water quality modeling
 3. Policies and Procedures
- Development Needs
 - Real-time microbial monitoring
 - Tools that identify specific contaminants and sources
 - Direct linkage between contamination events and public health outcomes

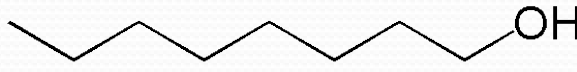


Hydraulic model using
EPANET (www.epa.gov)

Water Contaminants



Water Chemistry & Equilibria

- Salts: $\text{NaCl}_{(s)} + \text{H}_2\text{O} \rightarrow \text{Na}^+ + \text{Cl}^- + \text{H}_2\text{O}$
- Ionic compounds and pH scale (0-14)
 - $\text{CH}_3\text{COOH} \leftrightarrow \text{CH}_3\text{COO}^- + \text{H}^+$
 - Acidity constant: K_a or $\text{p}K_a$
- Partitioning
 - Air-Water: Henry's Constant, $K_H = C_A/C_W$
 - Octanol-Water: K_{OW} 
 - Soil-Water: K_d
 - Biomass-Water: Bioconcentration Factor, BCF
- Solubility: Hydrophilic vs Hydrophobic interactions



Liquid-liquid extraction

Lake Washington

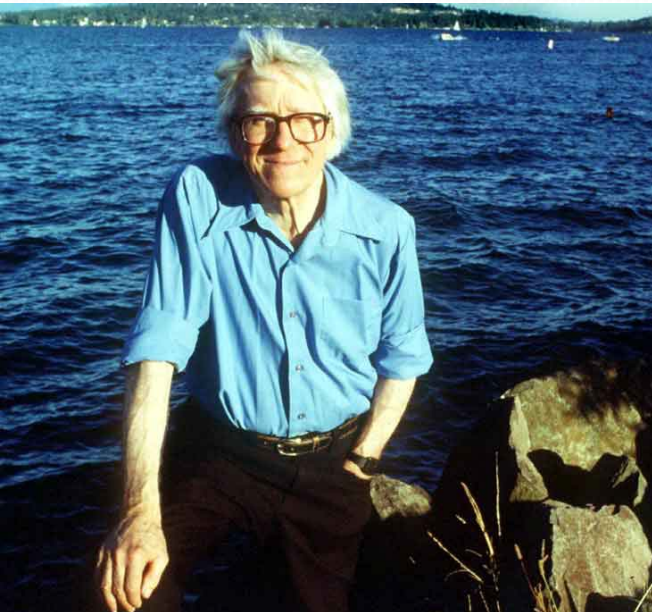
Is dilution the solution to pollution?



**WARNING
POLLUTED WATER
UNSAFE
FOR BATHING**



James Ellis



W. Thomas Edmondson

Rescuing Lake “Stinko” *

- Pollution of Lake and Sound
 - 10 sewage treatment plants dumping effluent into Lake Washington
 - Stormwater caused sewage overflow
 - Untreated sewage to Puget Sound
- 1953 – Jim Ellis, “Father of Metro”
 - Campaign to address problems of population growth & suburbanization
- 1955 – UW Ecology Prof. Edmondson
 - observed increased cyanobacterial algae growth and high phosphorus
 - Predicted eutrophication of Lake WA
 - Shared findings with Ellis and community

* Seattle P.I. (1963)

Aging of a lake

- Caused by algal growth that depletes oxygen, kills fish, and steadily degrades water quality
 - Trophic Stages
 - Oligotrophic
 - Mesotrophic
 - Eutrophic
 - Hypereutrophic
- ↑ Clarity,
Water quality,
Dissolved O₂
- ↓ Nutrients,
Primary productivity,
Algal growth



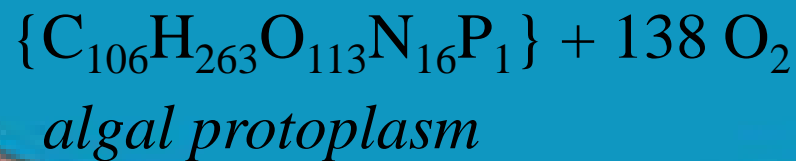
Redfield Equation: Algal growth



Photosynthesis



Respiration



METRO

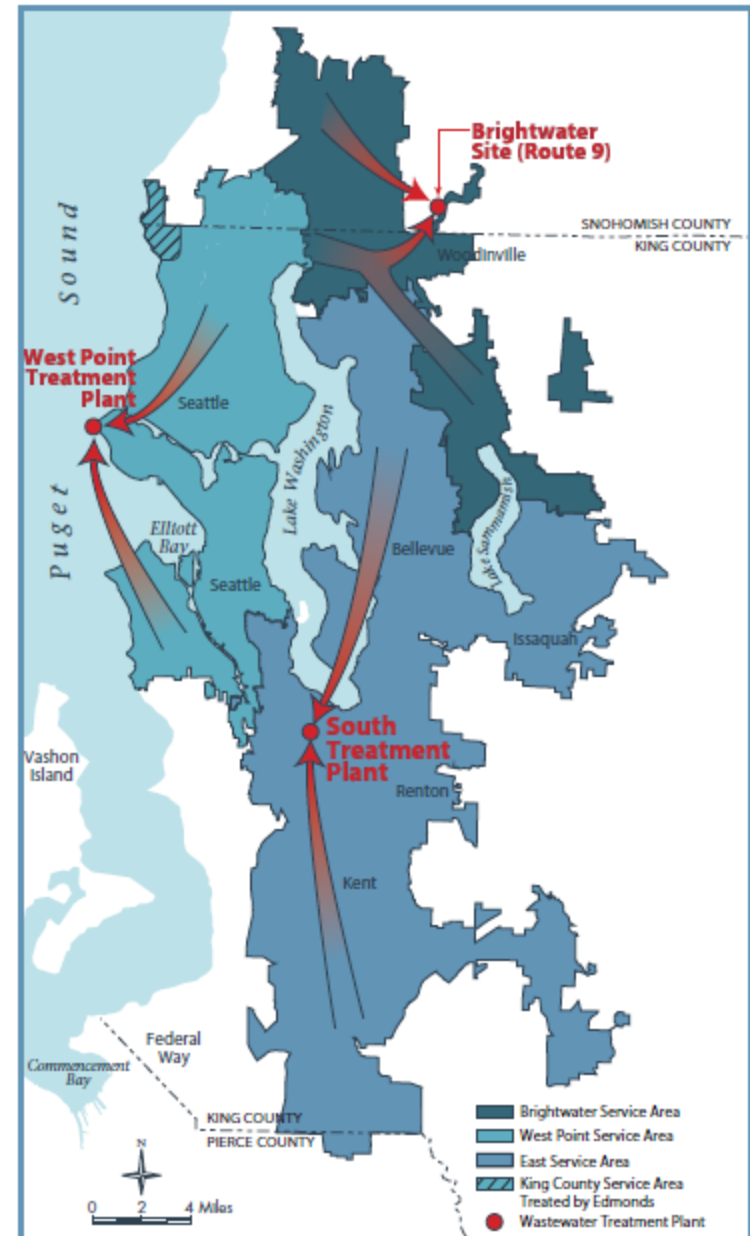
- Municipality of Metropolitan Seattle
- 1958 – voted to authorize Metro to address wastewater treatment for region (completed by 1967)
 - Build sewer system
 - Close sewage treatment plants on Lake Washington
 - Treat sewage entering Puget Sound
- 1960's – Eutrophication of Lake Washington
- 1968 – Improved water clarity & quality: nutrients decline, algal blooms rare, dissolved oxygen increases
- 1972 – Metro takes over transportation
- 1994 – Metro incorporated into King County



Children posing at Matthews Beach in favor of a new METRO to improve Lake Washington water quality, 1958. Seattle Municipal Archive, www.historylink.org

Metro - King County Wastewater Treatment

- 1965: **Renton (South)**
 - Primary + Secondary treatment
 - doubled capacity in 1985
- 1966: **West Point**
 - Primary treatment
 - upgraded with Secondary treatment in 1995
- 2011: **Brightwater**
 - Primary + Secondary
 - opened in Sept



Wastewater Treatment Processes

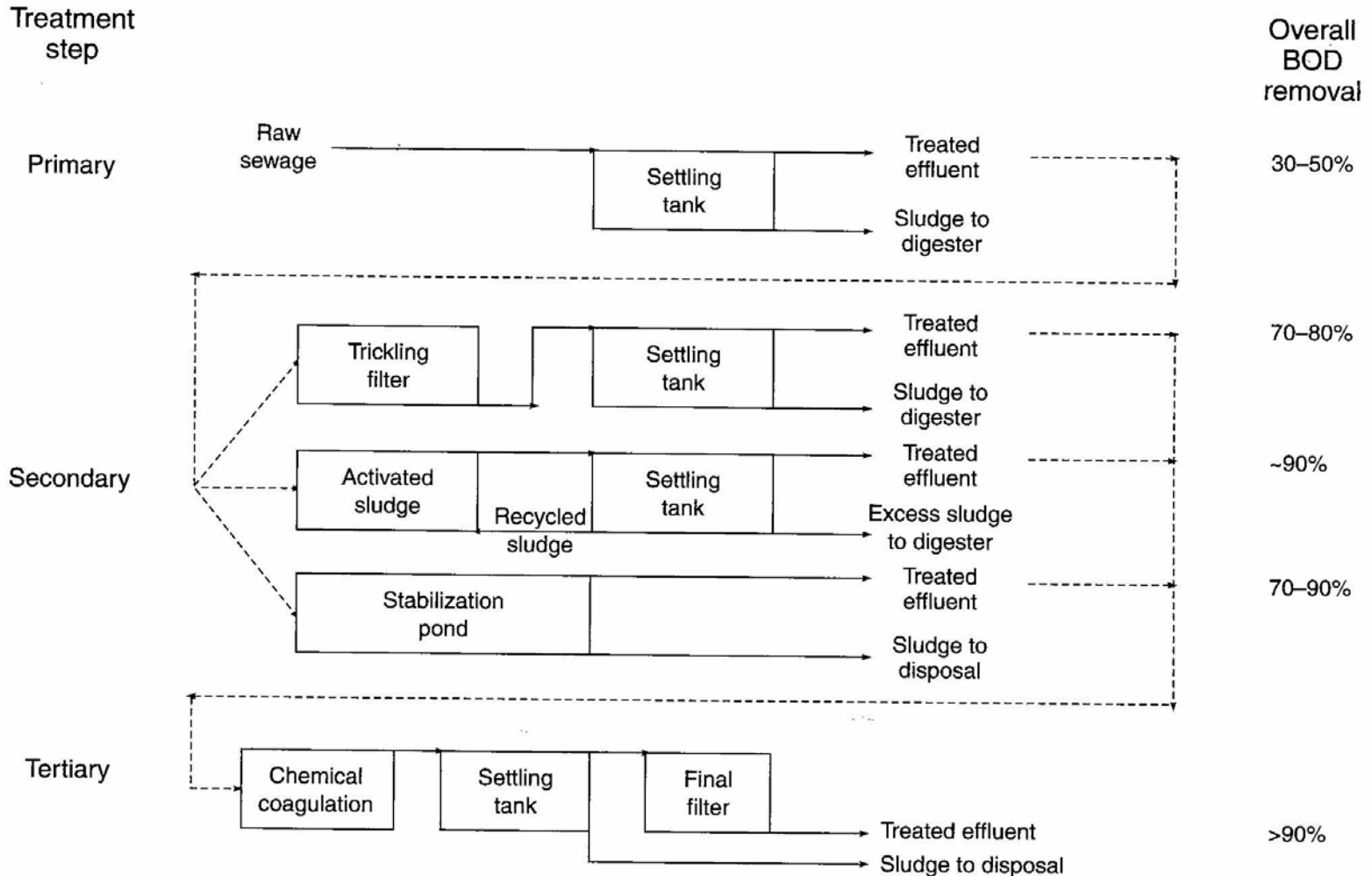


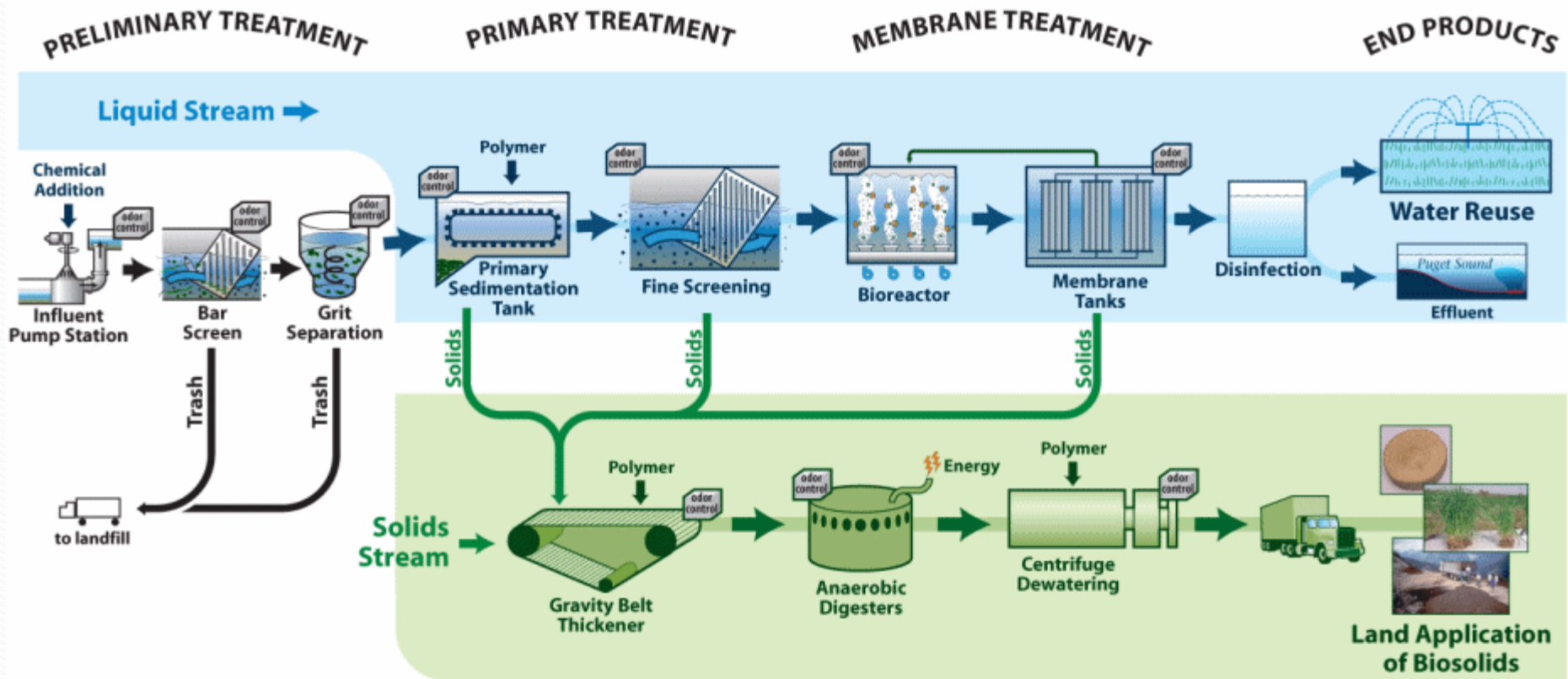
Figure 8.5 Primary, secondary, and tertiary stages in the treatment of municipal sewage

Wastewater Treatment Processes

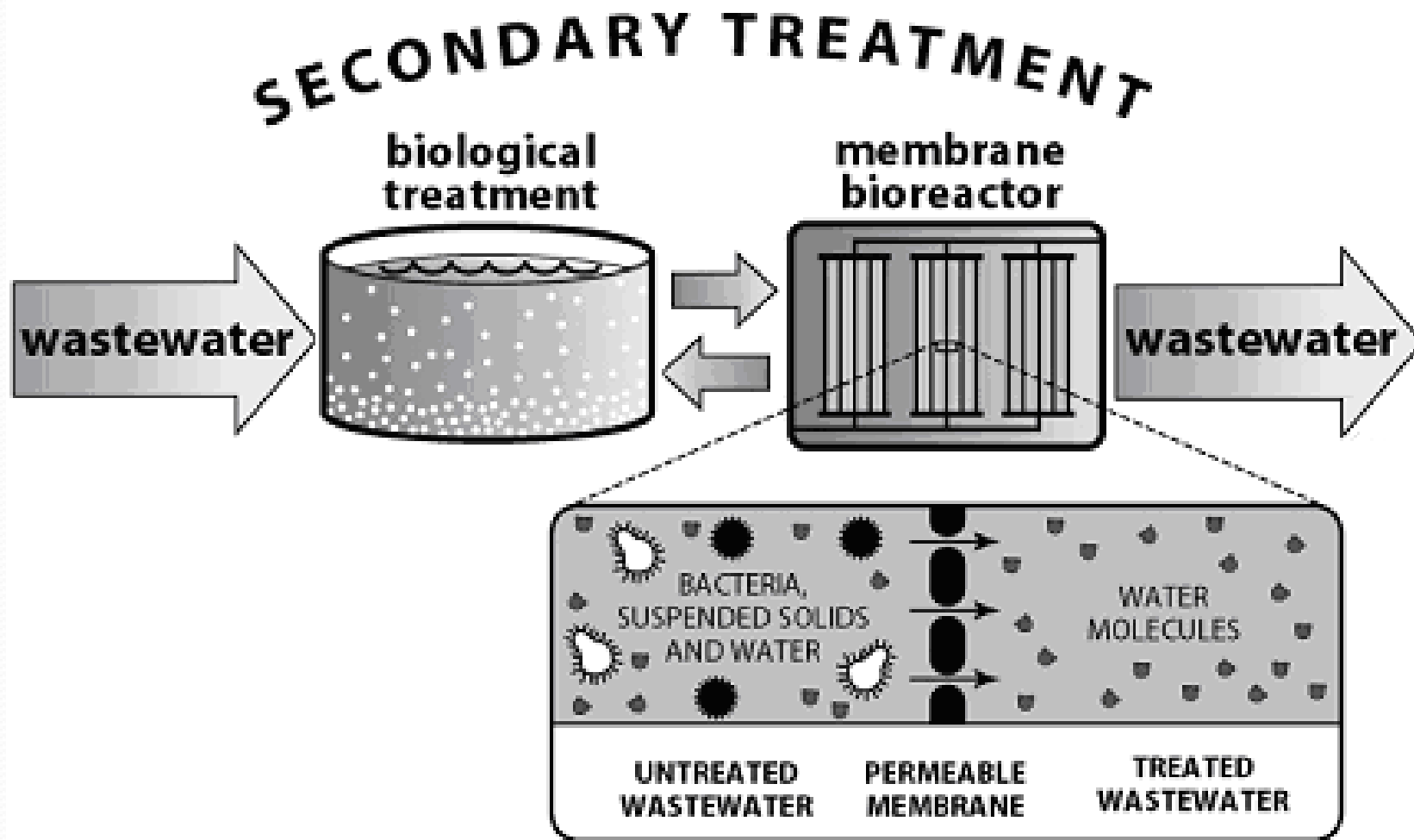
- Primary
 - Settling of solids out of liquid waste
- Secondary
 - Biological oxidation of organic matter in liquid waste
 - Trickling Filter
 - Activated Sludge
 - Stabilization Basin
- Tertiary
 - Process depends on required effluent quality
 - Organic compound removed by granular activated carbon (adsorptive filtration) and ozone disinfection
 - Heavy metals and viruses by coagulation & sedimentation

Brightwater Treatment System

(Secondary)



Brightwater Membrane Bioreactor



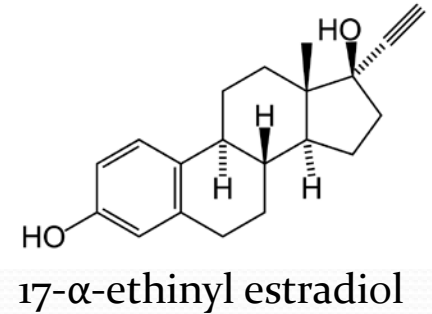
Wastewater Treatment Targets

- Oxygen demand
 - Primary and secondary treatment
- Pathogens
 - Disinfection, digestion
- Nutrients
 - Tertiary treatment
- Chemical toxicants
 - Incidental to other processes
- Industrial waste
 - Pretreatment by industry

Water-Pollution Regulations

- Water Pollution Control Act of 1948 and amendments:
 - 1956 – Standards for interstate and navigable waters
 - 1972 – Secondary treatment required for wastewater
- **Clean Water Act of 1977**
 - Treatment of point sources of industrial waste
 - Standards for discharge into sewers or public waters
 - Required secondary sewage treatment
- Water Quality Act of 1987
 - Control of nonpoint sources of waste
- Wet Weather Control Act of 2000
 - Nonpoint sources of pollution in stormwater

Endocrine disruptors from agriculture & WW effluent



- Endocrine disruptors
 - Pharmaceuticals: ethinyl estradiol, steroid hormones
 - Plasticizers: nonyl phenol
 - Pesticides: atrazine
- Key studies:
 - WW effluent changes gender of fish in British streams
 - Jobling et al (1998) Environ Sci Technol 32:2498-2506
 - USGS study of US streams
 - Kolpin et al (2002) Environ Sci Technol 36:1202-1211
 - Atrazine changes gender of frogs
 - Hayes et al (2010) PNAS 207:10:4612-4617

USGS study: Kolpin et al (2002)

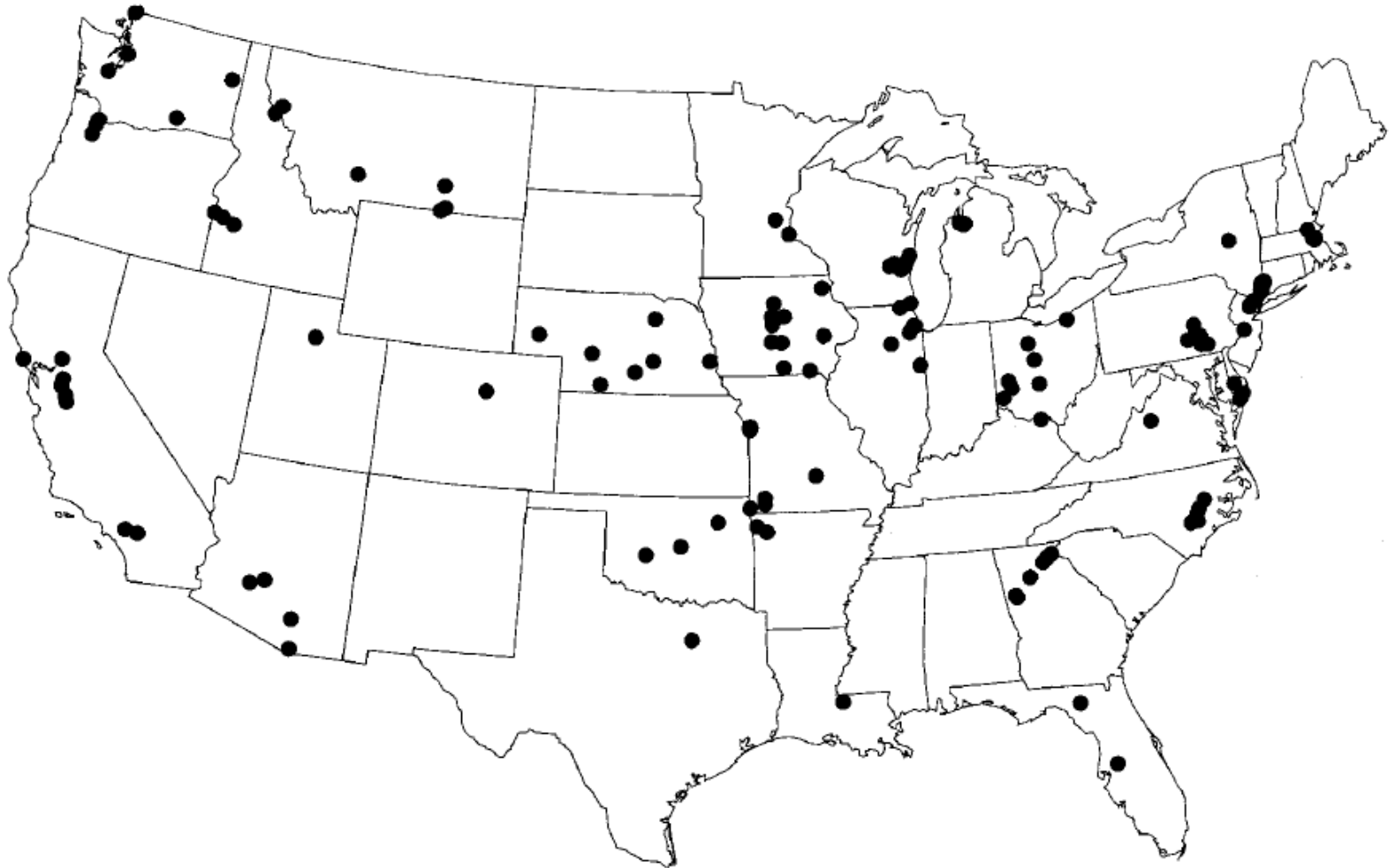


FIGURE 1. Location of 139 stream sampling sites.

USGS study:
Kolpin et al
(2002)

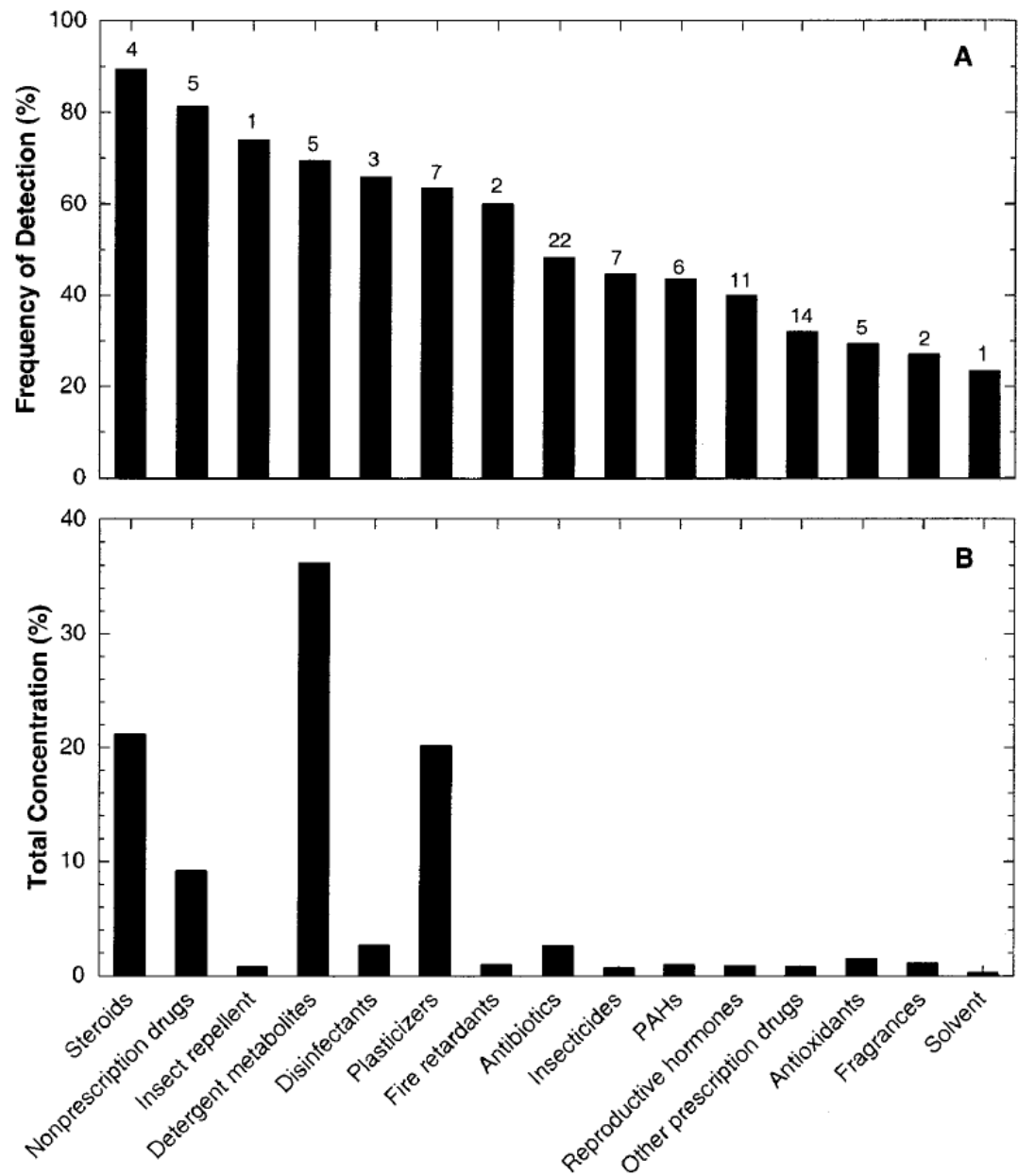
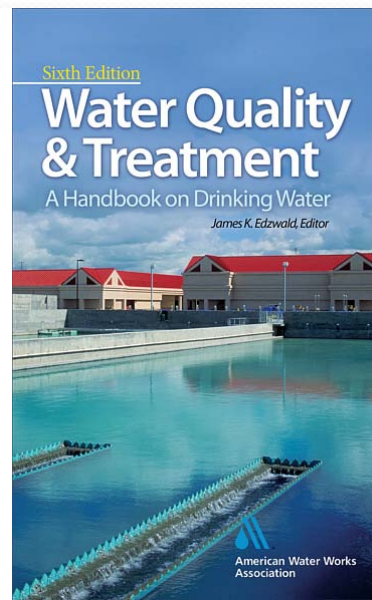
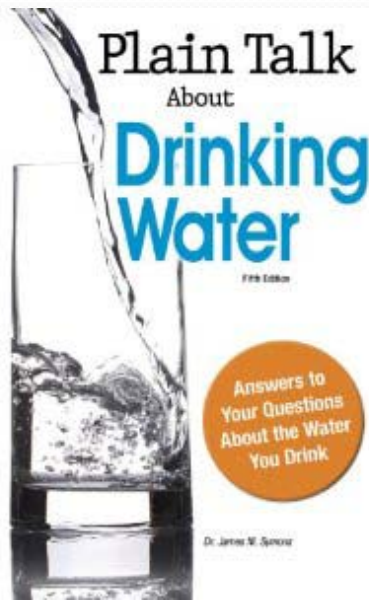


FIGURE 4. Frequency of detection of organic wastewater contaminants by general use category (4A), and percent of total measured concentration of organic wastewater contaminants by general use category (4B). Number of compounds in each category shown above bar.

References

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- Symons, J.M. (2010) *Plain Talk About Drinking Water: Questions and Answers About the Water You Drink*. American Water Works Association
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The Free Online Encyclopedia
of Washington State History



seattle.gov



American Water Works Association
The Authoritative Resource on Safe Water

Portable Water Treatment and Trihalomethane Formation

ENVH 432: Sampling & Analysis II

Environmental Health undergraduates

Data from 2008-2009 courses

Undergraduate Researcher: Benjamin Vellek



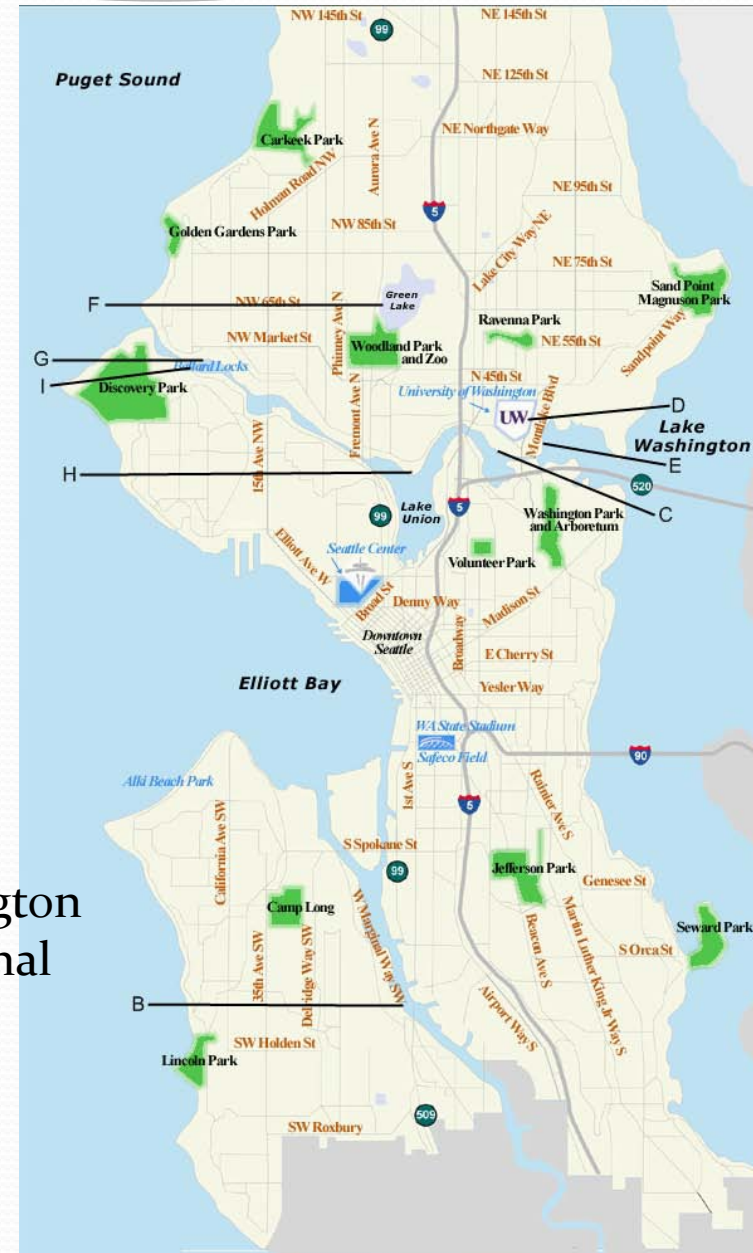
Camping Water Treatment

- Questions:
 - Will treatment techniques exceed maximum contaminant limit (MCL) of 80 ug/L for THM₄?
 - How do treatments compare?
 - What THMs are formed from different source DOC?
 - How does THM formation relate to chlorine consumption?

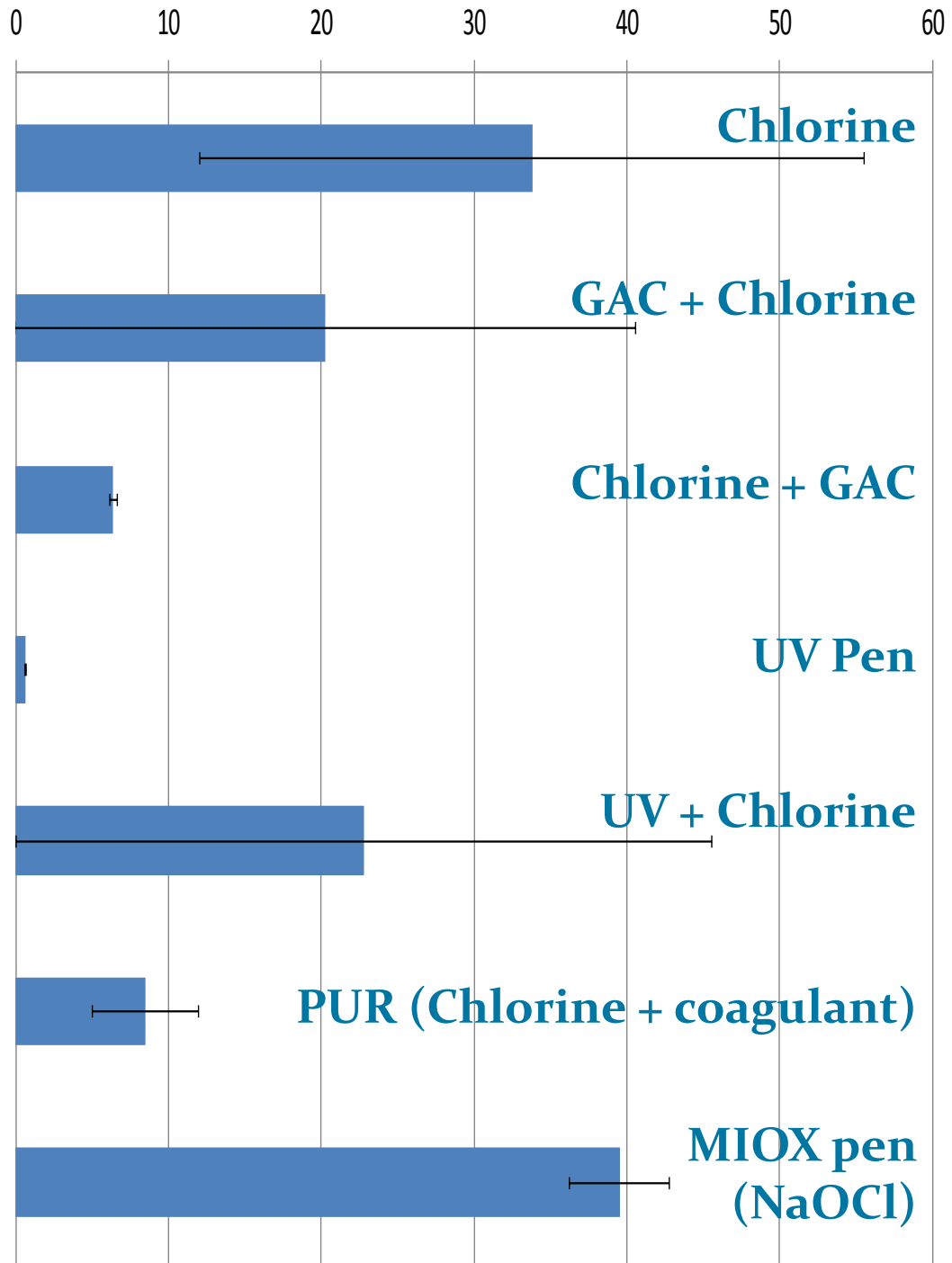
Sampling Locations DOC Sources

Location	Letter
Silver Lake, Everett	A
Duwamish River	B
Portage Bay	C
Drumheller Fountain	D
Union Bay Waterfront Activities Center	E
Green Lake	F
Ballard Locks	G
Gas Works Park	H
Ship Canal Trail	I

Lake
Washington
Ship Canal



Portage Bay: THMs ($\mu\text{g/L}$)





2



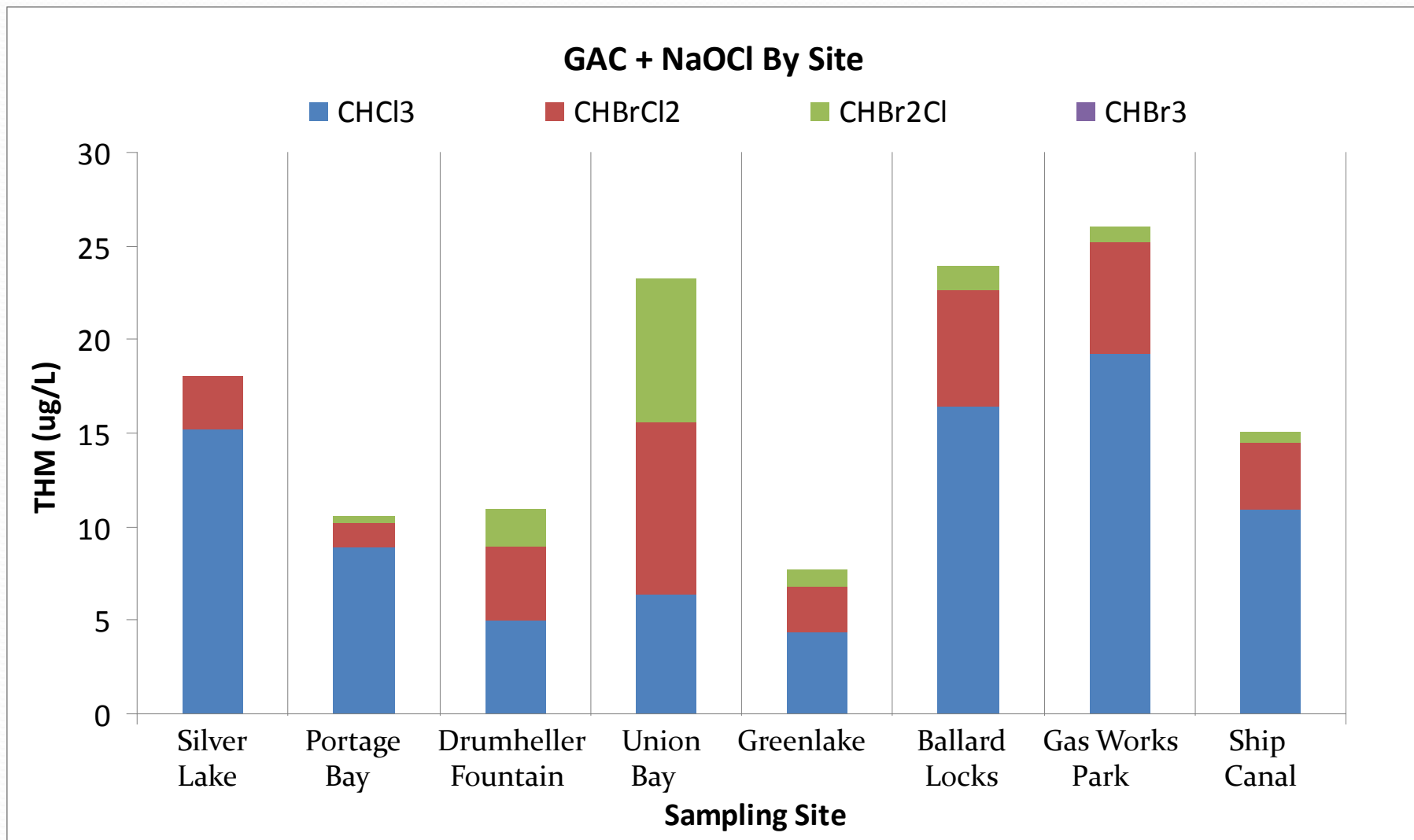
Granular Activated Carbon (GAC)



MSR Sweetwater (NaOCl)

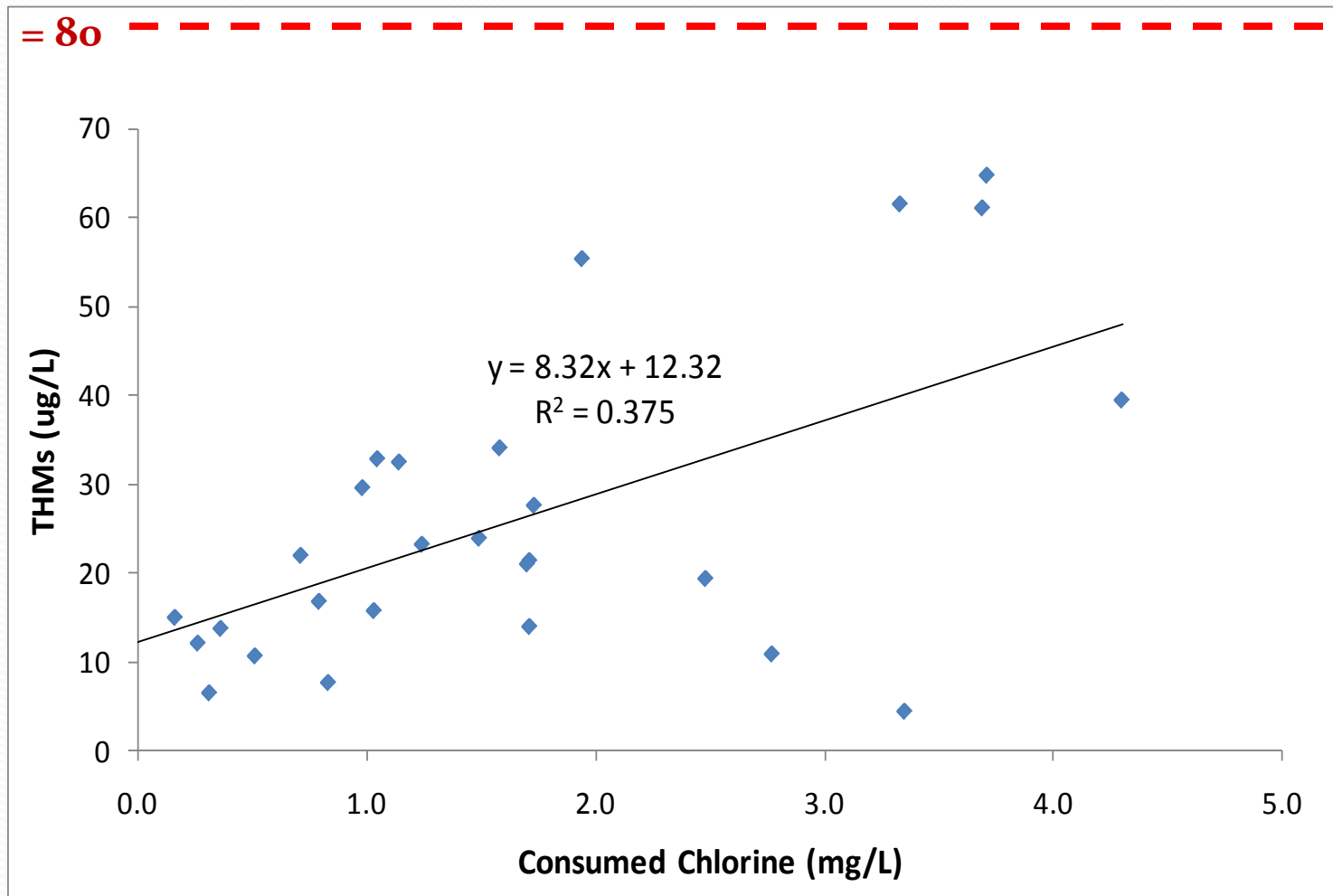
GAC + NaOCl

THM Speciation



THMs and Chlorine Consumption

MCL = 80



Disinfectant	Formula	Regulated DBPs	Target Organism
Chlorine Hypochlorite	Cl ₂ NaOCl	THMs & HAAs	Giardia, most bacteria & viruses
Chlorine Dioxide	ClO ₂	Chlorite Less THMs	Cryptosporidium Giardia, viruses
Chloramines	NH ₂ Cl	Less THMs	Giardia, viruses
Ozone	O ₃	Bromate	Cryptosporidium Giardia, viruses, bacteria
Ultraviolet radiation	UV	none	Cryptosporidium Giardia, viruses, bacteria

Oxidation Kinetics of Freshwater Cyanotoxins

Gretchen Onstad and Urs von Gunten

Swiss Federal Institute of Aquatic Science and Technology,
EAWAG

“TOXIC – Barriers Against Cyanotoxins in Drinking Water”
European Union Water Initiative



Cyanobacteria and -toxins

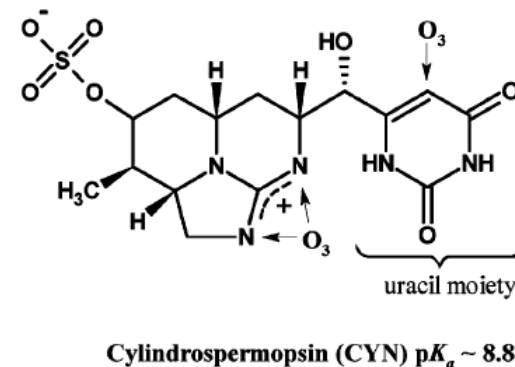
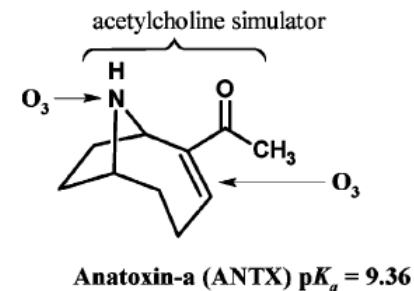
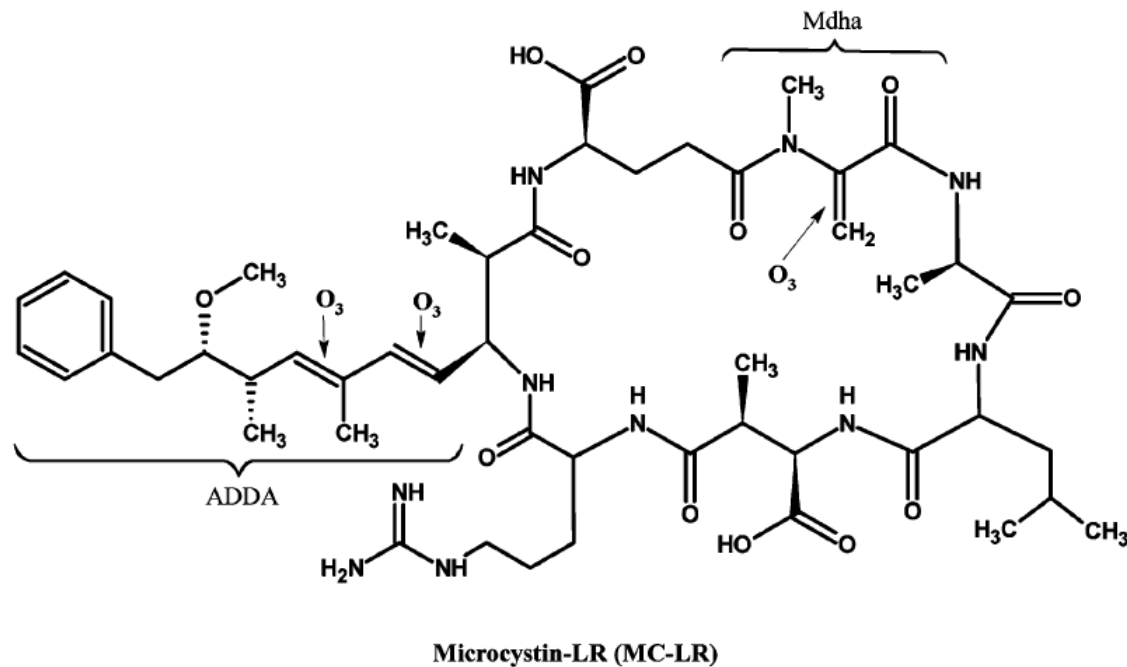
- Blue-green algae = photosynthetic bacteria
- Cyanobacterial “blooms” → eutrophic waters (temp, nutrients, sunlight, stratification)
- Early warning important for water resource protection and Drinking Water Treatment:
 - Removal of cell-bound toxins by filtration
 - Removal of free dissolved toxins
 - Oxidation by ozone or chlorine
 - Adsorption onto GAC or PAC

Toxicity

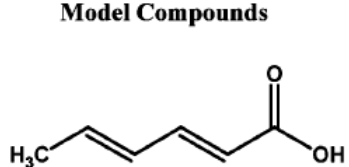
- **Microcystin-LR: Hepatotoxin**
 - Toxicity: side chain **ADDA**
 - Damages **liver** cells → death by fatal circulatory shock or liver failure
 - WHO drinking water guideline **1 μg/L MCLR**
- **Cylindrospermopsin: Hepato-, Cyto-, Genotoxin**
 - Toxicity: **Uracil-RNA or Uracil-DNA**
 - Suppresses protein synthesis or causes DNA strand breakage
 - Target organs: **liver** and **kidneys**
 - Proposed WHO drinking water guideline **1 μg/L CYN**
- **Anatoxin-a: Neurotoxin**
 - Toxicity: **simulates acetylcholine**
 - Overstimulates **respiratory muscles** → death by paralysis
 - NO proposed WHO drinking water guideline for ANTX

Selective Oxidation of Key Functional Groups in Cyanotoxins during Drinking Water Ozonation

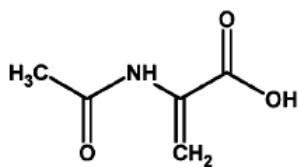
GRETCHEN D. ONSTAD,^{‡,†}
 SABINE STRAUCH,[‡] JUSSI MERILUOTO,[§]
 GEOFFREY A. CODD,^{||} AND
 URS VON GUNTEN^{*,‡,‡}



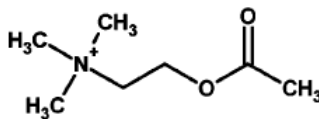
Model Compounds



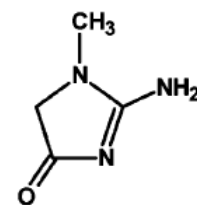
Sorbic Acid



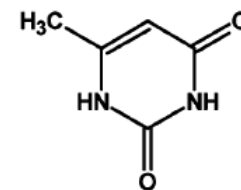
2-Acetamidoacrylic acid



Acetylcholine



Creatinine



6-Methyluracil

FIGURE 1. Structures of target cyanotoxins, model compounds, and proposed sites of ozone attack.

pH Dependence

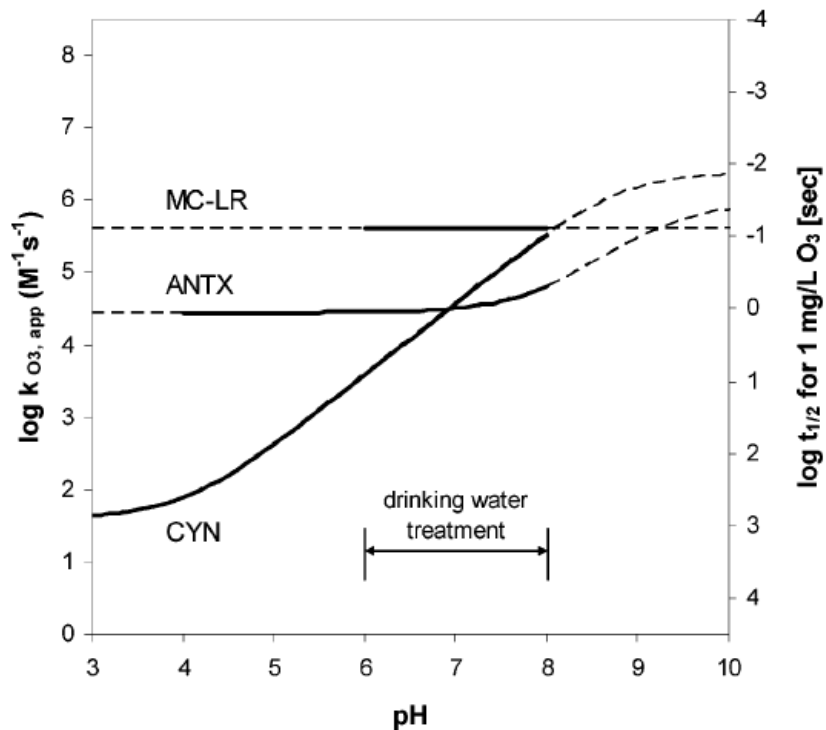


FIGURE 4. pH dependence of the rate constants for the reaction of cyanotoxins with ozone. Half-life times are given for an ozone concentration of 1 mg/L: MC-LR, microcystin-LR; CYN, cylindrospermopsin; ANTX, anatoxin-a.

DBP formation

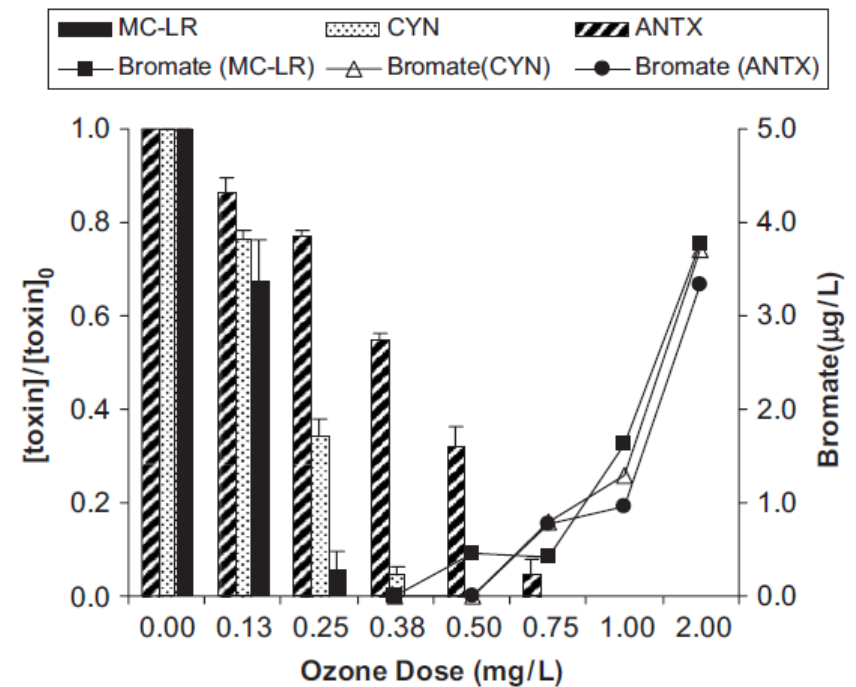


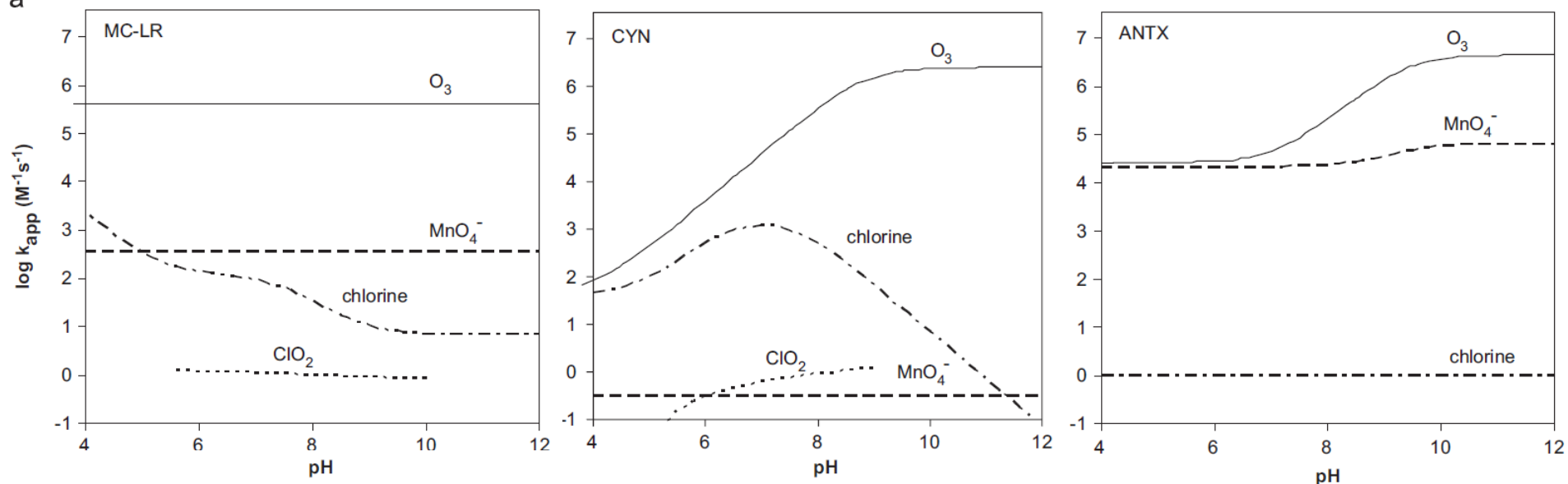
Fig. 5 - Relative decrease of cyanotoxin and increase of bromate formation as a function of the ozone dose. Error bars indicate standard deviation. Experimental conditions: pH 8, 20 °C, 3.6 mg/L DOC, 100 $\mu\text{g/L}$ ammonia, 3.6 mM alkalinity as HCO_3^- and 50 $\mu\text{g/L}$ Br^- .

Oxidative elimination of cyanotoxins: Comparison of ozone, chlorine, chlorine dioxide and permanganate



Eva Rodríguez^a, Gretchen D. Onstad^{b,1}, Tomas P.J. Kull^c, James S. Metcalfe^d,
 Juan L. Acero^{a,*}, Urs von Gunten^{b,e,**} WATER RESEARCH 41 (2007) 3381–3393

a



pH 7 {

- MCLR: $O_3 > MnO_4^- > HOCl > ClO_2 > NH_2Cl$
- CYN: $O_3 > HOCl \gg ClO_2 > MnO_4^- \approx NH_2Cl$
- ANTX: $O_3 \approx MnO_4^- \gg ClO_2 > HOCl > NH_2Cl$

TOXIC project outcomes

- Cyanotoxin removal by water treatment efficient if:
 - raw water management and early warning systems
 - application of precise and optimized analytical techniques
 - optimized treatment trains
- 2009 CCL₃ (US)
 - These 3 cyanotoxins for prioritized research (among 104 chemical contaminants)