

## General and Dilution Ventilation

### Dilution Ventilation

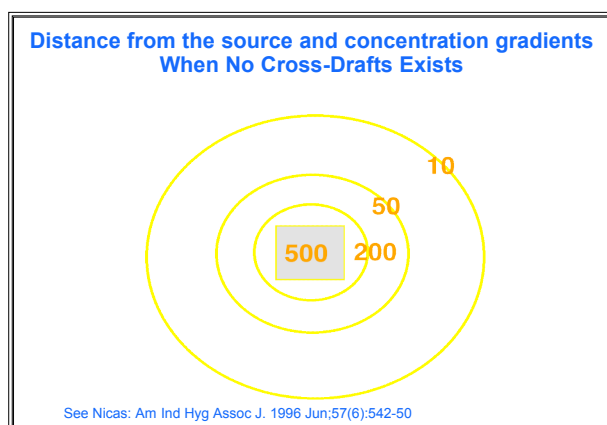
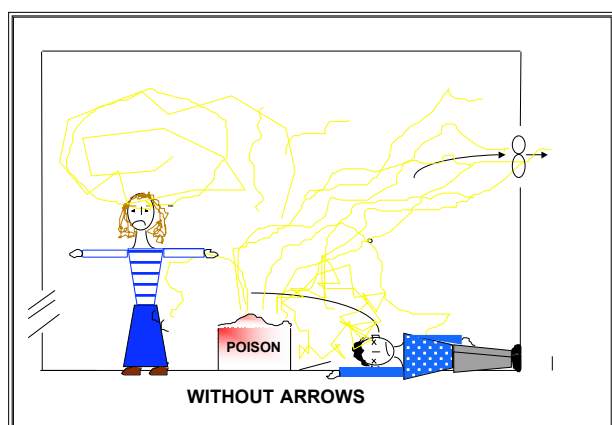
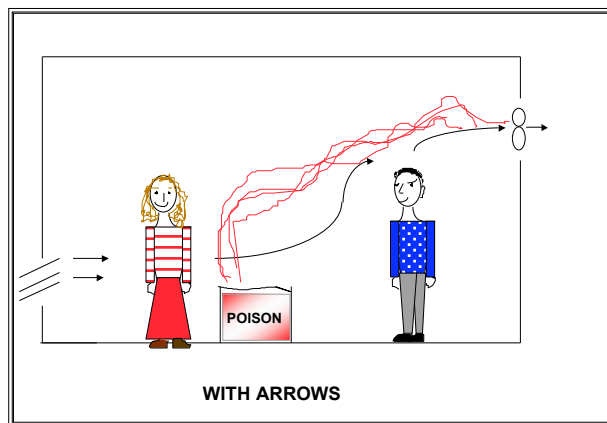
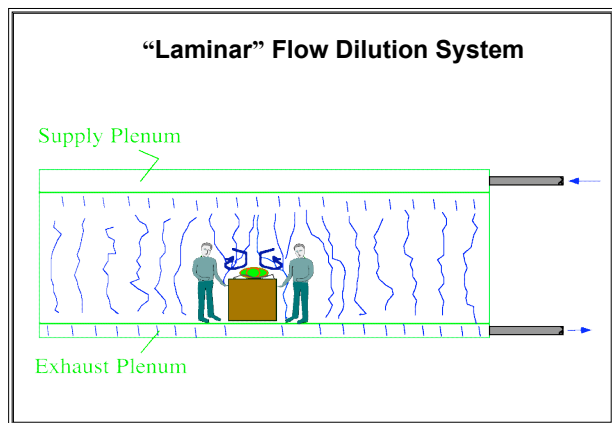
- ❑ The solution to pollution is dilution?
- ❑ Do you want to move a lot of air?
- ❑ What happens in the winter?
- ❑ How do you get a sweeping effect?
- ❑ Why bother with local exhaust if there are too many sources to vent them all?
- ❑ To have effective DV we need to:
  - ❑ Mix contaminated air with large volume of fresh air
  - ❑ Have sufficient air changes/hour to prevent build-up
  - ❑ Create air movement and mixing at all required locations

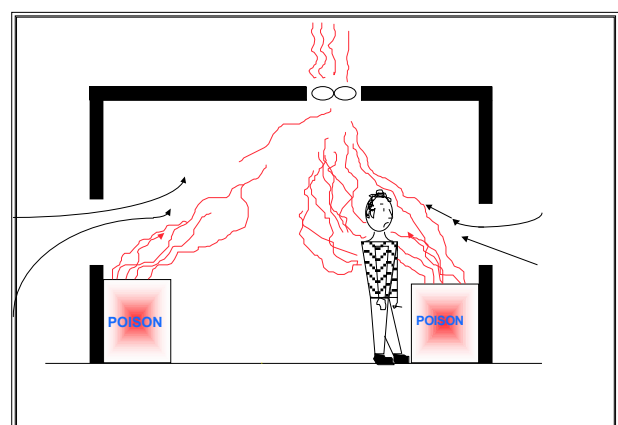
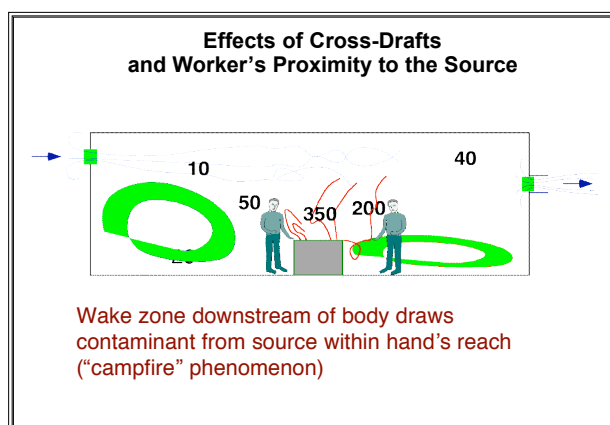
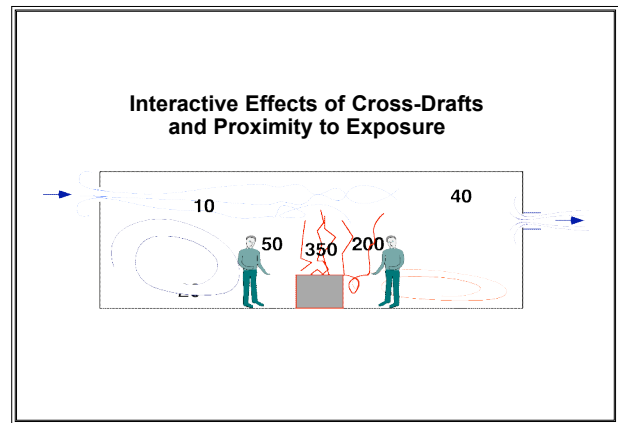
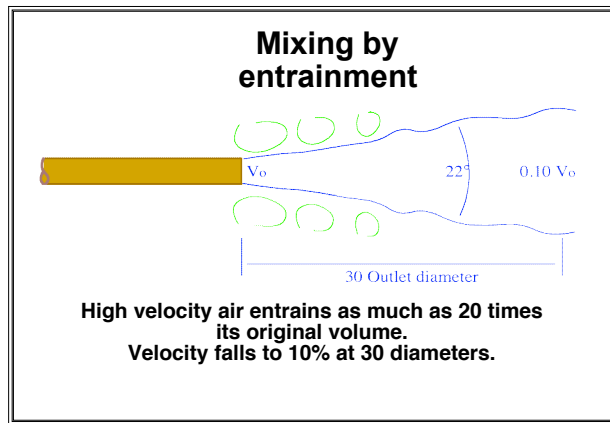
### General Ventilation - Purpose

- ❑ General ventilation
  - ❑ Provide heating or cooling
  - ❑ Provide make-up air
  - ❑ Provide dilution and reduction of contaminants such as CO<sub>2</sub> and body odor
- ❑ Dilution ventilation
  - ❑ Provide dilution of contaminants to safe levels (<TLV or LEL)
  - ❑ Constrained by comfort and other factors
  - ❑ Usually initial cost: DV cost << LEV cost
  - ❑ Usually for operation: DV cost >> LEV cost

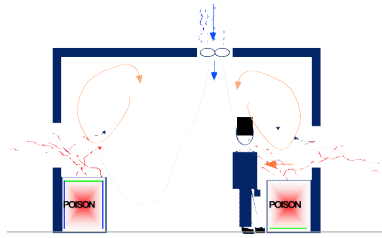
### Dilution Ventilation - applications

- ❑ Toxicity of contaminant is low to moderate (High TLV)
- ❑ Velocity and generation rate of contaminant low to moderate – must consider periodic generation too
- ❑ Sources are not well localized or identifiable
- ❑ Mobile sources or variable work process
- ❑ Energy costs are not a significant concern

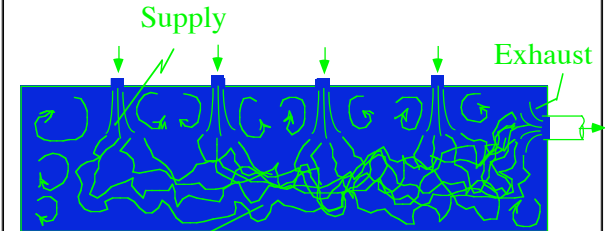




### Reverse the flow?



### Zones of Ventilation



General Elevation

Blowing with entrainment produces large scale eddies & mixing

### Computing Generation Rate

- We are concerned about dynamic conditions
- Assume constant generation rate

$$G = \frac{d(Vol_{vapor})}{dt}$$

$$G = \frac{Amount\ Evaporated}{t_2 - t_1}$$

### Calculating dilution volumes

$$VolumeCont = \frac{MassOfTheLiquid}{MassForOneMole} * VolumeForOneMole$$

$$VaporVol = \frac{MassOfTheLiquid}{MW} * 24.04\ L * \left( \frac{273.15C + T}{293.15} \right) \left( \frac{760\ mmHgO}{P_{atm}} \right)$$

$$VaporVol = \frac{(sp.grav. * \rho_{H2O} * Vol_{liquid})}{MW} * 24.04\ L * \left( \frac{273.15C + T}{293.15} \right) \left( \frac{760\ mmHgO}{P_{atm}} \right)$$

### Target Concentration

Toxicity	TLV ppm	C <sub>i</sub> as a % TLV
highly toxic, radioactive or carcinogenic	< 20	local exhaust only
moderately toxic	20-100	25
somewhat toxic	100-200	50
slightly toxic	> 200	75

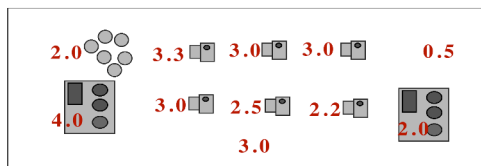
### Concentration if perfect mixing

$$C = \frac{G}{Q}$$

Concentration if not perfect mixing

$$C = \frac{G}{Q/m}$$

### Values of m<sub>i</sub> for each work station



$$m_i = \frac{C_i}{C_{exhaust}}$$

m due to non-uniformity  
Supply air mixing  
Contaminant release  
Distance from worker

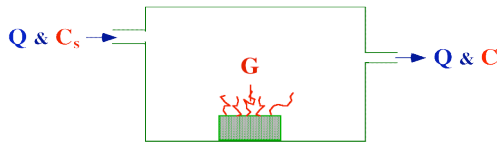
### Mixing factors

$$m_i = \frac{C_i}{C_{exhaust}}$$

$$C_{avg} = \frac{1}{8 \text{ hours}} \sum_i C_i t_i \quad C_{avg} = \frac{C_{exhaust}}{8 \text{ hours}} \sum_i m_i t_i$$

$$M_{avg} = \frac{1}{8 \text{ hrs}} \sum_i m_i t_i \quad M_{peak} = \frac{\bar{C}_{15 \text{ min}}}{C_{exhaust}}$$

### Accumulation = Generation – Removal



$$V \frac{dC}{dt} = \left( G + \frac{Q}{m} C_s - \frac{Q}{m} C_t \right)$$

$$C_2 = C_1 e^{-Qt/mV} + \left( \frac{mG}{Q} + C_s \right) (1 - e^{-Qt/mV})$$

### ROOM VOLUME

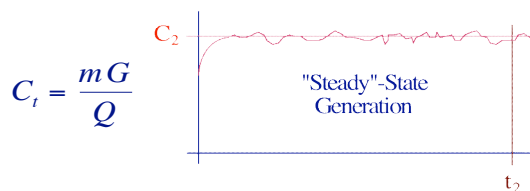
$$C_2 = C_1 e^{-Qt/mV} + \left( \frac{mG}{Q} + C_s \right) (1 - e^{-Qt/mV})$$

Important for transient conditions  
Irrelevant for steady state

$$C_t = \frac{mG}{Q}$$

### Application of Equations

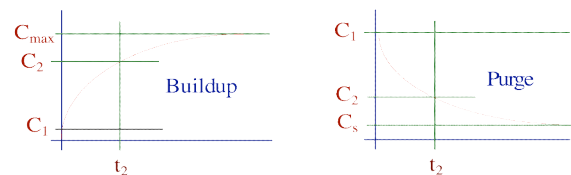
$$C_2 = C_1 e^{-Qt/mV} + \left( \frac{mG}{Q} + C_s \right) (1 - e^{-Qt/mV})$$



$$C_t = \frac{mG}{Q}$$

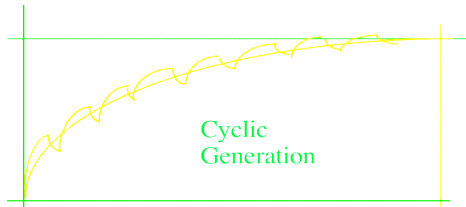
### Application of Equations

$$C_2 = C_1 e^{-Qt/mV} + \left( \frac{mG}{Q} + C_s \right) (1 - e^{-Qt/mV})$$



### Cyclic operations

- If short cycles, effects “average out”



### When conditions change with time

- Solve for time interval during which all conditions are constant
- If conditions change continuously, make interval one minute
- Use result as initial conditions for next interval

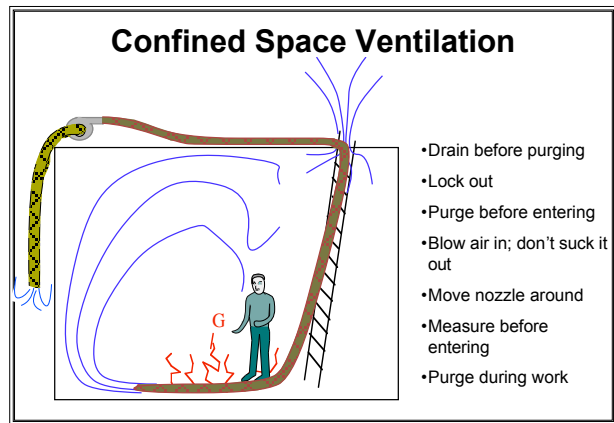
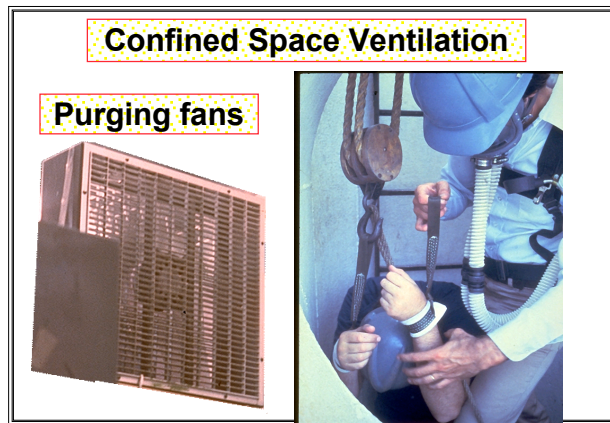
$$C_2 = C_1 e^{-Qt/mV} + \left( \frac{mG}{Q} + C_s \right) (1 - e^{-Qt/mV})$$

### DESIGNING NEW SYSTEM

- Locate sources near exhaust fans**
- Locate supply air outlets to direct air away from face and towards exhaust fans
  - Separate sources from traffic using barriers
  - Block undesirable cross drafts and competing sources of motion using barriers.

### EXISTING SYSTEM IMPROVEMENTS

- Substitute less volatile or toxic chemicals
- Install or improve local exhaust hoods
- Reduce incidence of spills and leaks
- Relocate supply and exhaust points
- Relocate workers or the sources — or both
- Increase airflow



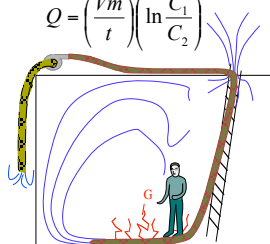
## Confined Space Ventilation-Purge Time

If supply air clean, then:

$$t = \left( \frac{Vm}{Q} \right) \left( \ln \frac{C_1}{C_2} \right)$$

$$Q = \left( \frac{Vm}{t} \right) \left( \ln \frac{C_1}{C_2} \right)$$

- Should be computed
- Then test against measurements
- Wait minimum time even if measurement says okay
- Time less with bigger blower
- Bigger blower more costly, harder to maneuver



### Example Problem

Initial measurements indicate 10,000 ppm of xylene in a confined space. Assuming that  $C_t = 0.25 * TLV$ , how much should  $Q$  be to allow entry in 30 minutes if:

$R=1000 \text{ ft}^3, M=3, C_s = 0$

$$Q = \left( \frac{Rm}{\Delta t} \right) \ln \left[ \frac{G + QC_s / m - Q C_0 / m}{G + QC_s / m - Q C_2 / m} \right] = \left( \frac{1000 \text{ ft}^3 * 3}{30 \text{ min}} \right) \ln \left[ \frac{0.010^{-2}}{0.025 * 10^{-6}} \right] = 599 \text{ ft}^3/\text{min}$$

b.	R=2000 ft <sup>3</sup> , M=3, C <sub>s</sub> = 0 :	Solution: Q = 1198 ft <sup>3</sup> /min
c.	R=1000 ft <sup>3</sup> , M=6, C <sub>s</sub> = 0 :	Solution: Q = 1198 ft <sup>3</sup> /min
d.	R=1000 ft <sup>3</sup> , M=6, C <sub>s</sub> = 15 ppm:	Solution: Q = 1381 ft <sup>3</sup> /min



## Summary

- Estimating G and m is difficult
- Reduce G as much as possible
- Reduce greatest contributors to exposure and perceived exposure first
- Use sweeping, but be realistic about it
- Complement local exhaust systems
- Provide winter and summer
- Purge before and during confined space entry

**STOP HERE**