

Ann Mescher

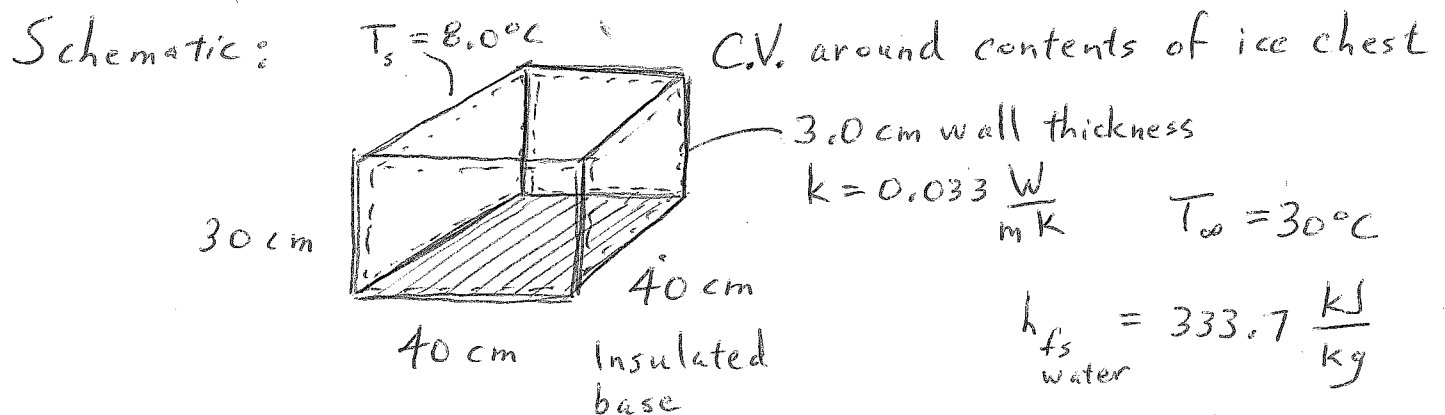
March 30, 2017

# HW Assignment #1

## Problem #1

Statement: Ice chest ( $30\text{ cm} \times 40\text{ cm} \times 40\text{ cm}$ ) of wall thickness  $3.0\text{ cm}$  is filled with  $40\text{ kg H}_2\text{O}$ , initially solid ice at  $0.0^\circ\text{C}$ . Outer surface temperature of the ice chest is  $8.0^\circ\text{C}$ , and the  $40\text{ cm} \times 40\text{ cm}$  base is insulated.

Find: Time for ice to completely melt.



Assumptions: Inner surfaces of ice chest are at uniform temperature  $0.0^\circ\text{C}$ . Outer surfaces of top and four side walls are at uniform temperature  $8.0^\circ\text{C}$ . 1-D conduction is assumed through these five walls. Base of ice chest is insulated. No sensible energy change.

Analysis: Over a period of time  $\Delta t$ ,

$$E_{\text{in}} - E_{\text{out}} + E_g = \Delta E_{\text{stored}}$$

Energy flows (No thermal generation  $\dot{q}$ )  
from ambient (within  $\text{H}_2\text{O}$ )  
environment, through  
ice chest walls to ice  
chest contents.

$$E_{in} = q \Delta t = q'' A \Delta t = m_{H_2O} h_{fg, H_2O} = \Delta E_{stored}$$

Where  $q'' = -k \frac{\Delta T}{\Delta x} = -0.033 \frac{W}{mK} \cdot \frac{(0.0 - 8.0)K}{0.03m} = 8.8 \frac{W}{m^2}$

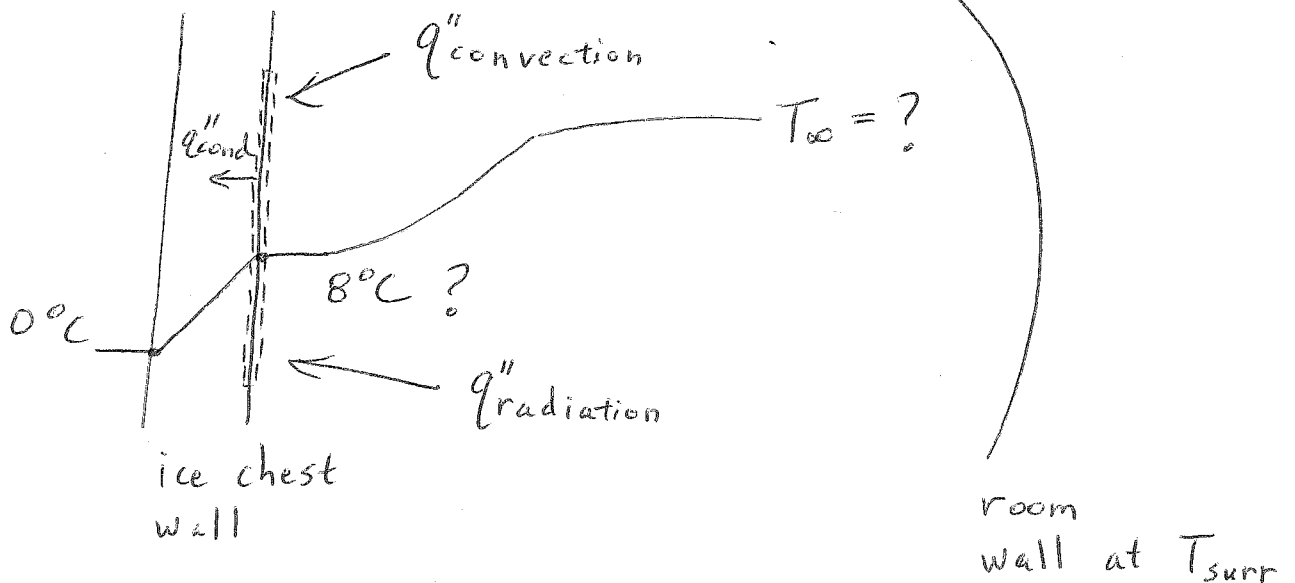
$$A = 4(0.37m \cdot 0.27m) + 0.37m \cdot 0.37m = 0.5365 m^2$$

$$\Delta t = \frac{m_{H_2O} h_{fg, H_2O}}{q'' A} = \frac{40kg \cdot 333.7 \frac{kJ}{kg}}{8.8 \times 10^{-3} \frac{kW}{m^2} \cdot 0.5365 m^2} = 2.827 \times 10^6 \text{ seconds}$$

$$\Delta t = 785 \text{ hours} = 32.7 \text{ days}$$

Comments: This seems a long time!

Is it realistic that the outer surface could be maintained at 8.0°C? It depends on the surrounding wall and air temperatures. Perform surface energy balance.



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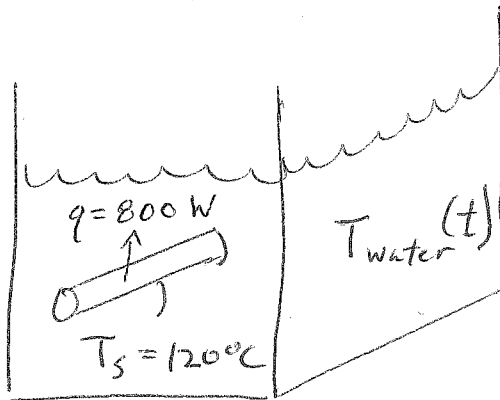
## HW Assignment #1

### Problem #2

Statement: 800 W electric resistance heating element (cylindrical with  $L = 50$  cm,  $D = 0.50$  cm) has constant surface temperature  $T_s = 120^\circ\text{C}$  and is immersed in 40 kg of water at an initial temperature of  $20^\circ\text{C}$ .

Find: How long to raise the temperature of the water from  $20^\circ\text{C}$  to  $80^\circ\text{C}$ , and the initial and final heat transfer coefficients?

Schematic:



Control Volume is around the water, excluding the electric resistance heater.

Assumptions: The tank is insulated, so that there is negligible heat loss from the water to surroundings. Water circulates so that the water temperature is approximately uniform. The specific heat of the liquid water from  $20^\circ\text{C}$  to  $80^\circ\text{C}$  is treated as constant. Negligible radiation transfer.

Problem #2 Continued

$$\text{Analysis: } \dot{E}_{in} - \dot{E}_{out} + \dot{E}_g = \dot{E}_{stored} \quad \text{for CV surrounding water}$$

no heat loss from tank  
no thermal generation in water

$$q = 800 \text{ W} = m_w c_w \frac{dT_w}{dt} = 40 \text{ kg} \cdot 4181 \frac{\text{J}}{\text{kg K}} \frac{dT_w}{dt}$$

where the specific heat of water  $c_w = 4181 \frac{\text{J}}{\text{kg K}}$  is evaluated at the average water temperature of  $50^\circ\text{C} = 323 \text{ K}$ .

Over a time interval  $\Delta t$ ,  $E_{in} = \Delta E_{stored}$

$$q \Delta t = 800 \text{ W} \cdot \Delta t = m_w c_w \Delta T_w = 40 \text{ kg} \cdot 4181 \frac{\text{J}}{\text{kg K}} (80 - 20) \text{ K}$$

Answer:  $\Delta t = 12,543 \text{ seconds} = \boxed{3.48 \text{ hours}}$

At the initial water temperature,  $T_\infty = 20^\circ\text{C}$ ,

$$q = h A (T_s - T_\infty) \quad \text{where } A = 2\pi r^2 + 2\pi r L$$

$$h = \frac{800 \text{ W}}{(120 - 20) \text{ K} \cdot \pi \cdot 0.005 \text{ m} \left(0.50 \text{ m} + \frac{0.005 \text{ m}}{2}\right)} = \frac{\pi D^2}{2} + \pi D L$$

Answer:  $h = 1013.5 \frac{\text{W}}{\text{m}^2 \text{K}}$   $\boxed{h_{\text{initial}} = 1010 \frac{\text{W}}{\text{m}^2 \text{K}}}$

At the final water temperature,  $T_\infty = 80^\circ\text{C}$ ,

Answer:  $h = \frac{800 \text{ W}}{(120 - 80) \text{ K} \cdot A} = 2533.8 \frac{\text{W}}{\text{m}^2 \text{K}}$   $\boxed{h_{\text{final}} = 2530 \frac{\text{W}}{\text{m}^2 \text{K}}}$

Comments: In this transient process it is impossible for  $T_s$  to be maintained at  $120^\circ\text{C}$  for a constant  $q = 800 \text{ W}$  from the electric heater. For forced or natural convection,  $h_{\text{initial}} > h_{\text{final}}$ ; therefore  $(T_s - T_\infty)_{\text{initial}} < (T_s - T_\infty)_{\text{final}}$ . Since  $T_\infty$  is increasing,  $T_s$  must also increase.

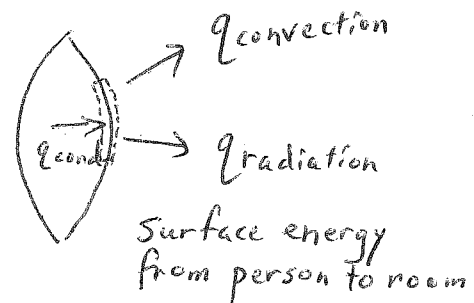
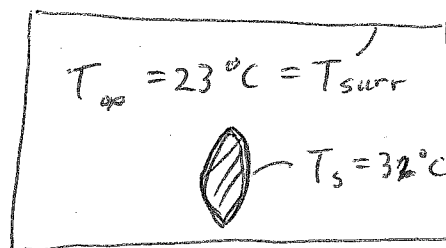
## HW Assignment #1

## Problem #3

Statement: Person of surface area  $1.7 \text{ m}^2$  at surface temperature of  $32^\circ\text{C}$  with gray emissivity of  $\epsilon = 0.90$  standing in a large room enclosure at  $23^\circ\text{C}$  with room air at the same temperature of  $23^\circ\text{C}$  with a natural convection coefficient of  $h = 5.0 \frac{\text{W}}{\text{m}^2\text{K}}$ .

Find: Convective and radiative heat losses from person.

Schematic:



Assumptions: For gray surface,  $\epsilon_{\text{person}} = 0.90$ , in a

large enclosure,  $q_{\text{radiation}} = \epsilon \sigma A (T_s^4 - T_{\text{surr}}^4)$

Analysis:  $q_{\text{radiation}} = 0.90 \cdot 5.67 \times 10^{-8} \frac{\text{W}}{\text{m}^2\text{K}^4} \cdot 1.7 \text{ m}^2 (305.15 \text{ K}^4 - 296.15 \text{ K}^4)$

$q_{\text{convection}} = hA(T_s - T_{\infty}) = 5.0 \frac{\text{W}}{\text{m}^2\text{K}} \cdot 1.7 \text{ m}^2 (32 - 23) \text{ K}$

Answer:  $q_{\text{total}} = q_{\text{radiation}} + q_{\text{convection}} = 84.9 \text{ W} + 76.5 \text{ W}$

$q_{\text{total}} = 160 \text{ W}$  High heat loss for high  $T_s$ !

## HW Assignment # 1

## Problem # 4

Statement: Heat diffusion equation is given in simplest form for a medium as:

$$\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} = \frac{1}{\alpha} \frac{\partial T}{\partial t},$$

where  $\alpha = \frac{k}{\rho c_p}$  is the thermal diffusivity.

Find:

- transient or steady?
- 1-D, 2-D, or 3-D?
- thermal generation,  $\dot{q} = ?$
- constant or variable conductivity  $k$ ?

Assumptions: The medium is a solid or a non-convecting fluid.

- Answer:
- Transient because of the term  $\frac{\partial T}{\partial t}$ .
  - 2-D because of the terms  $\frac{\partial^2 T}{\partial x^2}$  and  $\frac{\partial^2 T}{\partial y^2}$ .
  - No thermal generation;  $\dot{q}$  must be 0.
  - Constant  $k$  because for variable  $k$ ,
 
$$\frac{d}{dx} \left( k \frac{\partial T}{\partial x} \right) + \frac{d}{dy} \left( k \frac{\partial T}{\partial y} \right) = \rho c_p \frac{\partial T}{\partial t}.$$

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## HW Assignment #1

### Problem #5

Statement: Stainless steel pan bottom conducts heat one-dimensionally to boil water @ standard  $P_{atm}$ .

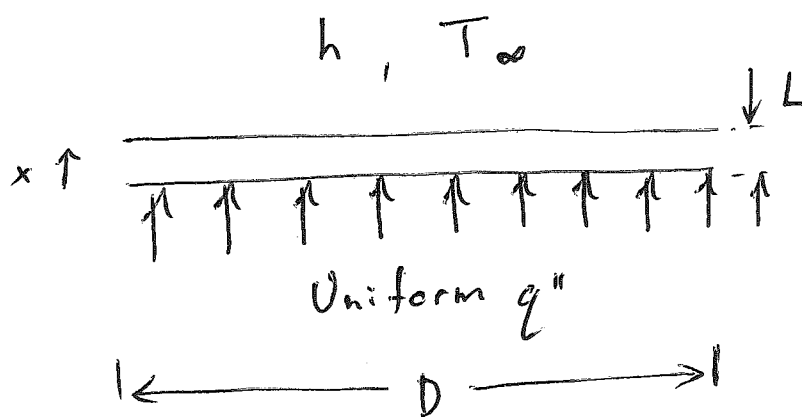
At steady-state,  $T_{water} = 100^\circ C$ ,  $h = 3400 \frac{W}{m^2 K}$ .

Pan geometry is known:  $D = 20 \text{ cm}$  and  $L = 0.30 \text{ cm}$ .

Heat flux to bottom of pan is  $q'' = \frac{0.85 \times 1000 \text{ W}}{A}$

Find: Steady-state temperature distribution through thickness of the pan bottom; differential equation and boundary / initial condition(s) for 1-D transient conduction in pan bottom.

Schematic:



Differential CV is used to obtain:

$$\frac{d}{dx} \left( k \frac{dT}{dx} \right) = \rho c \frac{dT}{dt}$$

Assumptions: • Standard atmospheric pressure with water boiling point at  $100^\circ C$ .

• Constant stainless steel thermal conductivity  $k_{ss} = 15 \text{ W/mK}$ .

Problem #5 continued.

Assumptions: • Shallow pan such that side walls  
(continued) have negligible effect on 1-D heat transfer through pan bottom.

• No thermal energy generation,  $\dot{q}''' = 0$

Analysis: For the more general 1-D transient case, both  $h$  and  $T_\infty$  are variable. The heat diffusion equation reduces to:  $k_{ss} \frac{d^2 T}{dx^2} = \rho c \frac{dT}{dt}$ .

Boundary conditions are imposed at  $x=0, L$ .

$$\begin{array}{l} \text{---} x=L \\ \uparrow x \\ \text{---} x=0 \end{array} \quad \begin{array}{l} -k_{ss} \left( \frac{dT}{dx} \right) \Big|_{x=L} = h \left[ (T) \Big|_{x=L} - T_\infty \right] \\ q'' = -k_{ss} \left( \frac{dT}{dx} \right) \Big|_{x=0} \end{array}$$

To solve the transient problem, an initial condition is required and hence a further assumption. For example, the water and pan can be assumed to be at an initial uniform temperature of  $T = 20^\circ\text{C}$ .

$$T(x, t=0) = 20^\circ\text{C} \quad \text{and} \quad T_\infty(t=0) = 20^\circ\text{C}.$$

$$\text{Also at } t=0, \quad h(t=0) = 0, \quad \text{or} \quad \left. \frac{dT}{dx} \right|_{x=L} = 0.$$

$t=0$

To solve transient conduction problem, convection analysis is also needed to obtain  $h$  and  $T_\infty$  as functions of time,  $h(t)$  and  $T_\infty(t)$ .



Steady-state

Analysis :  $k_{ss} \frac{d^2 T}{dx^2} = \rho c \frac{dT}{dt}$

Problem #5 continued

No initial condition.

$$T_{\infty} = 100^{\circ}\text{C} ; h = 3400 \frac{\text{W}}{\text{m}^2\text{K}} ; q'' = \frac{850 \text{ W}}{\pi (0.10\text{m})^2}$$

Boundary conditions as before with fixed  $T_{\infty}$  and  $h$ .

$$q'' = -k_{ss} \left. \frac{dT}{dx} \right|_{x=0} ; -k_{ss} \left. \frac{dT}{dx} \right|_{x=L} = h \left[ \left( T \Big|_{x=L} - T_{\infty} \right) \right]$$

From the reduced heat diffusion equation,

$$\frac{d^2 T}{dx^2} = 0 \quad \text{and therefore} \quad \frac{dT}{dx} = \text{constant} = -\frac{q''}{k_{ss}}$$

$$T(x) = C_1 x + C_2 = -\frac{q''}{k_{ss}} x + C_2 \quad \text{Linear!}$$

Using the boundary condition at  $x=L$ , solve for  $T(x=L)$ .

$$q'' = h \left[ T(x=L) - T_{\infty} \right] ; T(x=L) = \frac{q''}{h} + T_{\infty}$$

$$T(x=L) = \frac{850 \text{ W}}{\pi (0.10\text{m})^2 \cdot 3400 \frac{\text{W}}{\text{m}^2\text{K}}} + 100^{\circ}\text{C} = 107.96^{\circ}\text{C} \approx 108^{\circ}\text{C}$$

Obtain  $C_2$  by substituting  $T(x=L)$  at  $x=L$  into the general solution:

$$T(x=L) = \frac{q''}{h} + T_{\infty} = -\frac{q''}{k_{ss}} L + C_2$$

$$C_2 = \frac{q''}{h} + \frac{q''}{k_{ss}} L + T_{\infty} = T(x=L) + \frac{850 \text{ W} \cdot 0.003 \text{ m}}{\pi (0.10\text{m})^2 \cdot 15 \frac{\text{W}}{\text{m}^2\text{K}}}$$

$$C_2 = 113.37^{\circ}\text{C} = T(x=0)$$

Answer:

$$T = -1804 \frac{^{\circ}\text{C}}{\text{m}} x + 113.4^{\circ}\text{C}$$

Linear temperature decrease through pan thickness!