

Problem 1

Ex. 2.4 Dixon

Solution. From the definition of reheat factor, eqn. (2.39), the turbine total to total efficiency can be immediately determined:-

$$\eta_t = \eta_p^{R_H} = 0.85 \times 1.04 = 0.884$$

Note : if you assume that the given conditions are static then assume that the inlet and outlet turbine velocities are the same so that $\Delta h_o = \Delta h$.

Using the notation given in Q.2.1 the isentropic stagnation enthalpy drop, $h_{o1} - h_{o2s}$, can be determined using steam tables or, less accurately but more quickly, using a Mollier diagram for steam. From steam tables at $p_{o1} = 4 \text{ MPa}$ (40 bar) and $T_{o1} = 300^\circ\text{C}$ the initial steam condition is superheated (about 50°C of superheat) with $h_{o1} = 2963 \text{ kJ/kg}$ and $s_{o1} = 6.364 \text{ kJ/kg }^\circ\text{C}$. Inspection of the tables shows that at $p_{o2} = 0.35 \text{ MPa}$ (3.5 bar) the vapour saturation value of specific entropy $s_{go2} > s_{o1}$. This means the isentropic state point $o2s$ is in the liquid-vapour phase. The dryness fraction q can be evaluated for point $o2s$,

$$q = (s_{o2} - s_{fo2}) / (s_{go2} - s_{fo2})$$

$$= (6.364 - 1.727) / 5.214 = 0.8893$$

Hence, the specific stagnation enthalpy at point $o2s$ is

$$\begin{aligned} h_{o2s} &= h_{fo2} + q(h_{go2} - h_{fo2}) \\ &= 584 + 0.8893 \times 2148 = 2494 \text{ kJ/kg} \end{aligned}$$

Thus, the isentropic stagnation enthalpy drop is

$$h_{o1} - h_{o2s} = 2963 - 2494 = 469 \text{ kJ/kg}$$

As the total to total efficiency is known the actual stagnation enthalpy drop can be found, i.e.

$$h_{o1} - h_{o2} = \eta_t (h_{o1} - h_{o2s}) = 0.884 \times 469 = 414.6 \text{ kJ/kg}$$

and this is the specific work done by the steam, ΔW . The actual specific work delivered at the output shaft is less than this because of the mechanical losses. The shaft power delivered is

$$W_t = \eta_m \dot{m} \Delta W$$

where η_m is the mechanical efficiency. Thus, the rate of mass flow

$$\begin{aligned} \dot{m} &= W_t / (\eta_m \Delta W) \\ &= 20 \times 10^6 / (0.98 \times 414.6 \times 10^3) \\ &= \underline{\underline{49.22 \text{ kg/s}}} \end{aligned}$$

From the equation of continuity and assuming uniform flow at all radii,

$$\dot{m} = \rho A c_x = \rho \pi d_m h c \cos \alpha_1$$

Hence, the blade height is

$$\begin{aligned} h &= \dot{m} v / (\pi d_m c \cos \alpha_1) \\ &= 49.22 \times 0.0686 / (\pi \times 0.762 \times 244 \times .2419) = 0.0239 \text{ m} \\ &= \underline{\underline{23.9 \text{ mm}}} \end{aligned}$$

Problem 2

$$d_{sh_1} = 0.283 \text{ m}$$

$$d_{hb_{sh_1}} = 0.85 \quad R = 287 \frac{\text{J}}{\text{kgK}} \quad (\text{air})$$

$$\dot{m} = 8 \text{ kg/s}$$

$$T_{01} = 1550 \text{ K}$$

$$P_{01} = 15 \text{ bar}$$

$$\gamma = 1.4 \quad (\text{assume perfect gas})$$

flows enters nozzle w/o swirl

$$? \alpha_1 \text{ such that } M_1 = 1$$

$$\dot{m} = P_{01} A_{1\perp} \left(\frac{M_1}{RT_{01}} \right)^{\frac{\gamma+1}{2(\gamma-1)}} \quad ?$$

$$A_1 = \frac{\pi}{4} \left(d_{sh_1}^2 - d_{hb_1}^2 \right) = \frac{\pi}{4} d_{sh_1}^2 \left[1 - \left(\frac{d_{hb_1}}{d_{sh_1}} \right)^2 \right] = \frac{\pi}{4} 0.283^2 \left(1 - 0.85^2 \right) = 0.01745 \text{ m}^2$$

$$A_{1\perp} = A_1 \cos \alpha_1$$

$$A_{1\perp} = \frac{\dot{m}}{P_{01} M_1 \sqrt{\frac{RT_{01}}{\gamma}}} \left(1 + \frac{\gamma-1}{2} M_1^2 \right)^{\frac{\gamma+1}{2(\gamma-1)}} = \frac{8}{15 \times 10^5 \times 1} \sqrt{\frac{287 \times 1550}{1.4}} \left(1 + 0.2 \times 1 \right)^{\frac{3}{2}} = 0.00519 \text{ m}^2$$

$$\Rightarrow \cos \alpha_1 = \frac{A_{1\perp}}{A_1} = \frac{0.00519}{0.01745} = 0.297 \rightarrow \alpha_1 = 72.7^\circ$$

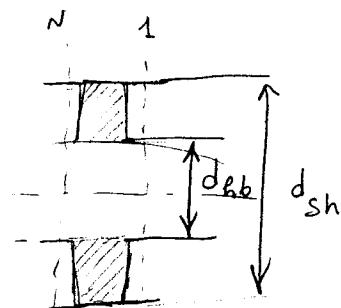
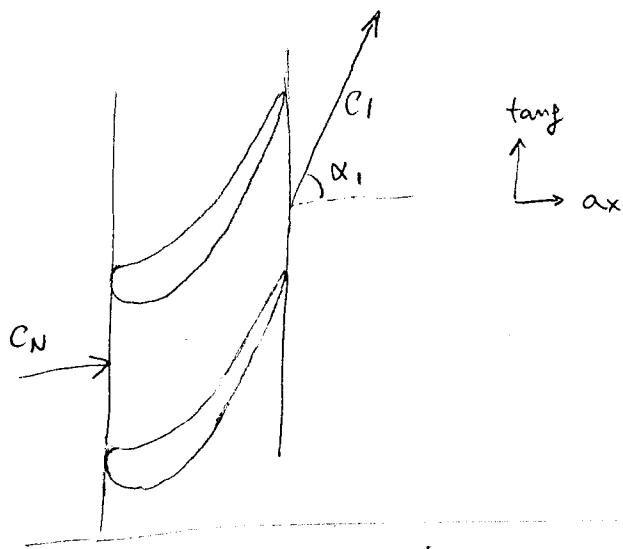
alternative way : $T_1 = T_{01} \left(1 + \frac{\gamma-1}{2} M_1^2 \right)^{-1} = 1550 \left(1 + 0.2 \right)^{-1} = 1291.7 \text{ K}$

to find $A_{1\perp}$ $P_1 = P_{01} \left(1 + \frac{\gamma-1}{2} M_1^2 \right)^{-\frac{\gamma}{\gamma-1}} = 15 \left(1.2 \right)^{-3.5} = 7.924 \text{ bar} = 702.4 \text{ kPa}$

$$\rightarrow \rho = \frac{P}{RT} = \frac{702.4 \text{ kPa}}{0.287 \frac{\text{kPa m}^3}{\text{kg K}} \times 1291.7} = 2.137 \text{ kg/m}^3$$

$$c_1 = M_1 \sqrt{\rho RT_1} = 1 \times \sqrt{1.4 \times 287 \times 1291.7} = 720.4 \text{ m/s}$$

$$\dot{m} = \rho c_1 A_{1\perp} \Rightarrow A_{1\perp} = \frac{\dot{m}}{\rho c_1} = \frac{8}{2.137 \times 720.4} = 5.19 \times 10^{-3} \text{ m}^2$$



problem 3
compressor

$$c_1 = 300 \text{ m/s}$$

$$A_1 = 0.08 \text{ m}^2$$

$$T_{en} = 300 \text{ K}$$

$$P_{en} = 100 \text{ kPa}$$

$$\dot{W} = 3 \text{ MW}$$

$$\text{air perfect gas } \gamma = 1.4, C_p = 1005 \text{ J/kgK}$$

Evaluate :

a) $T_{01}, P_{01}, \rho_{01}, T_1, P_1, \rho_1$ (inlet)

b) $P_{02\max}$ (outlet)

c) entropy generated if $P_{02} = 0.97 P_{02\max}$

a) environmental conditions are equal to stagnation conditions at inlet

$$P_{01} = P_{en} = 100 \text{ kPa} \quad \checkmark$$

$$T_{01} = T_{en} = 300 \text{ K} \quad \checkmark$$

$$\rightarrow \rho_{01} = \frac{P_{01}}{RT_{01}} = \frac{100 \text{ kPa}}{0.287 \frac{\text{kPa} \cdot \text{m}^3}{\text{kg K}} 300 \text{ K}} = 1.16 \frac{\text{kg}}{\text{m}^3} \quad \checkmark$$

$$T_1 = T_{01} - \frac{C_1^2}{2C_p} = 300 - \frac{300^2}{2 \times 1005} = 255.2 \text{ K} \quad \checkmark$$

$$M_1 = \frac{C_1}{\sqrt{RT_1}} = \frac{300}{\sqrt{1.4 \times 287 \times 255.2}} = 0.937 \quad \text{Note: (flow is compressible!)}$$

$$P_1 = P_{01} \left(1 + \frac{\gamma-1}{2} M_1^2 \right)^{-\frac{\gamma}{\gamma-1}} = 100 \text{ kPa} \left(1 + 0.2 \times 0.937^2 \right)^{-\frac{1.4}{0.4}} = 56.77 \text{ kPa} \quad \checkmark$$

$$\rho_1 = \frac{P_1}{RT_1} = \frac{56.77 \text{ kPa}}{0.287 \frac{\text{kPa m}^3}{\text{kg K}} 255.2} = 0.775 \frac{\text{kg}}{\text{m}^3} \quad \checkmark$$

b) $m = \rho_1 A_1 c_1 = 0.775 \frac{\text{kg}}{\text{m}^3} \times 0.08 \text{ m}^2 \times 300 \frac{\text{m}}{\text{s}} = 18.6 \frac{\text{kg}}{\text{s}}$

specific work : $W = \frac{\dot{W}}{m} = \frac{3 \cdot 10^6 \text{ J/s}}{18.6 \frac{\text{kg}}{\text{s}}} = 161.3 \text{ kJ/kg}$

$$w = h_{02} - h_{01} = C_p (T_{02} - T_{01}) \Rightarrow T_{02} = T_{01} + \frac{w}{C_p} = 300 + \frac{161.3 \times 10^3}{1005} = 460.5 \text{ K}$$

$P_{02\max}$ when process $1 \xrightarrow{isentropic} 2$ is isentropic : $P_{02\max} = P_{01} \left(\frac{T_{02}}{T_{01}} \right)^{\frac{\gamma-1}{\gamma}} = 100 \text{ kPa} \left(\frac{460.5}{300} \right)^{\frac{1.4}{0.4}} = 448 \text{ kPa}$

c) entropy generated if $P_{02} = 0.97 P_{02\max} = 0.97 \times 448.1 = 434.7 \text{ kPa}$

$$\frac{\dot{S}_{gen}}{m} = S_{02} - S_{01} = C_p \ln \frac{T_{02}}{T_{01}} - R \ln \frac{P_{02}}{P_{01}} = 1005 \ln \left(\frac{460.5}{300} \right) - 287 \ln \left(\frac{434.7}{100} \right) = 8.93 \frac{\text{J}}{\text{K}}$$

$$\dot{S}_{gen} = m (S_{02} - S_{01}) = 18.6 \times 8.93 = 166.2 \frac{\text{W}}{\text{K}}$$