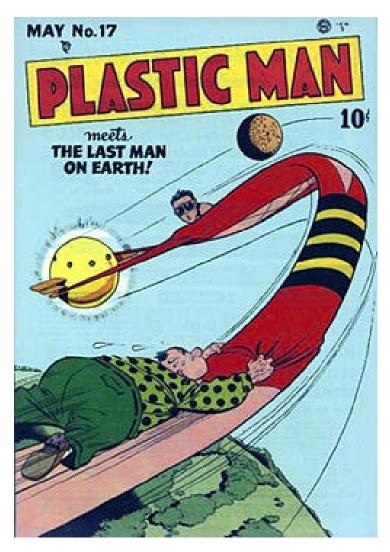
Polymers Continued....



Plastic Man #17 (May 1949) Cover art by Jack Cole.

From Monday:

- Polymer- Long chain molecule made from repeating units
- Monomer examples
- Boiling points of hydrocarbons
- •M_n- Number average molecular weight
- Linear, branched, cross linked

Polymers

Today:

- Nylon Demonstration
- Resin Codes of Common Polymers
- Special conducting polymers
- Mechanical Behavior of Polymers
- Glass transition temperature (Tg)

Two Solutions

- •NH₂(CH₂)₆NH₂
- •CIOC(CH₂)₄COCI

Produces Nylon 6,6

$$-N - \begin{bmatrix} H & O & H & O \\ -C & -N - C - \begin{bmatrix} H & O & H \\ -C & -N - C - \begin{bmatrix} H & O & H \\ -C & -C - \end{bmatrix} \\ H & H \end{bmatrix}_{6}^{O}$$

Poly (urethane) – A thermoset cross linked polymer

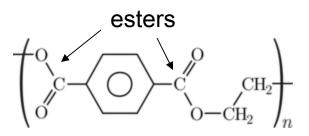
Note: Can be reused, but not recycled

Common Polymers- Resin Codes

Poly (ethylene terephthalate)

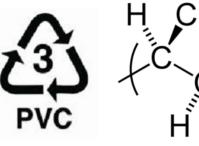


Drink Bottles
Plastic Parts
Clothing Fibers



poly (ester)

Poly (vinyl chloride)

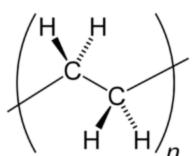


Pipe

High density poly (ethylene)



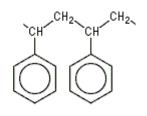
Food Containers Grocery Bags Injection Molded Parts



Poly (styrene)



Styrofoam CD cases

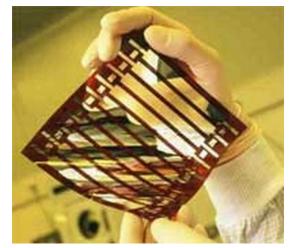


Uncommon Polymers- Conducting Polymers

poly (3-hexyl thiophene)

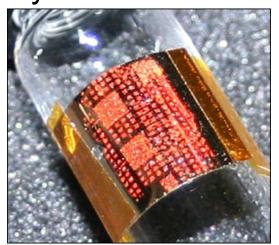
$$C_6H_{13}$$
 C_6H_{13} C_6H_{13} C_6H_{13} C_6H_{13} C_6H_{13}

Polymer Solar Cell

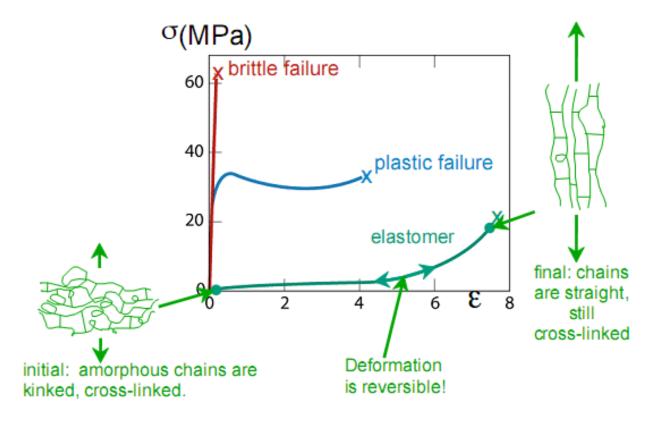


(Photo: UCSB)

Polymer Transistor



Tensile response: elastomers



Stress-strain curves adapted from Fig. 15.1, Callister 7e. Inset figures along elastomer curve (green) adapted from Fig. 15.15, Callister 7e. (Fig. 15.15 is from Z.D. Jastrzebski, The Nature and Properties of Engineering Materials, 3rd ed., John Wiley and Sons, 1987.)

- · Compare to responses of other polymers:
 - brittle response (aligned, crosslinked & networked polymer)
 - plastic response (semi-crystalline polymers)

Polymer crystalline structure

Polymers rarely 100% crystalline Too difficult to get all those chains aligned

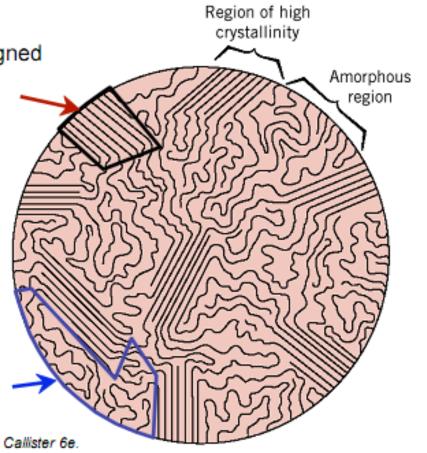
crystalline region

- % Crystallinity: % of material that is crystalline.
 - -- TS and E often increase with % crystallinity.
 - Annealing causes crystalline regions to grow. % crystallinity increases.

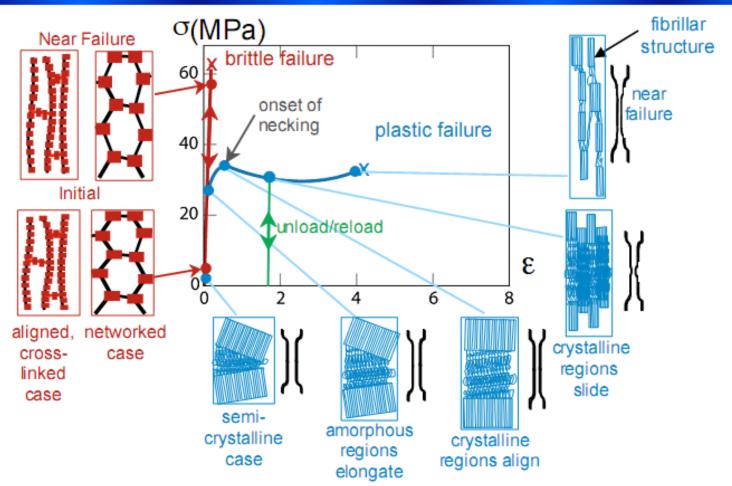
amorphous region

Adapted from Fig. 14.11, Callister 6e.

(Fig. 14.11 is from H.W. Hayden, W.G. Moffatt, and J. Wulff, The Structure and Properties of Materials, Vol. III, Mechanical Behavior, John Wiley and Sons, Inc., 1965.)



Brittle and plastic behavior



Stress-strain curves adapted from Fig. 15.1, Callister 7e. Inset figures along plastic response curve adapted from Figs. 15.12 & 15.13, Callister 7e. (Figs. 15.12 & 15.13 are from J.M. Schultz, Polymer Materials Science, Prenti Hall, Inc., 1974, pp. 500-501.)

T_q Glass Transiton Temperature

Tg – temperature at which a material transitions from a glassy, crystalline state to an amorphous state

HDPE Tg \sim -100°C

Will HDPE be mostly amorphous or crystalline at room temperature?

PS Tg ~ 95°C

Will PS be mostly amorphous or crystalline at room temperature?

Can use density (ρ) to calculate % crystallinity