

ENGR/MSE 170 final review

- ☐ Exam date: Section A: 6/5/2007 Tues, 8:30-10:20am; B: 6/7/2007 Thurs 8:30am-10:20pm
- ☐ Place: Mueller 153
- ☐ The final exam will be comprehensive
- ☐ Closed book, notes and neighbors
- ☐ 2 sheets of double-side letter-sized paper are allowed
- ☐ Material on the exam will be taken from text book, lecture, homework and lab
- ☐ Bring a calculator and straight edge/triangle

Bonding and atomic forces

- The Periodic table
- What types of bonding are there?
- How does bonding affect materials properties?

Bonding and atomic forces (continue)

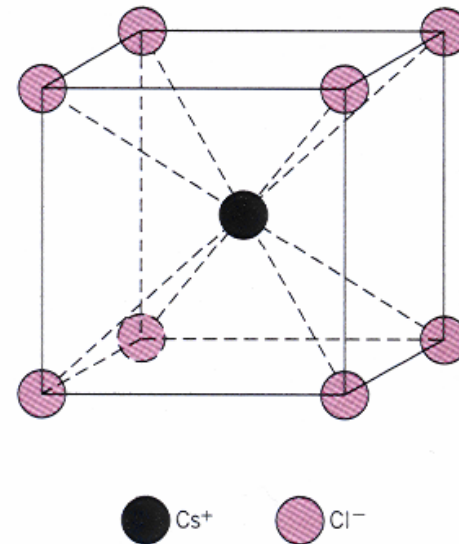
- Atomic forces & potential vs interatomic distance

Crystal structure

- ❑ Determine #atoms/unit cell, CN#, APF and density
- ❑ Draw and index crystallographic direction and planes
- ❑ Close-packed plane stacking sequence

Crystal structure (*continue*)

- Know the types of ceramic structures and identify their unit cells
- Determine coordination number for ceramic structures based on ionic radii and charge



Defects

- Distinguish point, linear (dislocation 1D), interfacial(2D), volume(3D)
- Draw and describe edge and screw dislocations
- Understand equilibrium of vacancies and effect of T

Diffusion

- Mechanisms, concentration gradients, diffusion coefficient
- Ficks first law
- Effects of T on diffusion coefficients
- Diffusion paths (interstitial vs. vacancy, grain boundary and surface)

Mechanical properties

- Definition of stress, strain, elastic modulus
- Analysis of stress-strain curves
- Yield strength, tensile strength, Poisson's ratio, ductility, resilience and toughness
- Hardness

Plastic deformation

- Slip plane, direction and system, resolved shear stress
- Mechanism of plastic deformation
- Strengthening mechanisms
- Recovery, recrystallization and grain growth

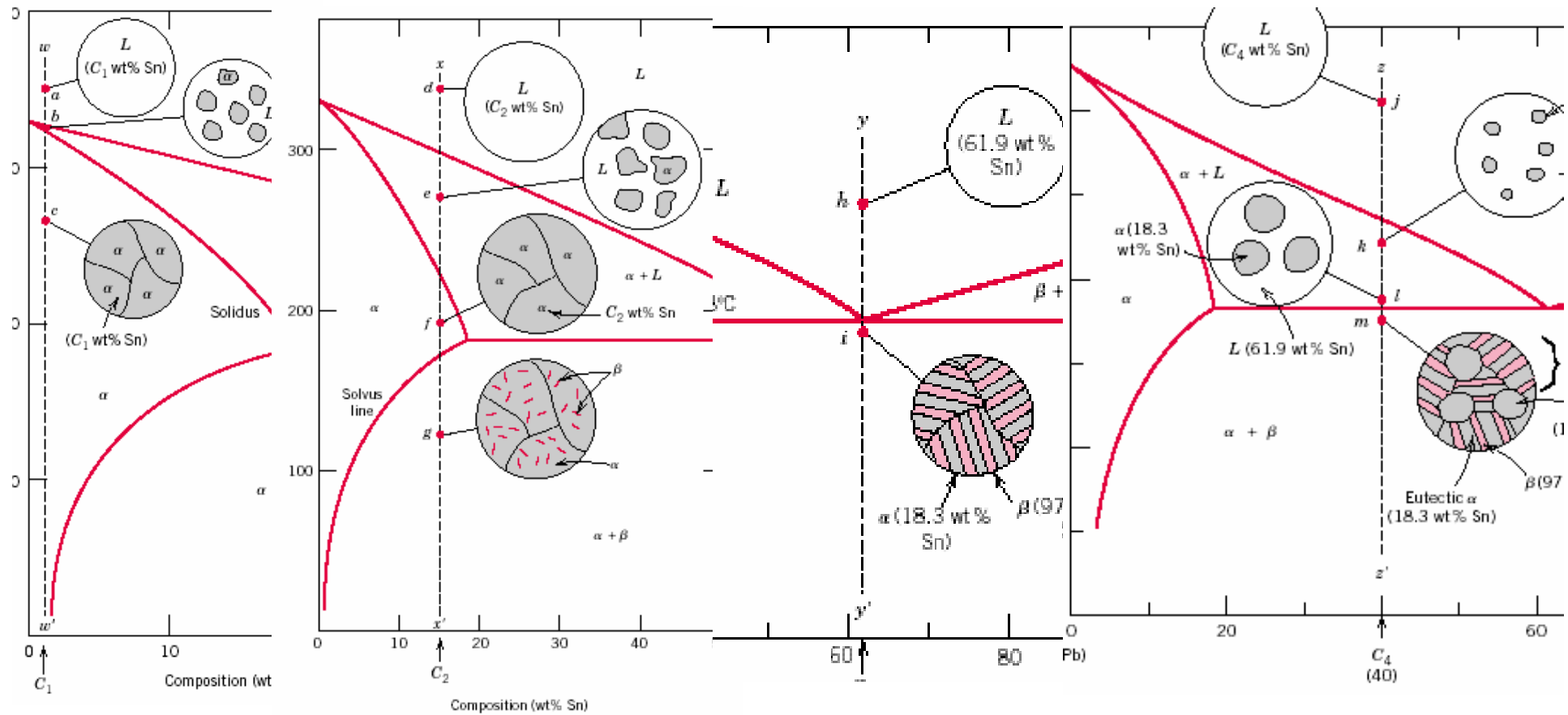
Failure

- Ductile vs brittle fracture
- Stress concentrations and fracture toughness
- Creep and fatigue failure

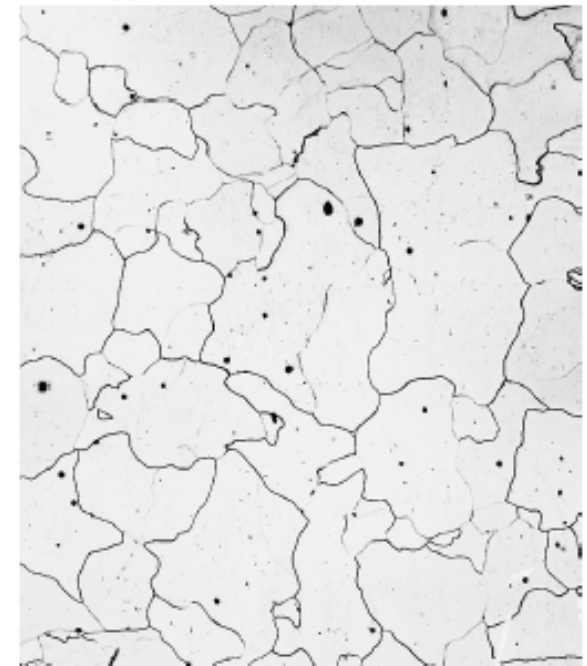
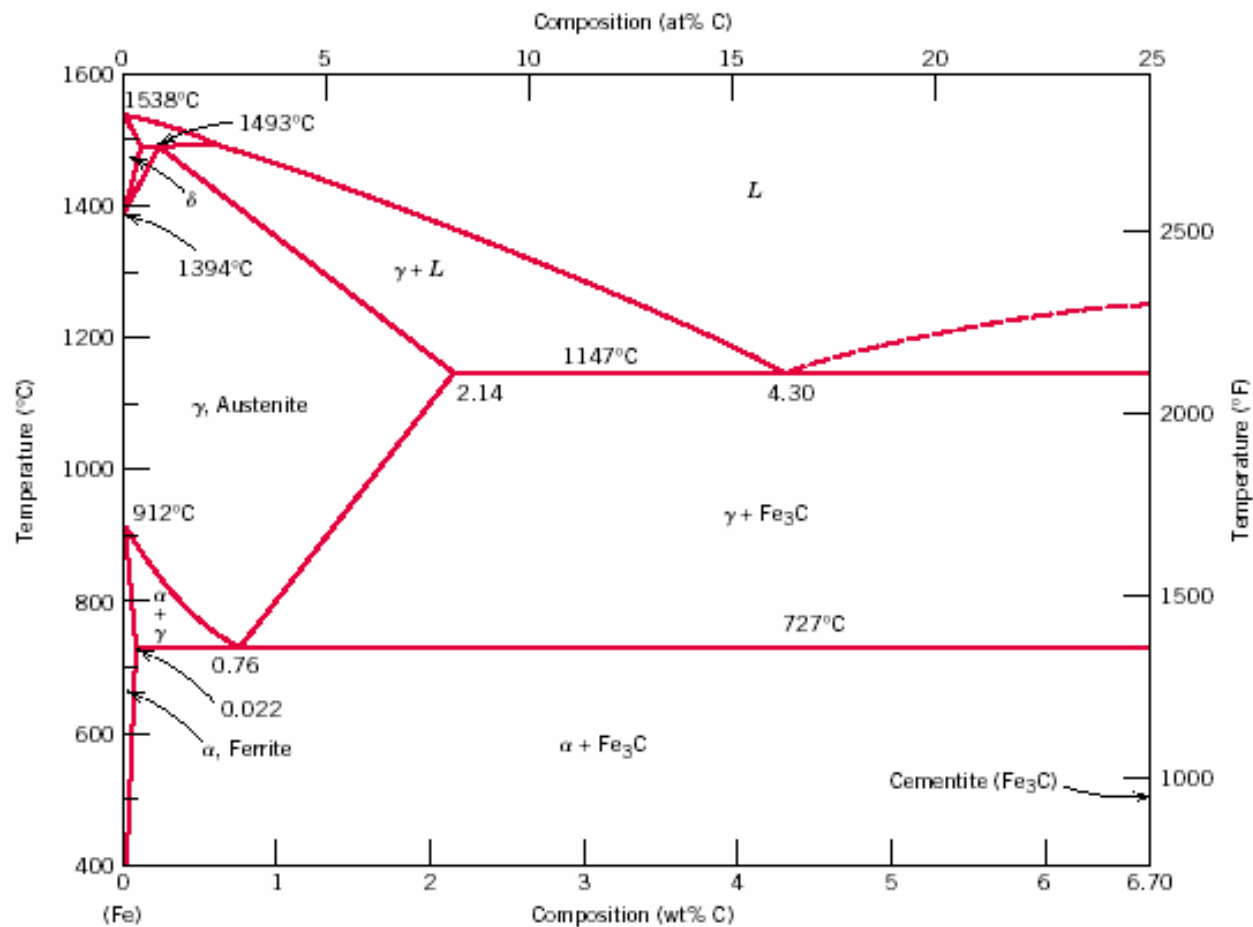
Phase diagram

- ☐ Solubility limits, solidus, liquidus
- ☐ Binary isomorphous and eutectic phase diagrams
- ☐ Eutectic, peritectic, eutectoid reactions
- ☐ Lever law, weight fraction of phases
- ☐ Composition and microstructure of equilibrium phases

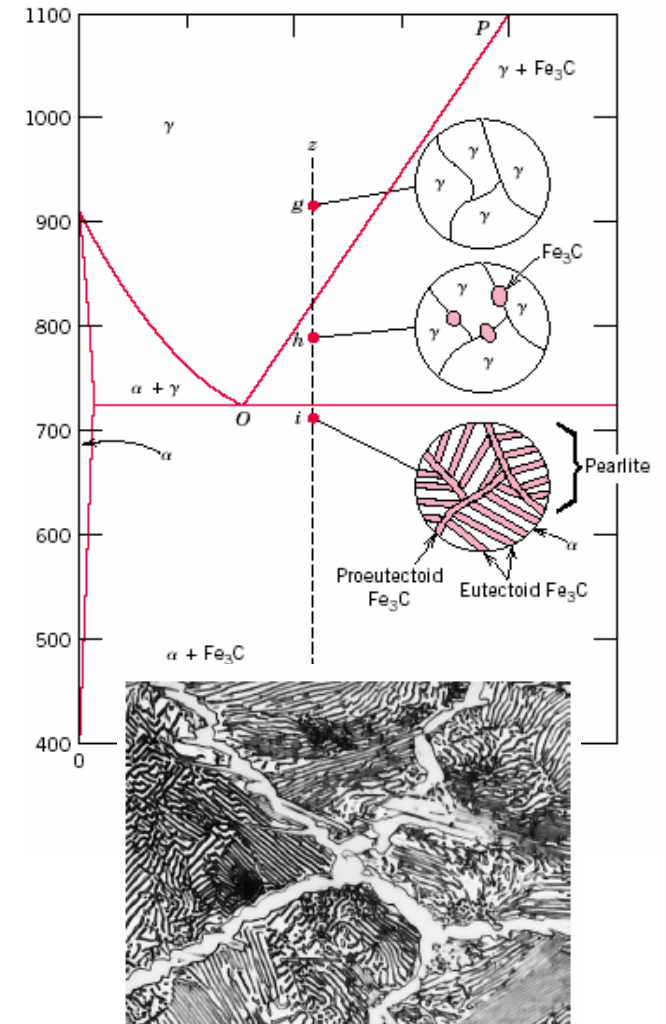
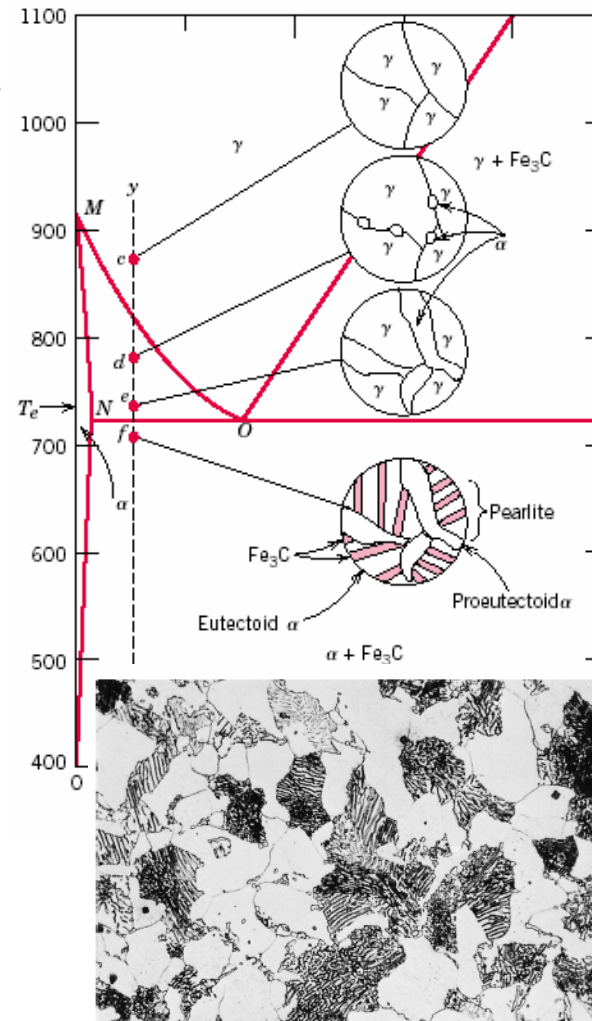
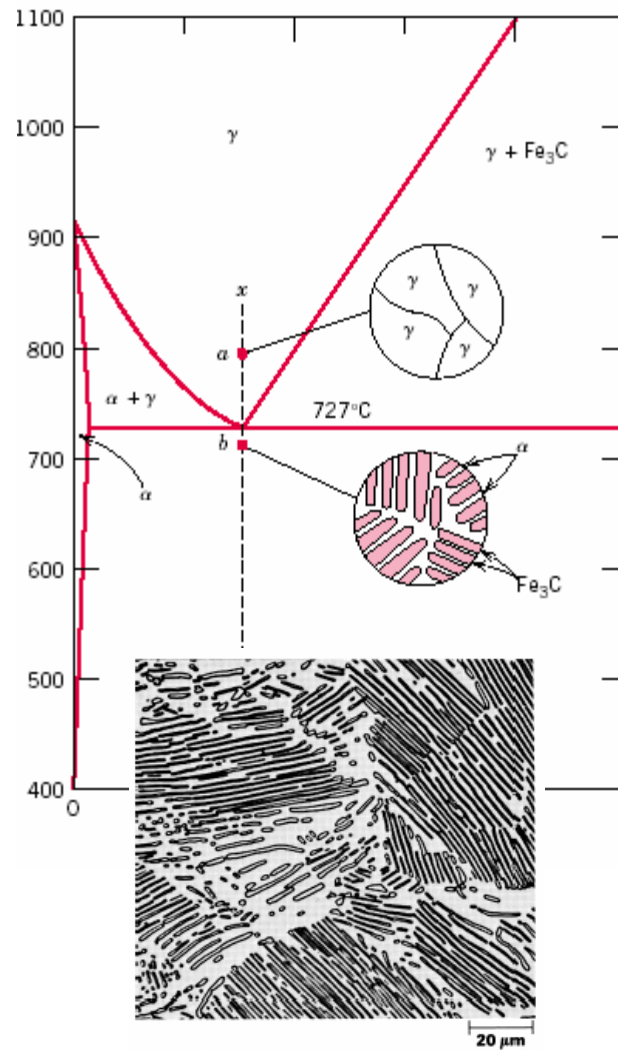
Phase diagram



Phase diagram



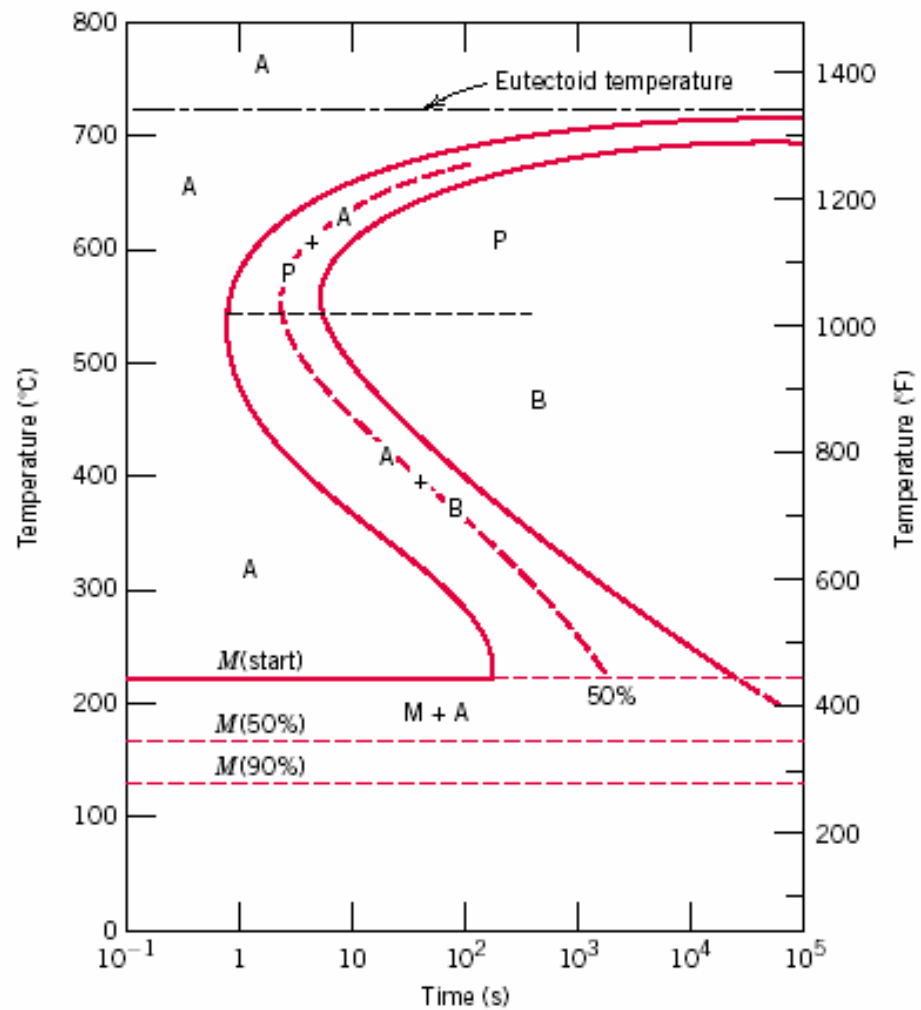
Phase diagram



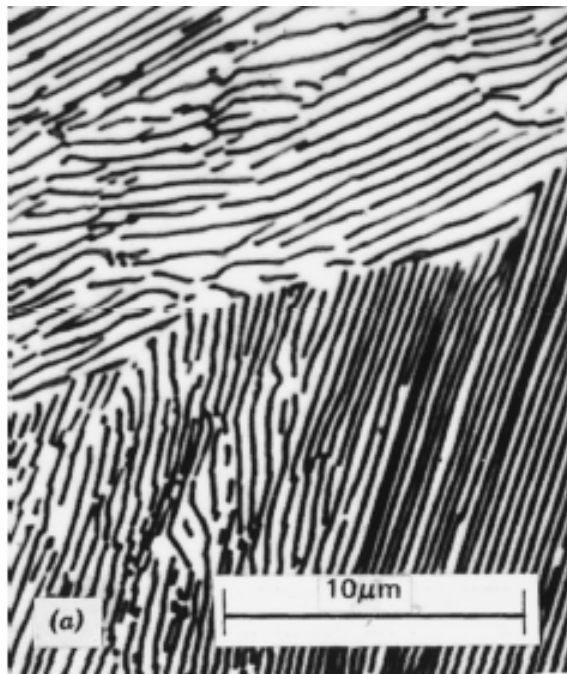
Phase transformation

- Nucleation/growth controlled phase transformation
- Diffusional vs diffusionless
- Rate of transformation
- Isothermal transformation diagrams

Phase transformation



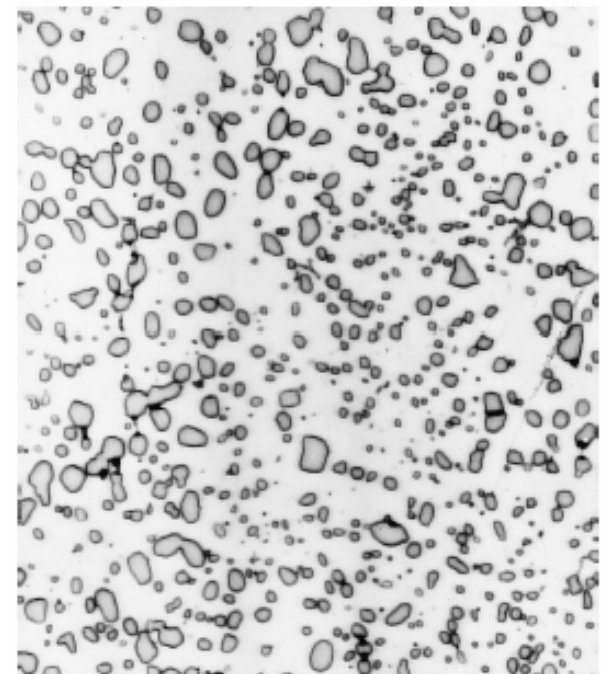
Phase transformation



3000X



3000X



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Electrical properties

- ☐ Intrinsic semiconductors
- ☐ Extrinsic semiconductors
- ☐ Energy band structure
- ☐ Electric conductivity, resistivity and mobility
- ☐ Carrier concentration
- ☐ P-N junctions and diodes

Polymer properties

☐ Hydrocarbon molecules

- Saturated and unsaturated
- Bonding
- Melting point
- Isomerism
- mer, monomer, polymer

☐ copolymer, homopolymer, blockpolymer

☐ molecular structures of polymers

☐ Mechanical properties of polymers

☐ Crystallization, melting, and glass transition phenomena

☐ Thermoplastic and thermosetting polymers

Structures and properties of ceramics

- Imperfection in ceramics
- Electric properties of ceramics
- Ceramic phase diagrams
- Mechanical properties of ceramics

Composites

- Composites, matrix, dispersed phase
- Types of composites and characteristics of each
- Mechanical properties of composites including upper bound and lower bound of elastic modulus