

Recap: Thermal Energy Conservation ()

Thermal Energy Conservation Eq (g const.):

Eq. 2.4-1: $\rho C_p \frac{DT}{Dt} = -\vec{\nabla} \cdot \mathbf{q} + H_V$ (const. ρ and C_p)

with Fourier:

Eq. 2.4-3: $\rho C_p \frac{DT}{Dt} = k \nabla^2 T + H_V$ (const. ρ, C_p, k)

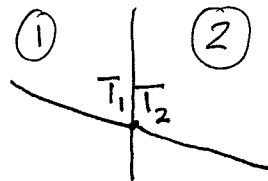
or $\frac{DT}{Dt} = \alpha \nabla^2 T + \frac{H_V}{\rho C_p}$; $\alpha \equiv \frac{k}{\rho C_p}$

At Interfaces: normal flux: $q_n|_{1,2}$; phases 1 and 2 separated by interface

Eq. 2.5-1 $q_n(\vec{r}_s, t)|_2 - q_n(\vec{r}_s, t)|_1 = H_s(\vec{r}_s, t)$
↑
interfacial location
↓
usual situation, $H_s = 0$
 $q_n|_1 = q_n|_2$ heat flux continuous at interface

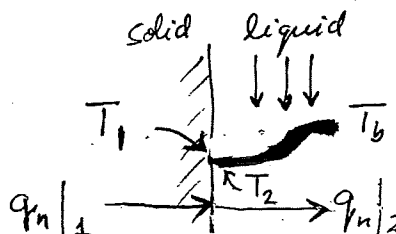
Zero resistance interface: $T_1 = T_2$

(applicable to liquid/solid interface)
(not applicable to solid/solid interface)



Convection Boundary Condition

Eq. 2.5-5: $q_n|_1 \equiv h(T_2 - T_b)$
↑ ↑ ↑
solid liquid bulk liquid temp.
↑
convection heat transfer coeff.
↓
interfacial temp. of liquid
in solid: $q_n|_1 = -k_1(\vec{n} \cdot \vec{\nabla} T)_1$
 $q_n|_1 = q_n|_2 \Rightarrow -k_1(\vec{n} \cdot \vec{\nabla} T)_1 = h(T_2 - T_b)$



Phase Change

mass conservation: $\rho_1 (v_n|_1 - v_{In}) = \rho_2 (v_n|_2 - v_{In})$ Eq. 2.5-10

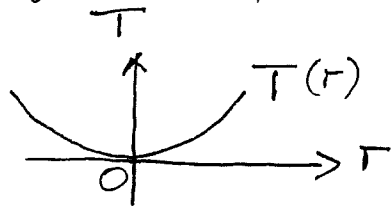
energy conservation: $\rho_1 \hat{H}_1 (v_n|_1 - v_{In}) + q_n|_1 =$
 $= \underbrace{\rho_2 \hat{H}_2 (v_n|_2 - v_{In})}_{\text{convective contribution}} + q_n|_2$ Eq. 2.5-11

See textbook for combined Eq. 2.5-12

Symmetry

At symmetry planes or symmetry points the normal fluxes vanish, i.e. $\bar{F}_n = 0$

Ex: Axisymmetric point in Heat Transfer



as $q_r = \begin{cases} -k \frac{\partial T}{\partial r} & ; \text{Fourier} \\ 0 & ; \text{Symmetry} \end{cases}$

$$\Rightarrow \left. \frac{\partial T}{\partial r} \right|_{r=0} = 0$$