

PCC 588 - January 6 and 8 2009

Winter 2009: ATMS/OCN/ESS 588
The Global Carbon Cycle and
Greenhouse Gases

T,Th 12:00-1:20 pm OSB 425

Course Goals

The course focuses on factors controlling the global cycle of carbon and the greenhouse gases (CO_2 , CH_4 , N_2O , O_3 and halocarbons).

- Abundance and distribution of carbon and greenhouse gases
- Physical, chemical and biological mechanisms that control ocean-atmosphere and terrestrial-atmosphere exchange of carbon and greenhouse gases
- The geologic evidence for climate change linked to greenhouse gases
- The fate of anthropogenic greenhouse gases, their impact on climate and strategies for sequestration of anthropogenic gases

Course Structure

- Greenhouse gases and radiative forcing (1.5 week)
- Non-CO₂ greenhouse gases and aerosols (2 weeks)
- Carbon cycle: past, present, future (2.5 weeks)
- Anthropogenic perturbation to C-cycle (1.5 week)
- Geoengineering solutions (1 week)
- Class presentations (1 week)

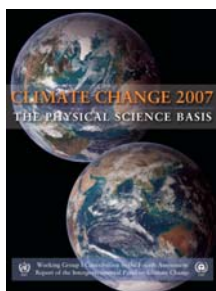
*3 Problem sets (30%); Midterm exam (25%);
Participation in class and during paper discussions
(15%); Term paper & oral presentation (30%)*

The IPCC Reports

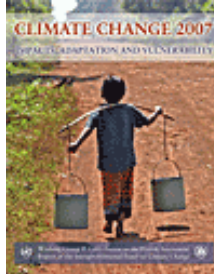
- Intergovernmental Panel on Climate Change (IPCC) established in 1988 by WMO and UNEP
- assess available scientific and socio-economic information on climate change and its impacts and on the options mitigation and adaptation
- Report every 5 years: 1991, 1996, 2001, and 2007
- Compiled by hundreds of scientists, reviewed by scientists, governments and experts: consensus document

<http://www.ipcc.ch/ipccreports/assessments-reports.htm>

The Scientific Basis



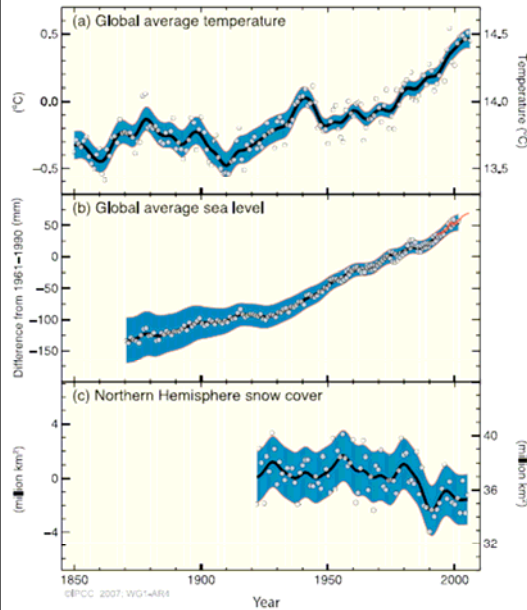
Impacts, Adaptation and vulnerability



Mitigation

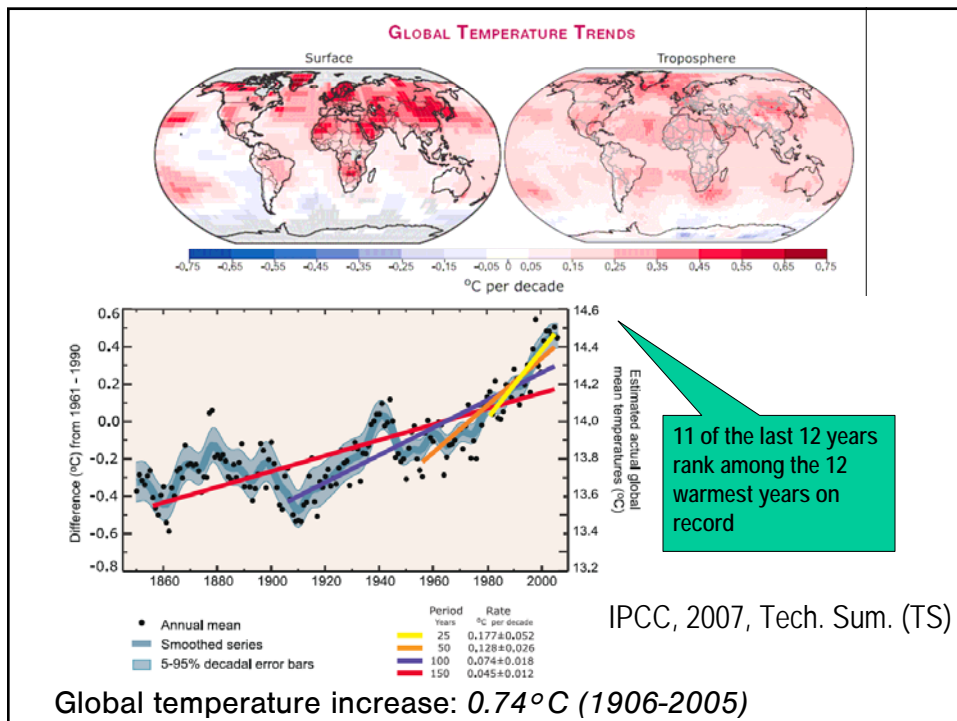


Observed changes in climate

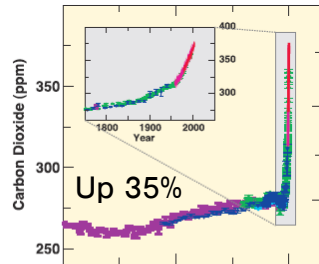


"Warming of the climate system is unequivocal, as is now evident from observations of increases in global average air temperatures, widespread melting of snow and ice, and rising average sea level."

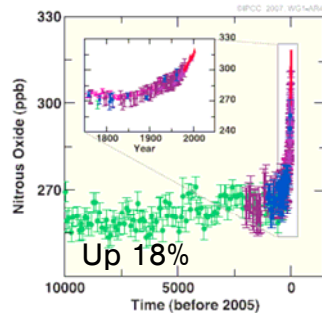
IPCC 2007, Summary for Policymakers (SPM)



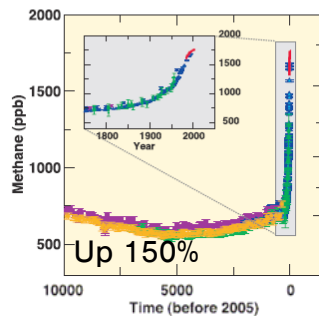
Human drivers of climate change



CO₂
Fossil
fuels +
land use
change



N₂O
agriculture



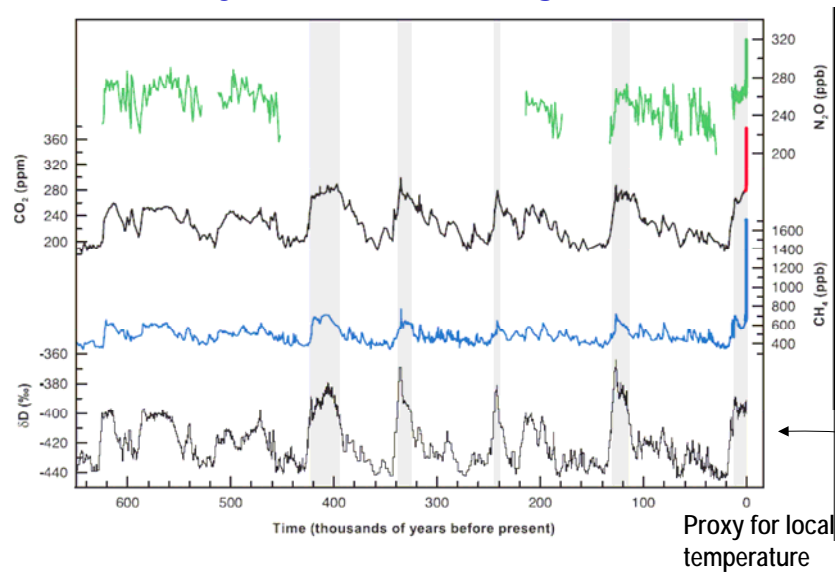
CH₄
Fossil fuel
use +
agriculture

CO₂, CH₄ and N₂O Concentrations

- far exceed pre-industrial values
- increased markedly since 1750 due to human activities

IPCC, SPM, 2007

Past 650,000 years: Glacial-interglacial ice core data



IPCC, TS (2007)

Today and Thursday: From greenhouse gases to climate change ...

- The climate system: what controls the temperature of the Earth?
- Greenhouse effect.
- Radiative forcing. Global warming potential. CO₂-equivalent emissions.
- Relating radiative forcing to temperature changes.

Reading for this week (will be on class web site):

IPCC WG1 Summary for Policymakers (2007) → in-class discussion next Tues!

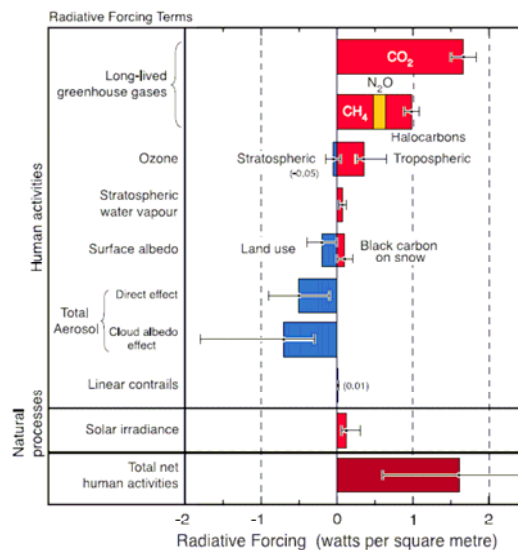
<http://www.ipcc.ch/pdf/assessment-report/ar4/wg1/ar4-wg1-spm.pdf>

For more information:

* IPCC WG1 Technical Summary (2007)

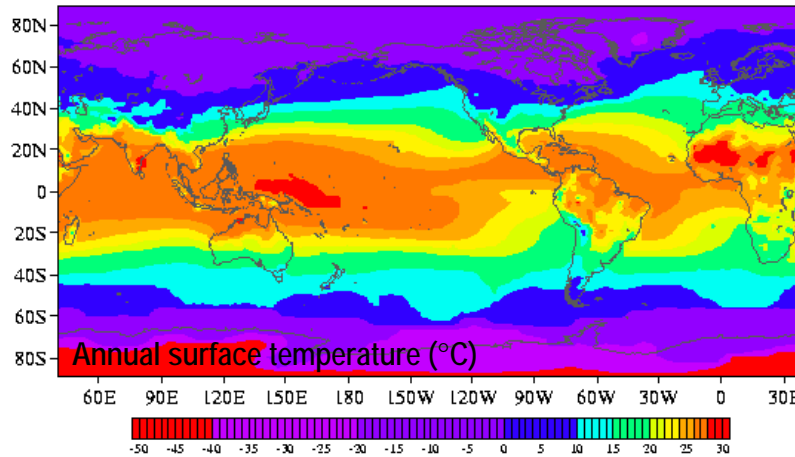
* Radiation tutorial/refresher: Chapter 7 in Jacob's "Introduction to Atmospheric Chemistry"

Human Impacts on the Climate System Radiative forcing of climate between 1750 and 2006



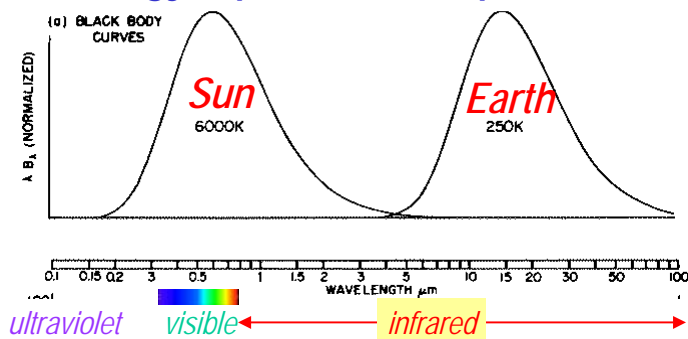
IPCC, Chap 2

Global Temperature



- Annual and global average temperature ~ 15°C, i.e. 288 K

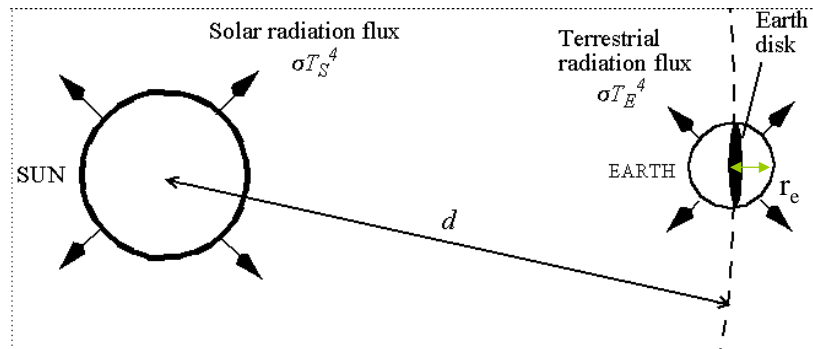
Energy Inputs and Outputs



- Both Sun and Earth behave as blackbodies (absorb 100% incident radiation; emit radiation at all wavelengths in all directions)
- Earth receives energy from sun in the form of shortwave radiation with peak in the visible ($\lambda = 0.4 - 0.7 \mu\text{m}$)
- Earth emits energy to space in the form of longwave radiation in the infrared ($\lambda = 5-20 \mu\text{m}$) → function of Earth's temperature

Total Solar Radiation Received By Earth

- Solar constant for Earth: $F_s = 1368 \text{ W m}^{-2}$ (Note: $1 \text{ W} = 1 \text{ J s}^{-1}$)



- Solar radiation received top of atmosphere unit area of sphere
 $= (1368) \times (\pi r_e^2) / (4 \pi r_e^2) = 342 \text{ W m}^{-2}$

A No-Atmosphere Earth

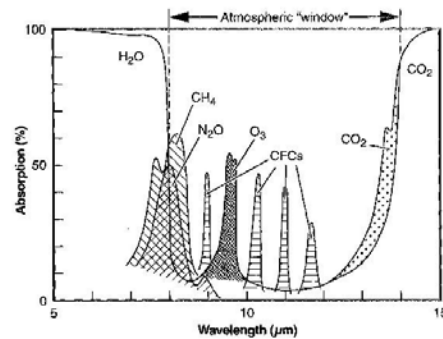
- Assume 30% of incoming solar energy is reflected by surface (albedo of surface = 0.3)
- Energy absorbed by surface = 70% of $342 \text{ W m}^{-2} = 239.4 \text{ W m}^{-2}$
- Balanced by energy emitted by surface
- Stefan-Boltzmann law: Energy emitted = σT^4 ; ($\sigma = 5.67 \times 10^{-8} \text{ W m}^{-2} \text{ K}^{-4}$)
- $239.4 \text{ W m}^{-2} = \sigma T^4 \rightarrow T = 255 \text{ K} (-18^\circ\text{C})$
 \rightarrow much less than average temperature of $288 \text{ K} (15^\circ\text{C})$
- What is missing?
 \rightarrow Absorption of terrestrial radiation by the atmosphere

The diagram illustrates the interaction of different photon wavelengths with atoms and molecules, leading to various transitions:

- Photon of wavelength**: The starting point for all interactions.
- Interacting with an atom**: A photon interacts with an atom, causing an **excitation of outer electrons**. This results in **Electronic transitions**, which can cause... (further effects).
- Interacting with a molecule**: A photon interacts with a molecule, leading to different types of transitions depending on the wavelength:
 - UV ($< 0.4 \mu\text{m}$)**: Causes **photo-dissociation**, leading to **Electronic transitions**.
 - Near IR ($0.7 < \lambda < 20 \mu\text{m}$)**: Causes **rotation**, leading to **Rotational transitions**, which are associated with **Greenhouse gases**.
 - Infrared + microwave ($\lambda > 20 \mu\text{m}$)**: Causes **vibration**, leading to **Vibrational transitions**, which are also associated with **Greenhouse gases**.

Philander, Is the temperature rising? 1998.



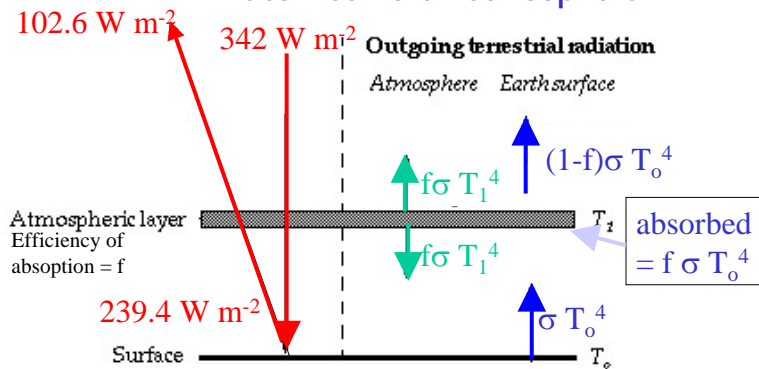


Not all molecules are equal...

Question:

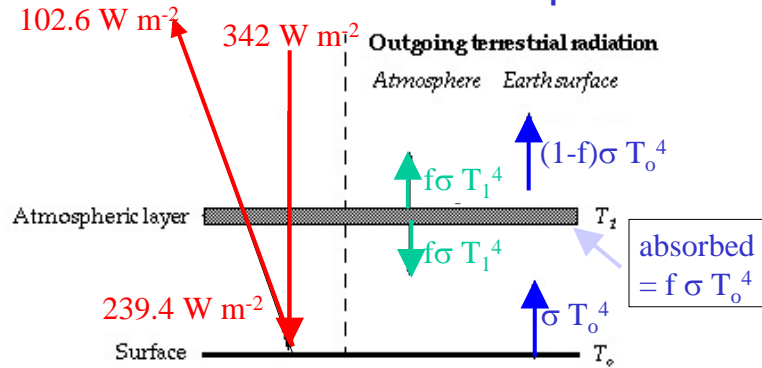
Consider CO_2 , CH_4 , N_2O , O_3 , and CFCs: on a per molecule basis, which do you expect to be most effective at absorbing infrared radiation? Least effective? Why?

An Idealized Earth+atmosphere



- Solar radiation at surface = 70% of $342 \text{ W m}^{-2} = 239.4 \text{ W m}^{-2}$
- Infrared flux from surface = σT_o^4
- Absorption of infrared flux by atmosphere = $f\sigma T_o^4$
- Kirchhoff's law: efficiency of absorption = efficiency of emission
- IR flux from atmospheric layer = $f\sigma T_1^4$ (up and down)

Radiation Balance Equations



- Balance at top of atmosphere

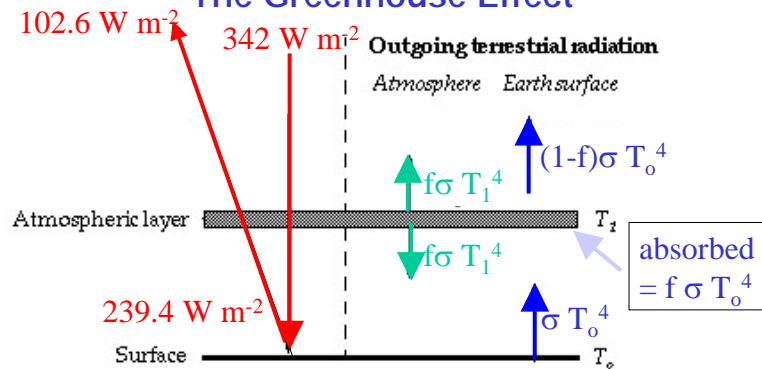
$$f\sigma T_1^4 + (1-f)\sigma T_o^4 = 239.4$$

- Balance for atmospheric layer

$$f\sigma T_1^4 + f\sigma T_1^4 = f\sigma T_o^4$$

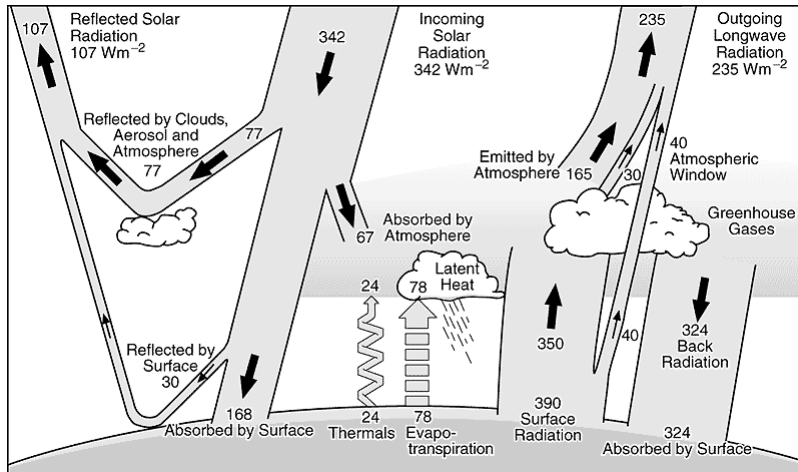
→ Solve for T_o and T_1

The Greenhouse Effect



- For $f=0.77$, $T_o=288 \text{ K}$ and $T_1=242 \text{ K}$ (33 K warmer than the no-atmosphere case)
- As f increases, T_o and T_1 increase
- Greenhouse gases → gases that affect f (absorption efficiency of atmos.)
- Earth has a natural greenhouse effect; human activities enhance effect

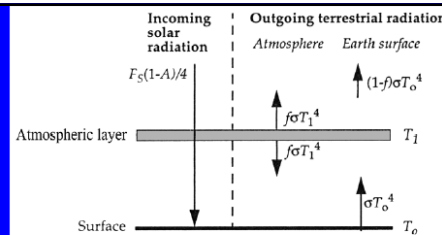
Energy Balance on Earth



IPCC, 2001

What are the differences between this picture and our simple model?

Sequence of Increasingly Complex Models



Division of atmosphere into elemental slabs

"Gray atmosphere" model

Exponential decrease in temperature $T_{\text{TOP}} \sim 210 \text{ K}$

Resolution of spectral absorption into lines and "band models"

Spectrally resolved radiative models

Inclusion of buoyancy, convection, energy balance

Radiative-convective models

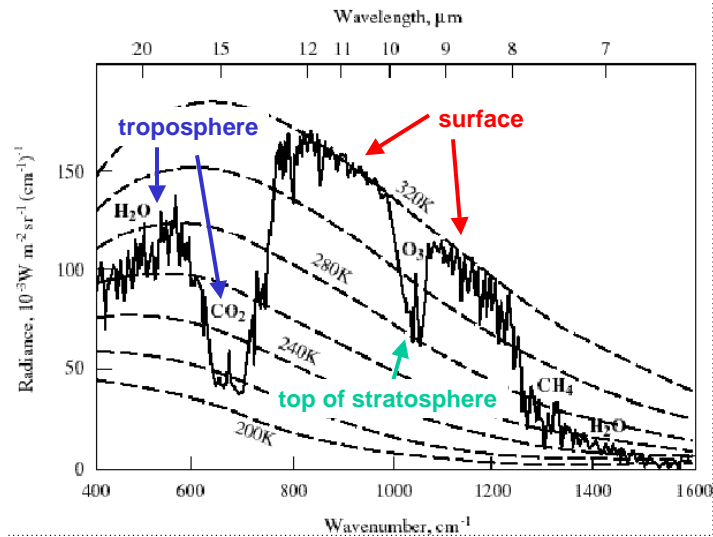
Global, three-dimensional equations for conservation of energy, mass and momentum

General circulation models

The ultimate models for climate research

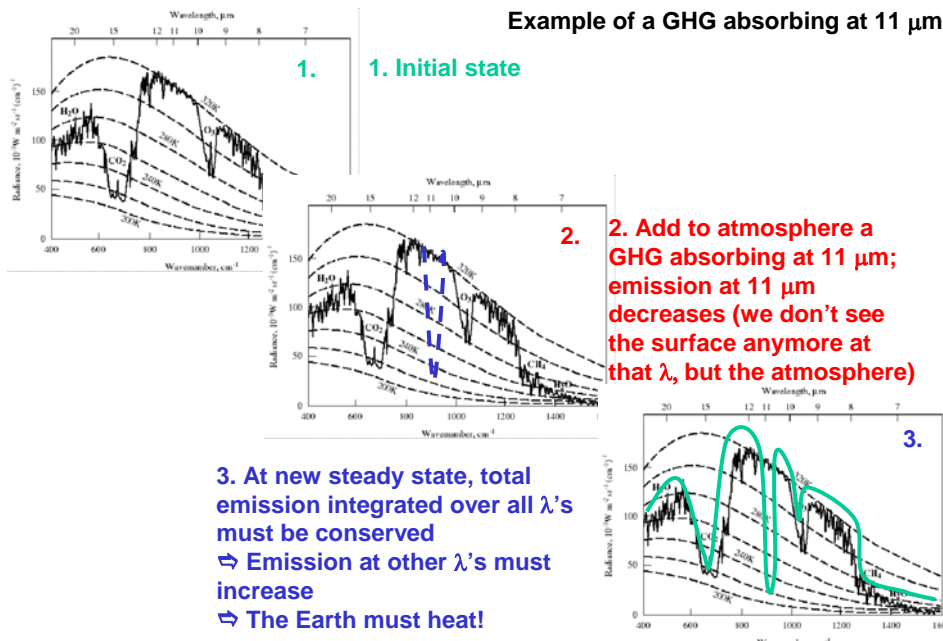
TERRESTRIAL RADIATION SPECTRUM FROM SPACE: composite of blackbody radiation spectra for different T

Scene over
Niger valley,
N Africa



How does the addition of a greenhouse gas warm the earth?

Example of a GHG absorbing at $11 \mu\text{m}$



Concept of Radiative Forcing

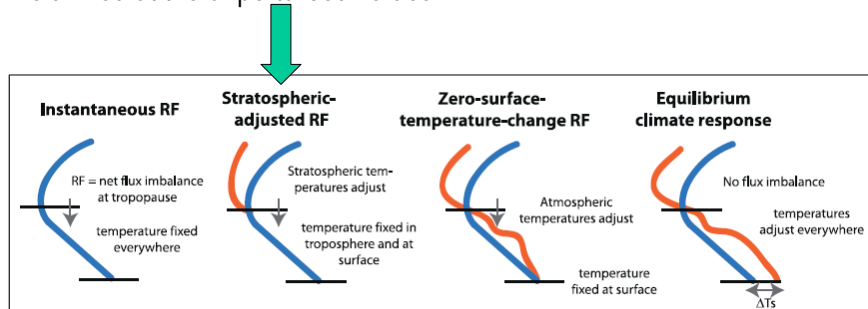
Measure of the climatic impact of a greenhouse gas or other forcing agent

- Radiative forcing = change in radiation balance (flux in minus out) at the top of the atmosphere due to a change in amount of greenhouse gas before the system relaxes to equilibrium. ΔF (W m^{-2})
- Consider atmosphere in radiation balance:
 - If concentration of a greenhouse gas increases and **nothing else changes** → outgoing terrestrial radiation decreases

IPCC definition of radiative forcing

- IPCC definition

"The radiative forcing of the surface-troposphere system due to the perturbation in or the introduction of an agent (say, a change in greenhouse gas concentrations) is the change in net (down minus up) irradiance (solar plus long-wave; in W m^{-2}) at the tropopause AFTER allowing for stratospheric temperatures to readjust to radiative equilibrium, but with surface and tropospheric temperatures and state held fixed at the unperturbed values."

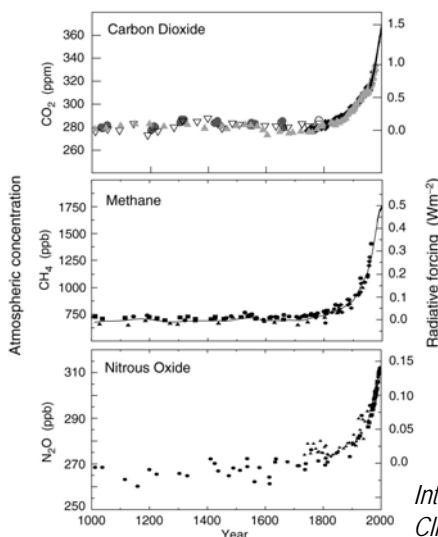


Why is radiative forcing useful?

- Simple measure to quantify and rank the many different influences on climate change.
- Robust. can be calculated with some accuracy
- Additive (globally/regionally/locally)
- Is used to calculate surface temperature changes, but avoids issue of climate sensitivity
- Near-quantitative comparison of anthropogenic forcing agents

Radiative forcing calculations

(a) Global atmospheric concentrations of three well mixed greenhouse gases



Simplified expressions for RF calculations

Gas	Radiative forcing
CO ₂	$F = f(c) - f(c_0)$, where $f(c) = 4.996 \ln(c + 0.0005c^2)$
CH ₄	$F = 0.0406(\sqrt{m} - \sqrt{m_0}) - [g(m, n) - g(m_0, n_0)]$
N ₂ O	$F = 0.136(\sqrt{n} - \sqrt{n_0}) - [g(m, n) - g(m_0, n_0)]$, where $g(m, n) = 0.5 \ln[1 + 2 \times 10^{-5}(mn)^{0.75}]$
CFC-11	$F = 0.264(x - x_0)$
CFC-12	$F = 0.323(y - y_0)$

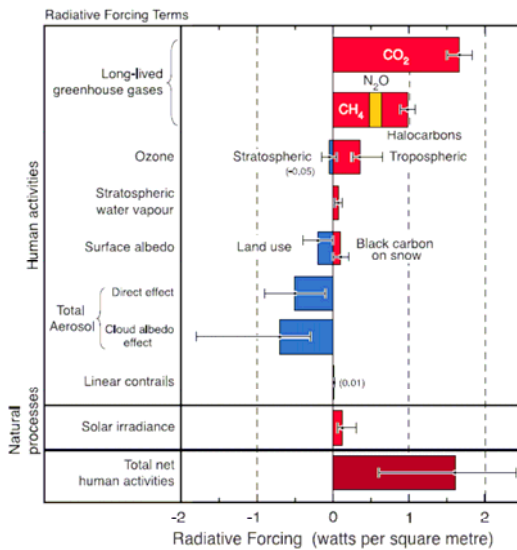
c , CO₂ (ppm); m , CH₄ (ppb); n , N₂O (ppb); x/y , CFC-11/12 (ppb).

Table from Hansen et al., PNAS, vol. 97 (18), 9875-9880, 2000

Intergovernmental Panel on Climate Change (IPCC), 2001.

Human Impacts on the Climate System

Radiative forcing of climate between 1750 and 2006



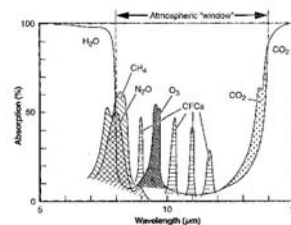
Radiative forcing gives first-order estimate of the relative climatic forcing of anthropogenic gases, aerosols, land-use change

IPCC, Chap 2

GLOBAL WARMING POTENTIAL (GWP): foundation for climate policy

- The GWP measures the integrated radiative forcing over a time horizon Δt from the injection of 1 kg of a species X at time t_o , relative to CO₂:

$$GWP = \frac{\int_{t_o}^{t_o + \Delta t} \Delta F_{1 \text{ kg X}} dt}{\int_{t_o}^{t_o + \Delta t} \Delta F_{1 \text{ kg CO}_2} dt}$$

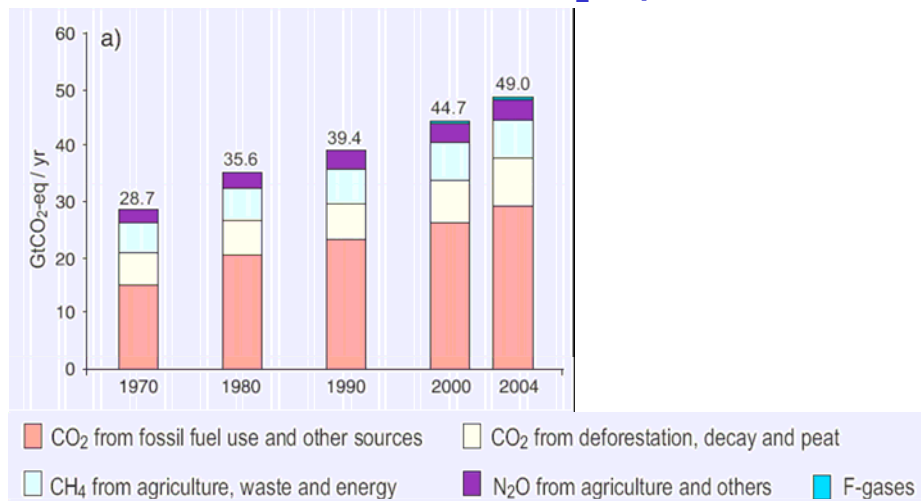


Gas	Mixing ratios (ppm) in 2005	Lifetime (years)	GWP for time horizon		
			20 years	100 years	500 years
CO ₂	379	~200	1	1	1
CH ₄	1.774	12	72	25	7.6
N ₂ O	0.319	114	289	298	153
CFC-12 (CF ₂ Cl ₂)	0.000538	100	11,000	10,900	5,200
SF ₆	0.000056	3200	16,300	22,800	32,600

CO₂ equivalent emissions

- Amount of CO₂ that would have the same radiative forcing as an emitted amount of a GHG for a 100 year horizon
= time integrated radiative forcing
- Example: Emitting 1 million ton of CH₄ (GWP(100 years) = 25) is the same as emitting 25 million tons of CO₂ or 25 Gt CO₂-equivalent.
- Question: are the impacts really the same?

Global anthropogenic GHG emissions in terms of CO₂-eq



IPCC, WGIII Mitigation, TS (2007)

Radiative Forcing and Temperature Change

- Response of system to energy imbalance: $\rightarrow T_o$ and T_1 increase
 \rightarrow may cause other greenhouse gases to change
 $\rightarrow T_o$ and T_1 may increase or decrease depending on internal climate *feedbacks*
 $\rightarrow \Delta f \rightarrow \Delta T \rightarrow$ etc... \rightarrow Ultimately, the system gets back in balance

- Radiative forcing is only a measure of initial change in outgoing terrestrial radiation

- How do we relate radiative forcing to temperature change?

$$\Delta T_o = \lambda \Delta F,$$

ΔT_o : Surface Temperature change ($^{\circ}\text{C}$)

ΔF : Radiative forcing (W/m^2)

λ : Climate sensitivity ($^{\circ}\text{C}$ per W/m^2)

- Climate models (GCMs) indicate that λ ranges from 0.3 to 1.4 $^{\circ}\text{C}$ per W/m^2 . On average $\lambda \sim 0.75$ $^{\circ}\text{C}$ per W/m^2

Feedbacks

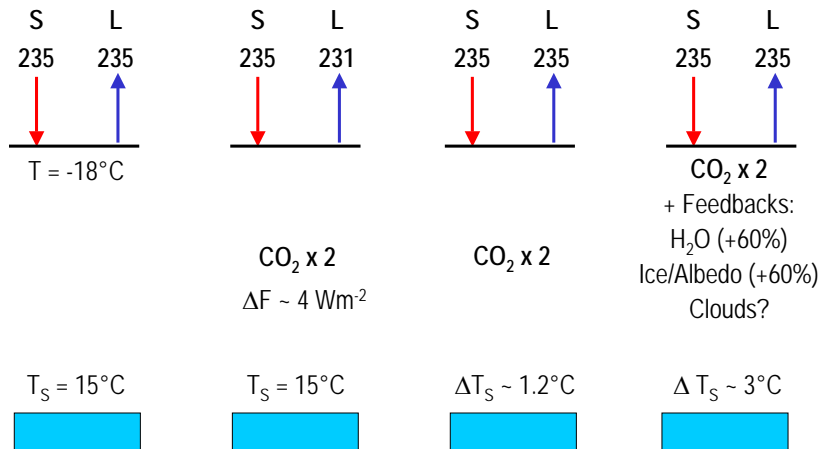
- **water vapor feedback:** positive.



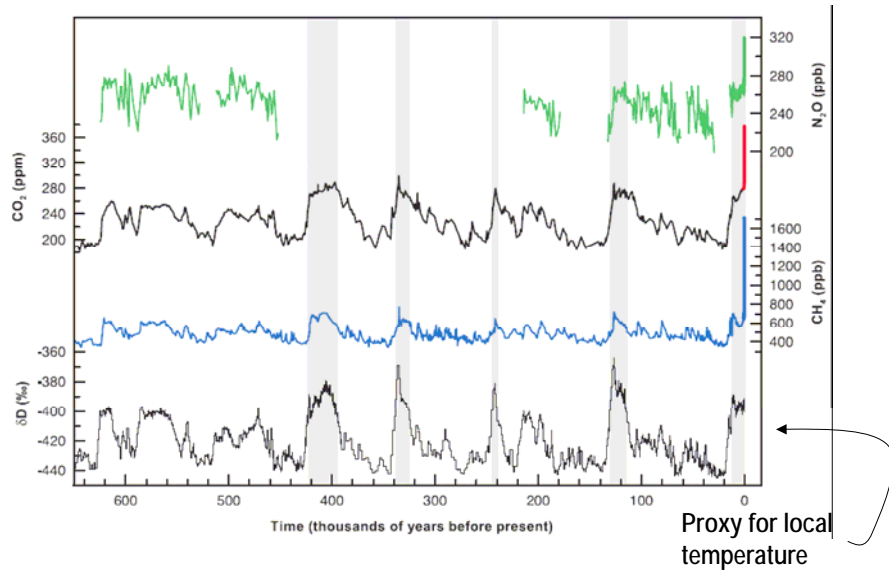
- **ice-albedo feedback:** positive.
- **cloud feedbacks:** positive or negative? potentially large
(clouds can reflect solar radiation or absorb IR radiation depending on their height, thickness and microphysical properties)
- **land surface feedback:** positive (deforestation and hydrological cycle)

Climate response to a doubling of CO₂

Solar (S) and longwave (L) radiation in Wm⁻² at the top of the atmosphere



Past 650,000 years: Glacial-interglacial ice core data



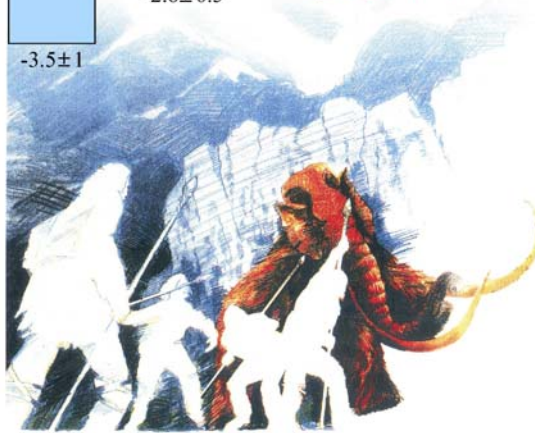
IPCC, TS (2007)

Ice Age Climate Forcings (W/m^2)

**Ice Age Forcings
Imply Global
Climate
Sensitivity**
 $\sim \frac{3}{4}^\circ\text{C per W/m}^2$.

ice sheets & vegetation	greenhouse gases	aerosols
	CO ₂	-0.5 ± 1
	CH ₄	
	N ₂ O	
	-2.6 ± 0.5	
-3.5 ± 1		

Forcing $\sim 6.6 \pm 1.5 \text{ W/m}^2$
Observed $\Delta T \sim 5 \pm 1^\circ\text{C}$
 $\rightarrow \frac{3}{4} \pm \frac{1}{4}^\circ\text{C per W/m}^2$



Source: Hansen et al.,
*Natl. Geogr. Res. &
Explor.*, 9, 141, 1993.