

Physics 115

General Physics II

Session 21

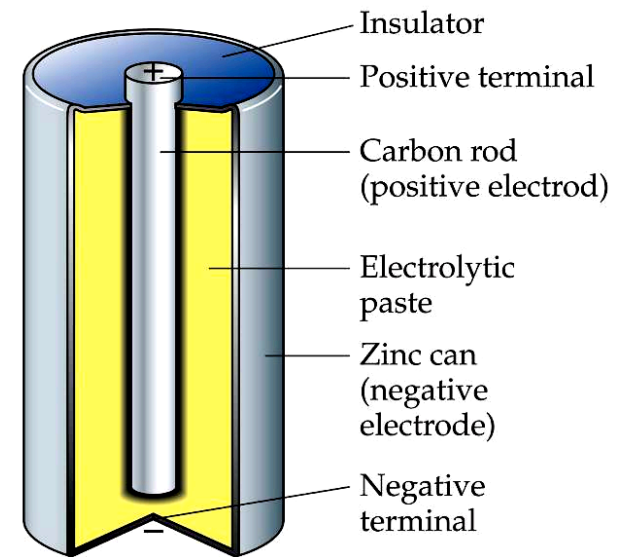
Energy in E fields

Electric Current

Batteries

Resistance

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Lecture Schedule

(up to exam 2)

21-Apr	Mon	12	Specific Heats	18.4-18.6
22-Apr	Tues	13	Second Law	18.7-18.10
24-Apr	Thurs	14	Entropy	18.8-18.10
25-Apr	Fri	15	Charges	19.1-19.4
28-Apr	Mon	16	E field	19.5-19.66
29-Apr	Tues	17	Gauss law	19.7
1-May	Thurs	18	Electrical potential	20.1-20.3
2-May	Fri	19	Potential, conductors	20.4
5-May	Mon	20	Capacitors	20.5-20.6
6-May	Tues	21	Current	21.1-21.2
8-May	Thurs	22	Power, Series & Parallel Circuits	21.3-21.4
9-May	Fri		EXAM 2 - Ch. 18,19,20	

Today

Announcements

- Exam 2 is this Friday 5/9
 - Covers material discussed in class from Chs 18, 19, 20
 - NOT Ch. 21
 - Same format and procedures as last exam
 - If you arranged to take exam 1 with section B, please do same for all remaining exams, OR email us to say you want to change
 - Practice questions have been posted in slides directory - we will review them in class Thursday

Last time Energy Storage in a Capacitor

In capacitors, charge is stored on electrodes with potential difference ΔV . It takes **work** to move charge against the E field represented by ΔV !

The first bit of charge is easy to move: for an uncharged capacitor, $V=0$

Thereafter each bit of charge takes more work: V grows linearly with total Q on the capacitor, since $V=Q/C$.

The stored charge represents the work done, in potential energy: $U = Q \Delta V$

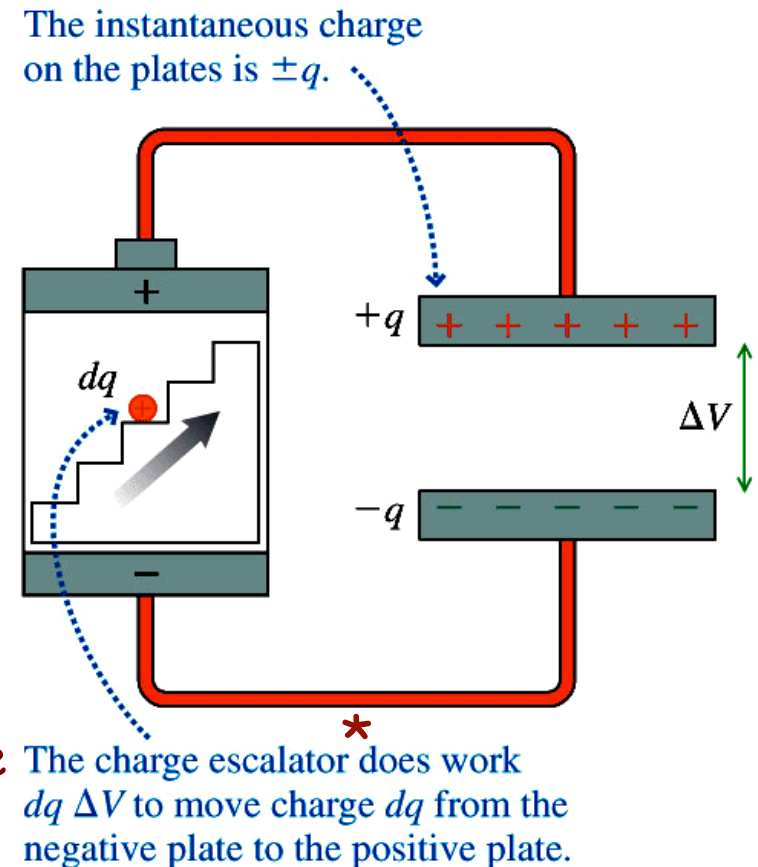
Using calculus we find the total work done is

$$W_{TOTAL} = U_C = \frac{1}{2} Q \Delta V_C$$

$$C = \frac{Q}{\Delta V_C} \rightarrow U_C = \frac{1}{2} C \Delta V_C^2 = \frac{Q^2}{2C}$$

Or, without calculus: since V grows linearly with total Q , average $V = \frac{1}{2} Q/C$, so total $W = QV_{AVG} = \frac{1}{2} Q^2 / C$

* What's this "charge escalator"? A source of energy (e.g, a battery) that "lifts" charge through the potential difference (against E force)



Energy **density** in an electric field

- The energy stored by a capacitor is the energy content of its electric field

$$U_C = \frac{1}{2} Q \Delta V_C, \quad Q = C \Delta V_C \rightarrow U_C = \frac{1}{2} C \Delta V^2$$

$$\frac{1}{2} C \Delta V^2 = \frac{1}{2} \left(\frac{\epsilon_0 A}{d} \right) (Ed)^2 = \frac{\epsilon_0}{2} (Ad) E^2$$

(Ad) = volume of space between plates

$$\text{energy density } u_E \equiv \frac{\text{energy stored}}{\text{storage volume}} = \frac{U_C}{Ad} = \frac{1}{2} \epsilon_0 E^2$$

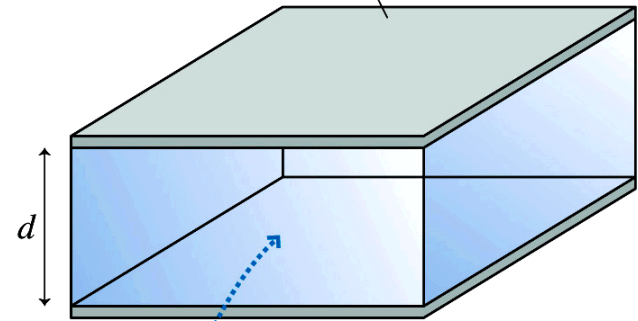
Example: capacitor has $d=1.0 \text{ mm}$, $\Delta V_C=500 \text{ V}$

We don't need to know A or Q :

$$E = \frac{\Delta V_C}{d} = \frac{500 \text{ V}}{1.0 \times 10^{-3} \text{ m}} = 5.0 \times 10^5 \text{ V/m}$$

$$\text{So } u_E = \frac{1}{2} \epsilon_0 E^2 = \frac{1}{2} (5.0 \times 10^5 \text{ V/m})^2 / (4\pi \times 9.0 \times 10^9 \text{ Vm/C}) = 1.1 \text{ J/m}^3$$

Capacitor plate with area A



The capacitor's energy is stored in the electric field in volume Ad between the plates.

$$u_E = \frac{1}{2} \epsilon_0 E^2$$

True for **any** region of space with **E** , not just inside capacitors

Energy in the field of a spherical conductor

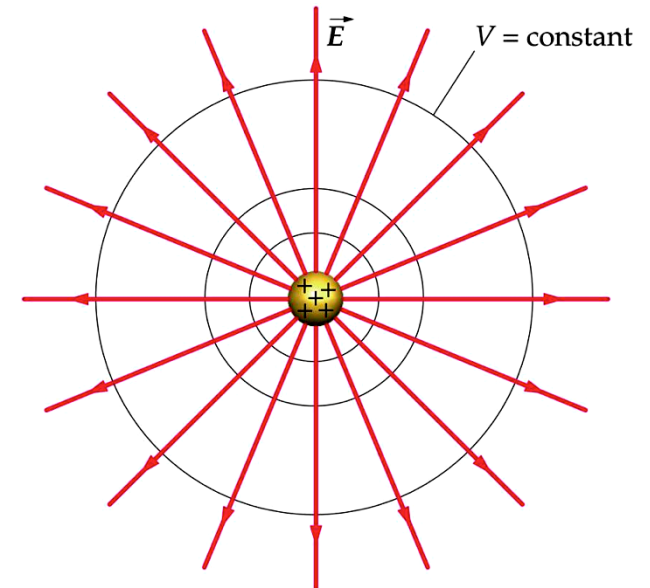
- For an isolated, spherical conductor with charge $+Q$ and radius R

$$E_{\text{outside}} = \frac{kQ}{r^2}, \quad V_{\text{surface}} = \frac{kQ}{R}, \quad \text{for } V = 0 \text{ at } \infty$$

$$C = Q / V = R / k \quad \leftarrow \text{Capacitance of an isolated sphere (other "electrode" is at infinity)}$$

$$U = \frac{1}{2} \frac{Q^2}{C} = \frac{kQ^2}{2R}$$

The energy in the electric field around the sphere can be regarded as the energy stored in its capacitance, relative to a (fictional) negative electrode at $R = \infty$.



Deep thought: Notice when $R \rightarrow 0$, $U \rightarrow \infty$.
So a true point charge should have *infinite energy*!
As far as we know, electrons are point particles...
(the Self-Energy Problem - how is an electron possible?)

Demo: dielectric 'wants to go into' capacitor gap

Charge up a parallel plate capacitor (so Q on plates is fixed)

Plastic dielectric slab is pulled into the gap – why?

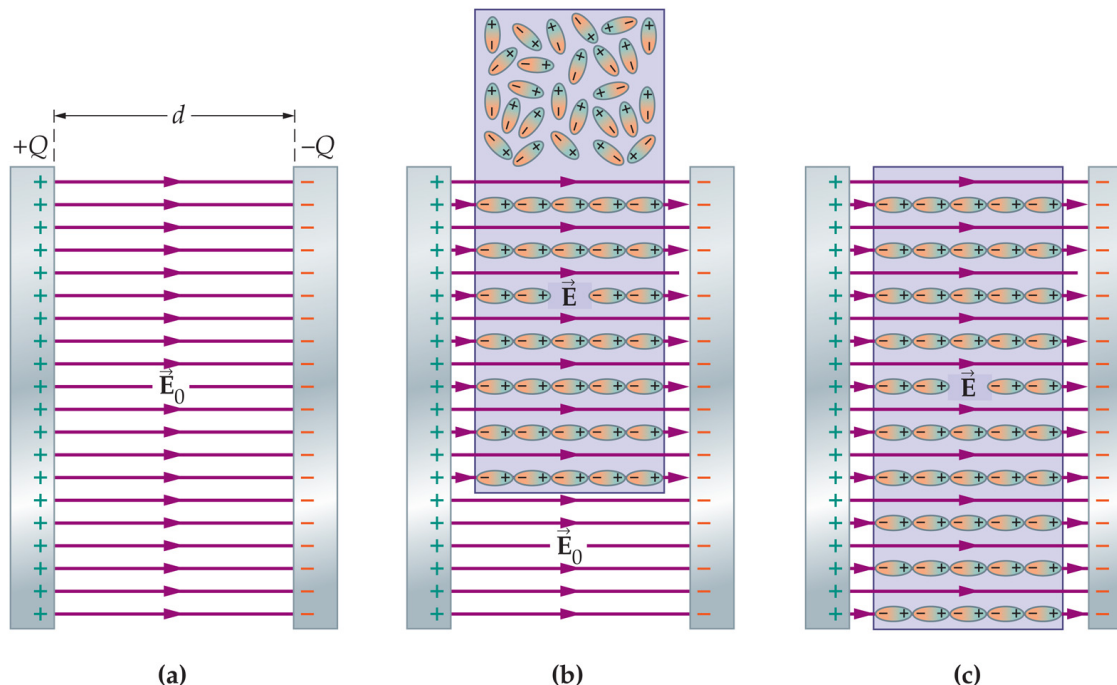
For fixed Q , more energy is stored in air-gap capacitor ($\kappa = 1.0$) than in dielectric-gap capacitor ($\kappa > 1$):

$$U_{AIR} = \frac{Q^2}{2C}, \quad U_D = \frac{Q^2}{2\kappa C} < U_{AIR}$$

Two ways to explain:

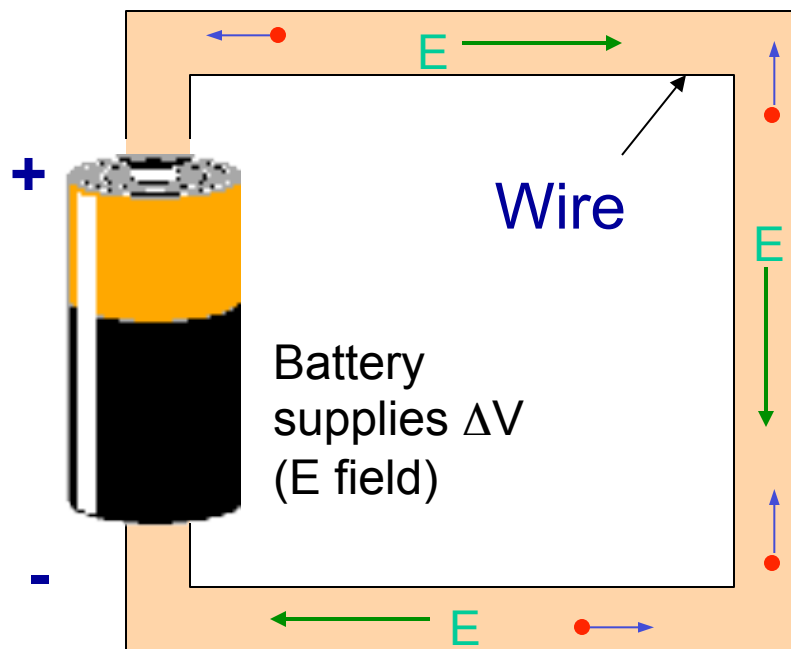
1. The polarized dielectric is attracted to the plates of the capacitor
2. Energy of system is reduced with dielectric in place (E field is reduced, so energy density is lower)
 - Systems move to lower energy states spontaneously

Did energy just disappear?
Where did the energy go?



Electrodynamics: Electric *Currents* and *Circuits*

- Wires (conductors) channel and contain electric fields
- Battery provides a source of potential difference
- Fields point *away* from positive terminal, towards negative
- We imagine **positive** charge flowing in direction of field lines
 - **Actually**, electrons (-) flow in *opposite* direction (Ben Franklin's error!)



- **Electrons' actual direction of motion**
- E** → **Electric field direction**
("conventional current")

Electric Current = flow of charge
1 **ampere** (A) = 1 C / sec

$$I = \frac{\Delta Q}{\Delta t}$$

Andre
Ampère,
1775-1836



Current and voltage

- We say the battery's \mathbf{E} field supplies an *electric potential* (or *Electromotive Force*, EMF) to charges in the conducting wires

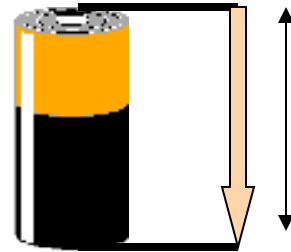
Useful analogy: electric current is like flow of water

- Voltage = **pressure** causing flow
- Current = **rate** of flow



Analogy to water flow
due to gravity

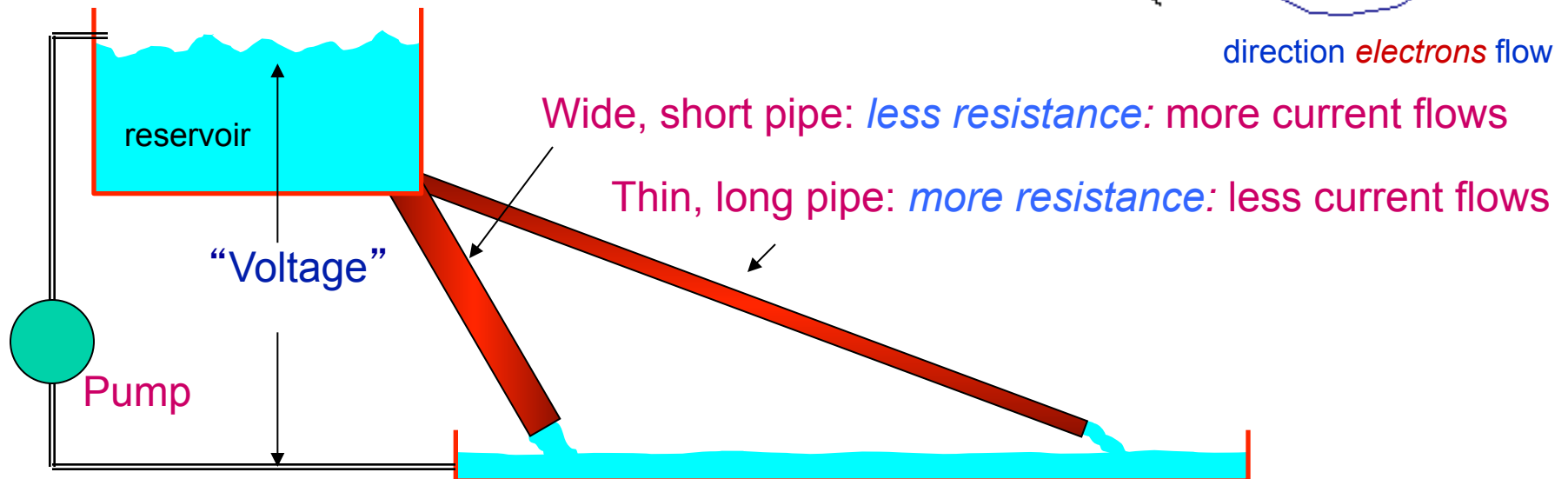
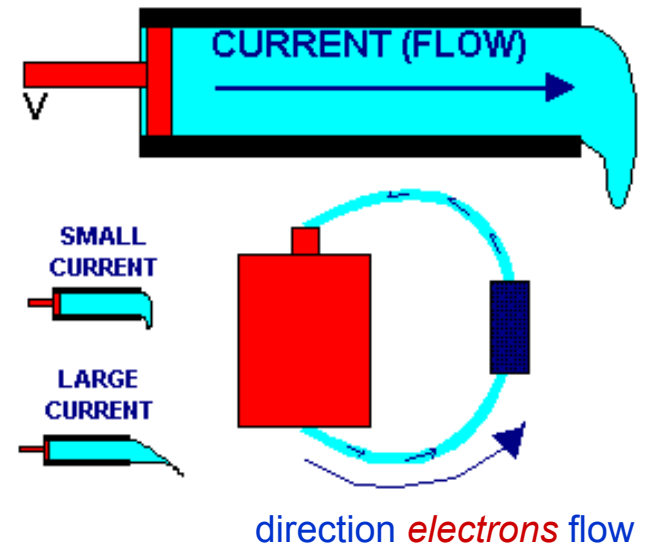
$$\begin{aligned} gh &= \text{PE per kg} \\ \text{kg/sec} &= \text{flow rate} \\ \text{Power} &= (\text{kg/sec}) \cdot g \cdot h \end{aligned}$$



$$\begin{aligned} V \text{ (volts)} &= \text{PE per coulomb} \\ I \text{ (C/sec=amps)} &= \text{flow rate} \\ \text{Power} &= V \cdot I \end{aligned}$$

Electrical voltage : current *as* Water pressure : current

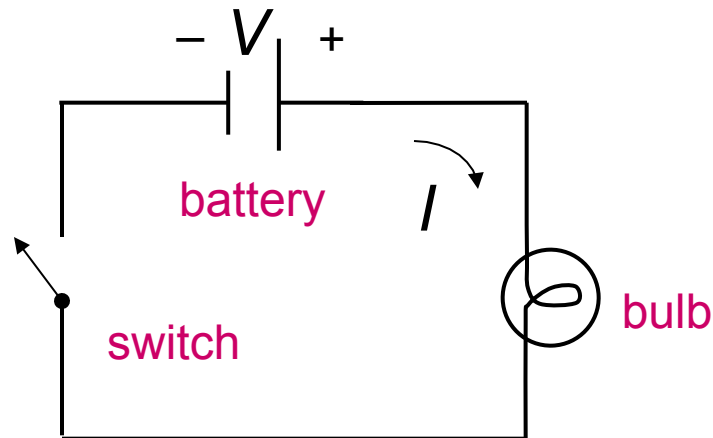
- Can think of electric current like water flow in a closed system
 - high current: lots of water per second
 - low current: trickle of water
 - current flows around a *loop* (circuit) of pipe
 - battery provides *pressure* to make water flow
- Battery is like reservoir of elevated water
 - Higher tank = bigger *potential*, or voltage
- Imagine wires as tubes that let water drain
 - *Resistance* depends on length and diameter



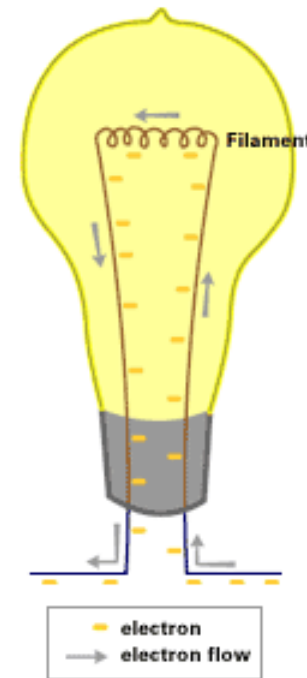
Need to make a *complete circuit* (path for charge flow)

- Must provide a *closed path* for current to flow through bulb under “pressure” from battery
 - Charge cannot just disappear! Battery has to have electrons returned
- Any gap in path *interrupts* current flow
 - We call that a *switch*
 - Acts like valve in water system

In “circuit diagram” form:



In a closed circuit,
current flows *around* the loop.
Switch *interrupts* flow, turns off bulb.



Current flowing through the high resistance filament heats it, and makes it white-hot.

Batteries and Electro-Motive Force (EMF)

Battery = **chemical source** of electric energy. Chemical reactions **create** potential difference by moving positive ions to one electrode and negative ions to the other.

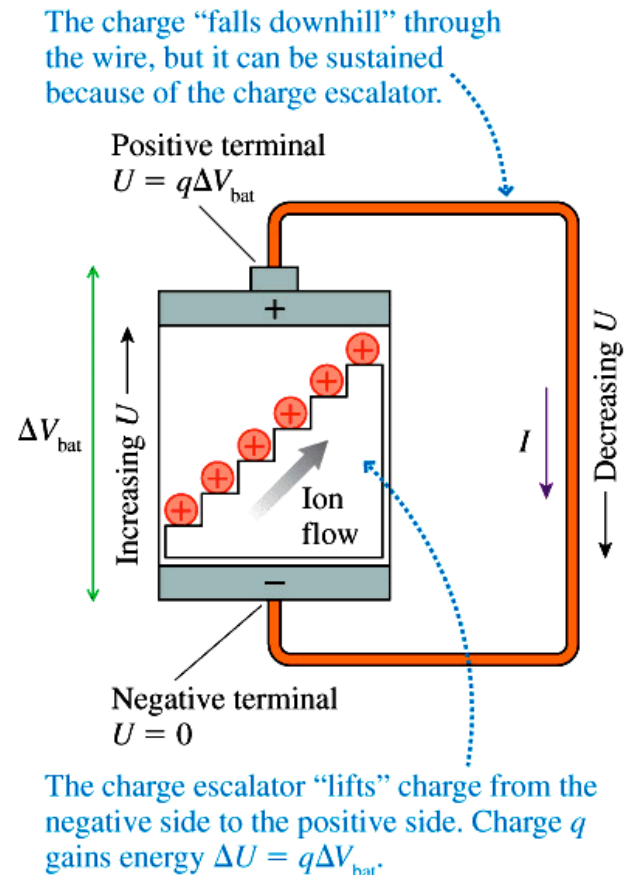
The potential difference ΔV_{bat} is determined by the **chemistry** of the battery (e.g., carbon and zinc in an old-fashioned dry cell)

ΔV_{bat} remains fairly constant until the chemicals are exhausted - the battery goes “dead”.

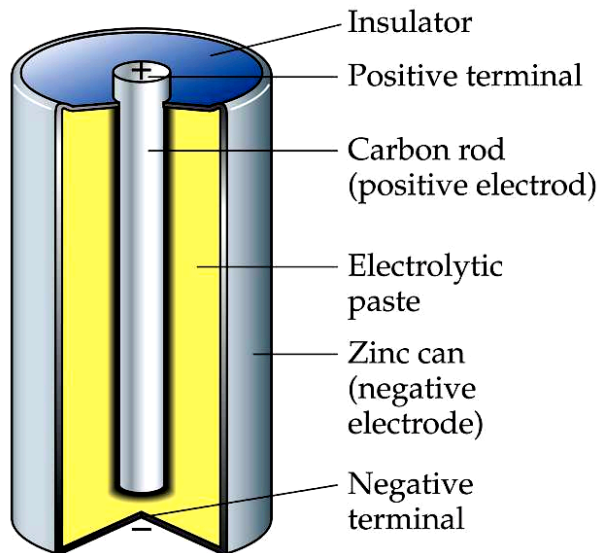
The term EMF (**electromotive force**), symbol \mathcal{E} , is used to describe the **work done per unit charge** by the battery: $\mathcal{E} = W_{\text{chem}}/q = \Delta V_{\text{bat}}$.

Remember: no force involved! EMF is just the potential difference maintained by a source.

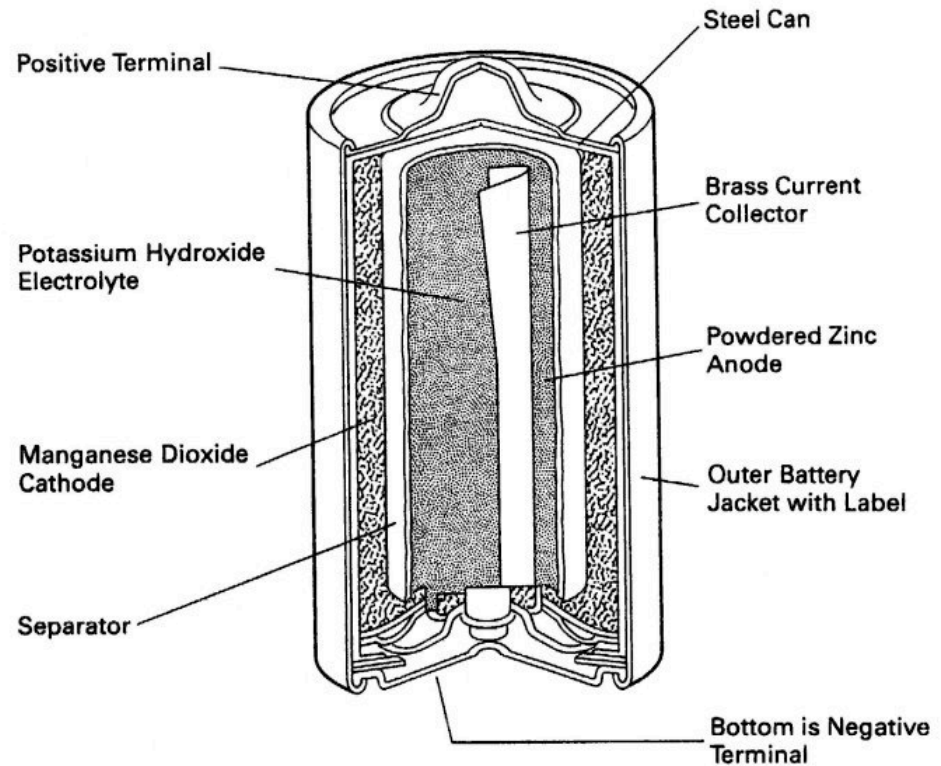
(A real battery has “internal resistance” that increases as the chemicals are used up, and limits current flow - more on this later)



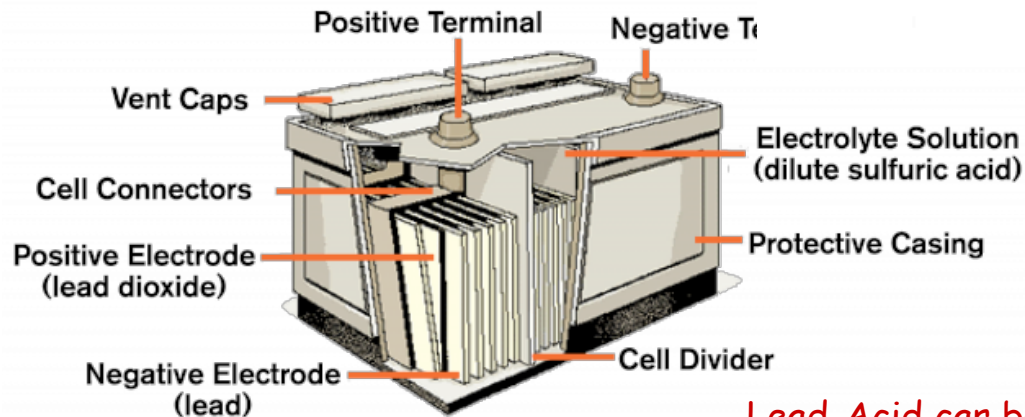
Inside batteries



Carbon-zinc
("dry cell": G. Leclanché, 1866)



"alkaline battery":
Cathode (-) = Zinc powder in alkaline gel, carbon-manganese anode (+)
Lewis Urry, c. 1950



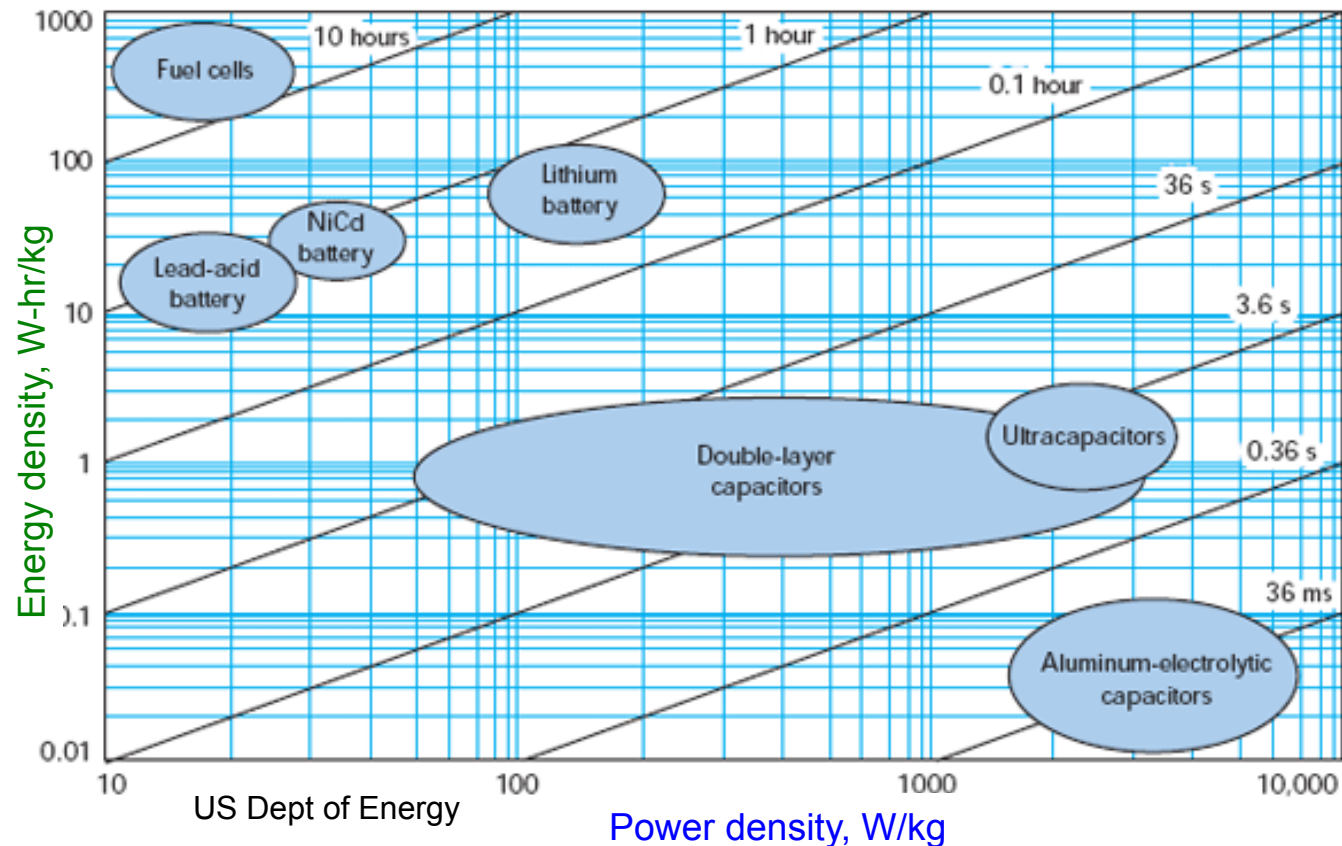
Lead-Acid car battery
("wet cell": 1859, Gaston Planté,)

BTW: Storing electrical energy is a hot topic

- Batteries are a major issue for “green” vehicles
- Two aspects to energy storage: how **much**, how **fast is it needed?**
 - **Energy** density = **joules** / kg (how much energy)
 - **Power** density = **kW** / kg (how fast energy can be delivered)

Batteries →
Lots of energy,
moderate
energy/sec for
a long time

Capacitors →
less total
energy but
available very
quickly



Current, voltage and resistance

- Conventional current I = flow of + charge
 - Really: electrons move in opposite direction
 - Electrons are very light, have to diffuse their way through atoms
 - Follow a long random walk from one end of battery to other!
 - May take an hour for a given electron to go through circuit
 - But bulb lights up right away...?
 - Like a garden hose: flow starts right away, but you must wait to get cold water (= water parcel just out of faucet)

- Current is proportional to V , inversely proportional to R

$$I \propto V \Rightarrow I = \text{const}(V) = \left(\frac{1}{R}\right)V$$

R = Resistance of circuit element

Unit of R = Volt per ampere = Ohm (Ω)

R is a property of a given object (device, wire, circuit element)

Ohm's Law

$$V = I R$$

Conductivity and Resistivity

Resistivity ρ -> intrinsic property of *material* -> like *density* vs *mass*
 Resistance R -> property of a particular *object*

$$R = \rho \frac{L}{A}$$

Resistivity units: Ohm-meters

*BTW: the inverse of resistance is conductance; its unit is the MHO * (no kidding) (Ω) and the inverse of resistivity is*

$$\sigma = \text{conductivity} = \frac{1}{\rho} \quad (\text{mho} / \text{m})$$

* Official SI unit is the siemens (S) but use of the mho cannot be suppressed...

TABLE 28.2 Resistivity and conductivity of conducting materials

Material	Resistivity ($\Omega \text{ m}$)	Conductivity ($\Omega^{-1} \text{ m}^{-1}$)
Aluminum	2.8×10^{-8}	3.5×10^7
Copper	1.7×10^{-8}	6.0×10^7
Gold	2.4×10^{-8}	4.1×10^7
Iron	9.7×10^{-8}	1.0×10^7
Silver	1.6×10^{-8}	6.2×10^7
Tungsten	5.6×10^{-8}	1.8×10^7
Nichrome*	1.5×10^{-6}	6.7×10^5
Carbon	3.5×10^{-5}	2.9×10^4

*Nickel-chromium alloy used for heating wires

Resistivity is temperature dependent, and Ohm's linear V vs I relation works only approximately for many materials.
 (Some circuit devices are specifically designed to be non-linear: transistors, capacitors... More later)

Clicker Quiz 14

- Which one of the following is **correct**?
 - A. If we put a sheet of dielectric ($k > 1$) into a parallel plate capacitor's air gap, the capacitance gets **smaller**
 - B. Resistance of a given wire does **not** depend upon the material of the conductor
 - C. The “ohm” is a joke devised by MIT students, not an accepted physical unit for resistance.
 - D. **All of the above are incorrect**