Paleoclimate and Ice Ages I: Ice core science and climate reconstruction

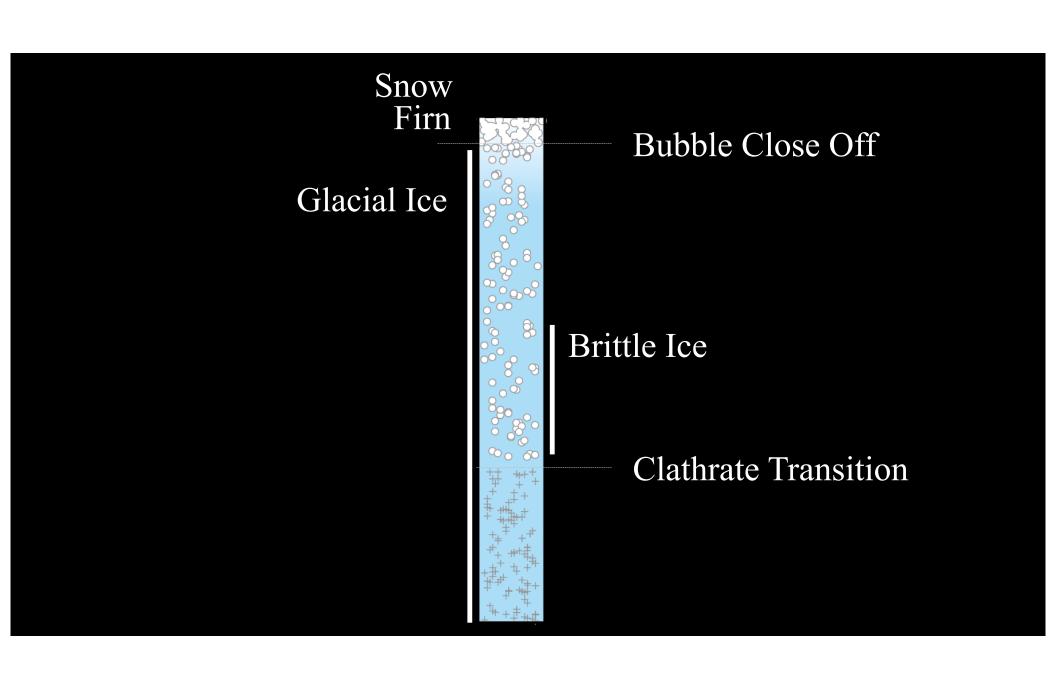
ESS431: Principles of Glaciology

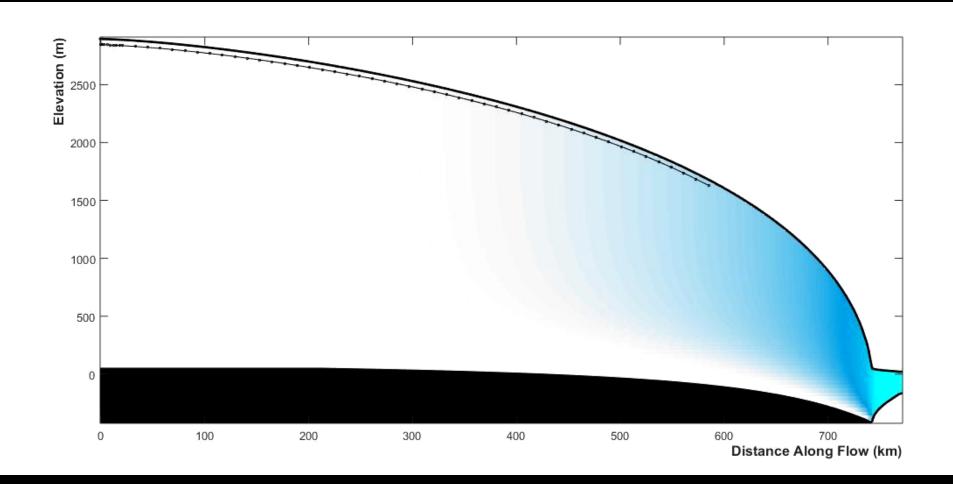
ESS505: The Cryosphere

Friday, 11/16 – Knut Christianson

Today's Objectives:

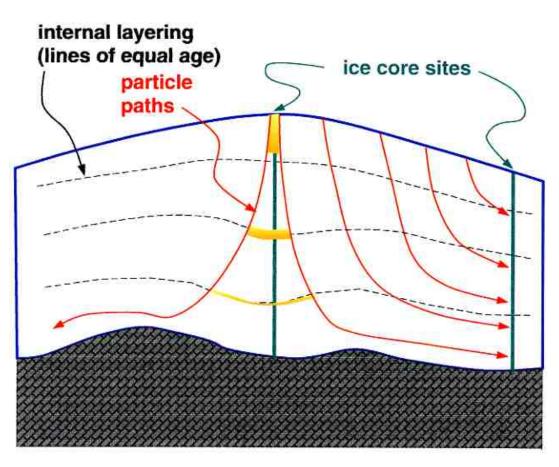
- What are ice cores? How are they collected, and how are their sites chosen?
- What types of information can ice cores provide about past climate?
- What are the challenges of interpreting the climate proxies?





Choosing an Ice-Core Site

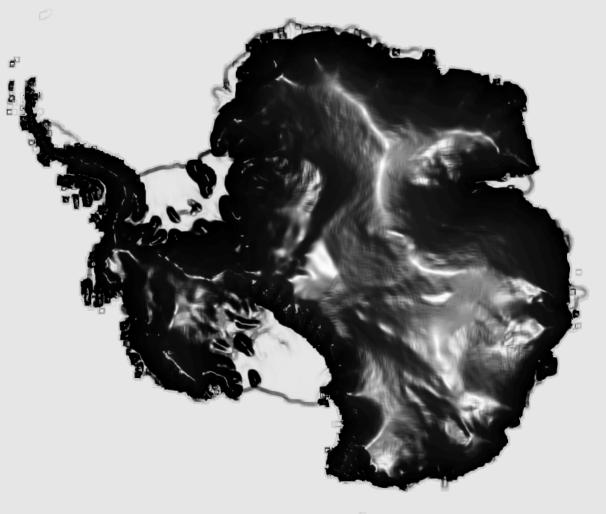
Near the summit is good.



• Climate signal relates to the same place for ice of all ages.

Layers get thinner over time due to flow.





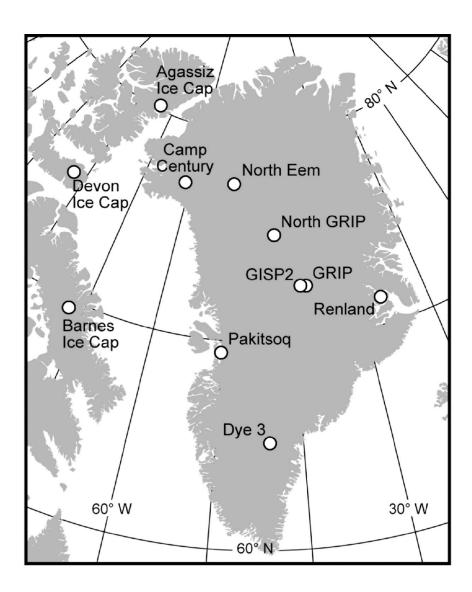
Collecting an Ice Core



Arctic Icecore Sites

Cuffey and Paterson 2010. *The Physics of Glaciers*. Fig. 15.1

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Ice Core at Greenland Summit

- Greenland Ice Sheet Project 2 (GISP2)
- The drill inside this dome reached bedrock 3 km (10,000 ft) below the ice-sheet surface.
- The science team recovered a record of past climate going back more than 100,000 years.





Inside the GISP2 Drill Dome

- The drill is lowered into the hole on a long cable.
- It brings up a 6-meter-long piece of ice core on each run.





Science in a Labyrinth

• Ice cores are analyzed in Labs connected by tunnels under the snow.



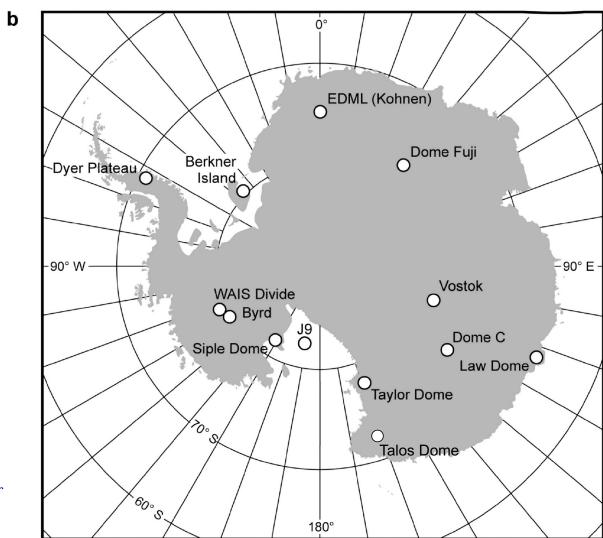
A Natural Freezer

- The cores are stored in a cave dug in the snow, where the temperature is always -30° Celsius.
- ooops ... the ceiling is sagging a little bit after 3 years ...

EGRIP



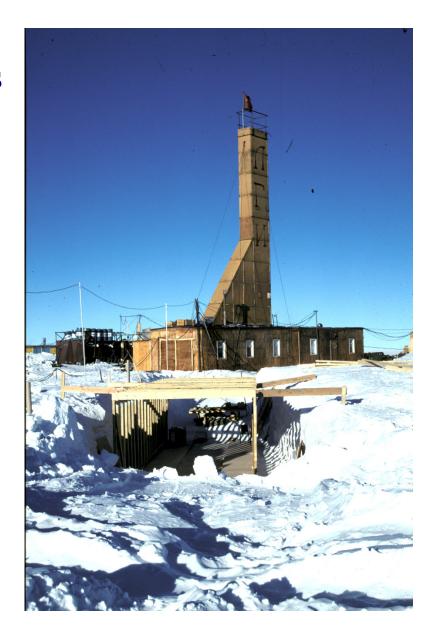




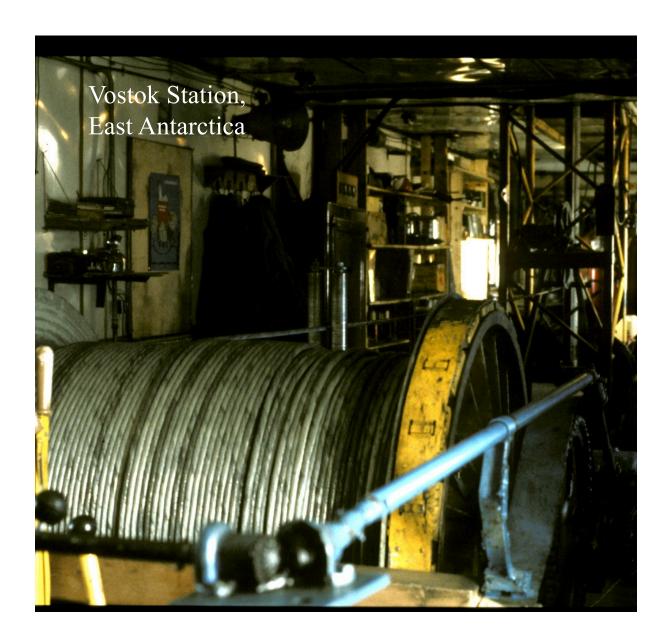
Base map: Cuffey and Paterson 2010. *The Physics of Glaciers*. Fig. 15.1

Where might this be?

Vostok Station, East Antarctica



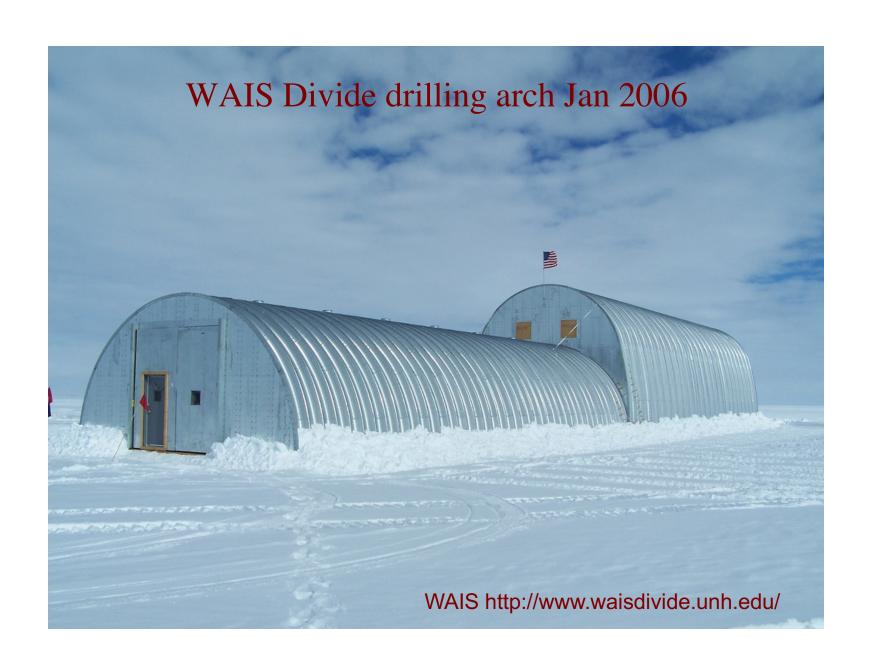
S.G. Warren photo



S.G. Warren



S.G.Warren



WAIS Divide drilling arch December 2014



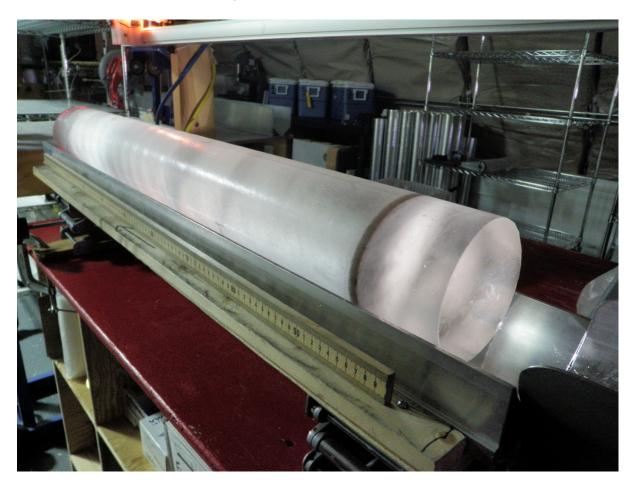
WAIS Tilting-Tower drill



WAIS http://www.waisdivide.unh.edu/



Ash layer in WAIS core



WAIS http://www.waisdivide.unh.edu/

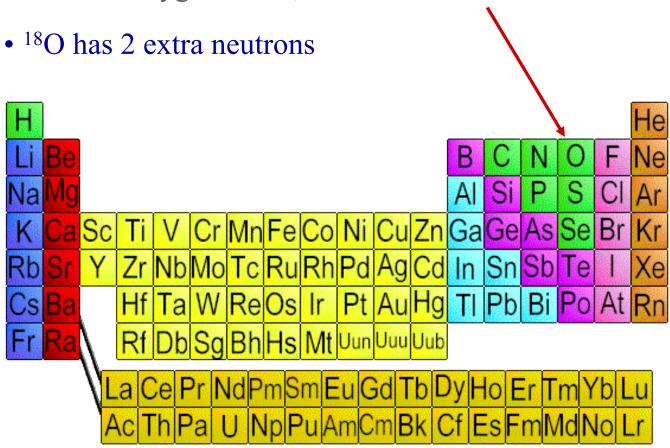
Measurements from the Core and Borehole

Core Measurements

Borehole Logging

^{18}O – What is it?

• Normal Oxygen is ¹⁶O, with 8 neutrons.



Oxygen isotopes in ice cores

H₂O is composed of both H₂¹⁶O and H₂¹⁸O

$$\delta^{18}O_{ice} = \frac{\left(\frac{H_2^{18}O}{H_2^{16}O}\right)_{ice} - \left(\frac{H_2^{18}O}{H_2^{16}O}\right)_{SMOW}}{\left(\frac{H_2^{18}O}{H_2^{16}O}\right)_{SMOW}} * 1000 \% o$$

Deuterium isotopes in ice cores

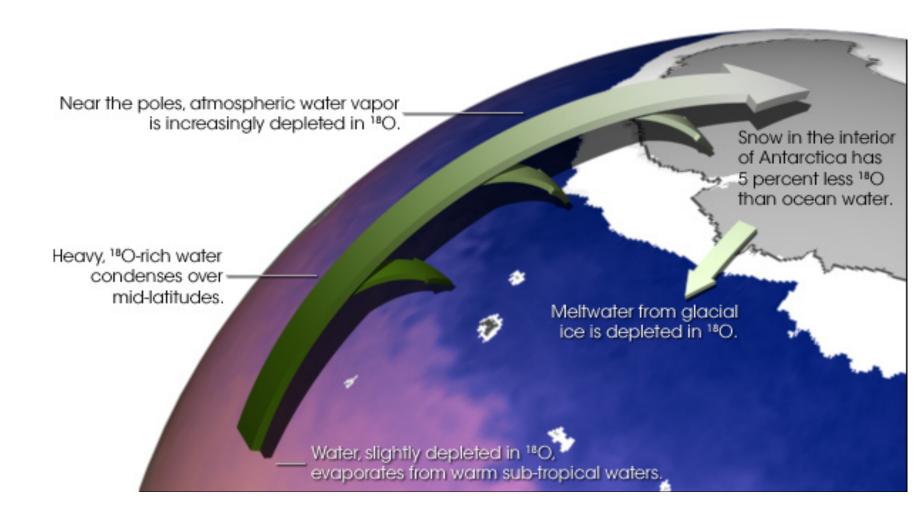
H₂O also contains heavy hydrogen (²H), also known as deuterium (D).

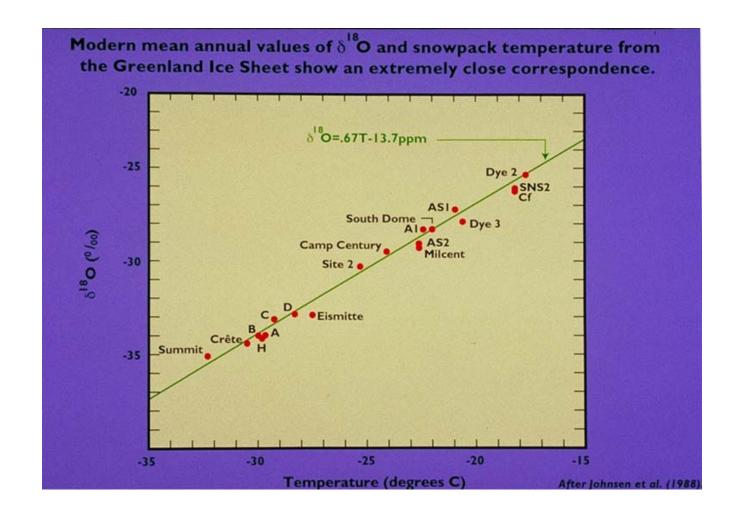
$$\delta D_{ice} = \frac{\left(\frac{D}{H}\right)_{ice} - \left(\frac{D}{H}\right)_{SMOW}}{\left(\frac{D}{H}\right)_{SMOW}} *1000 \% o$$

What is fractionation?

- ¹⁸O and ¹⁶O act the same way in chemical reactions (they are both oxygen molecules)
- They differ in subtle ways during phase changes.
- ¹⁸O has a slightly greater tendency to be in the lowerenergy state.
- During evaporation, the vapor is slightly depleted in ¹⁸O relative to the water.
- During condensation, the condensing water is slightly enriched in ¹⁸O relative to the vapor.

δ^{18} O as a Temperature Proxy

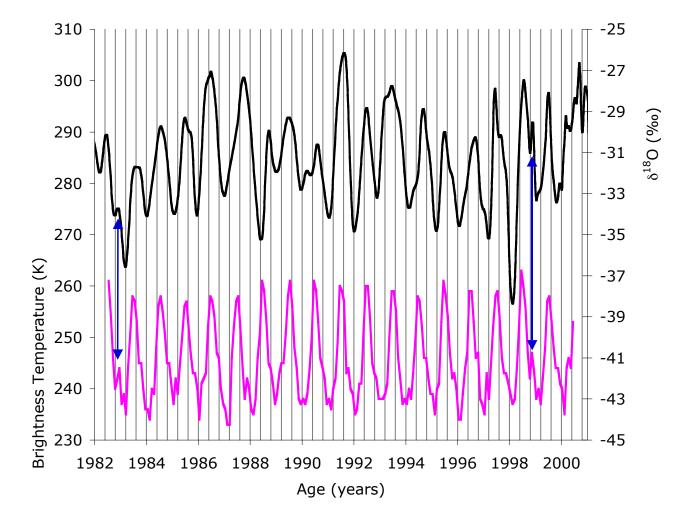




Rayleigh Distillation

Start with a saturated air mass above the ocean surface Assume that:

- 1) As the air travels poleward, it cools adiabatically.
- 2) The amount of water vapor remaining in the air at any point is determined by the saturation vapor pressure of water.
- 3) All isotopic exchange occurs in equilibrium.
- 4) Once water condenses, it does not re-evaporate.
- 5) No new water is added to the air mass.



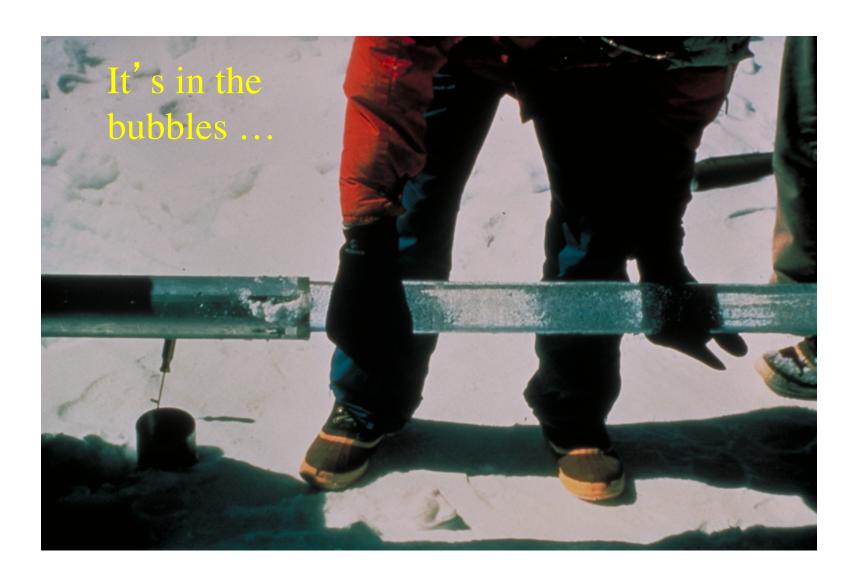


UW Stable Isotope Lab

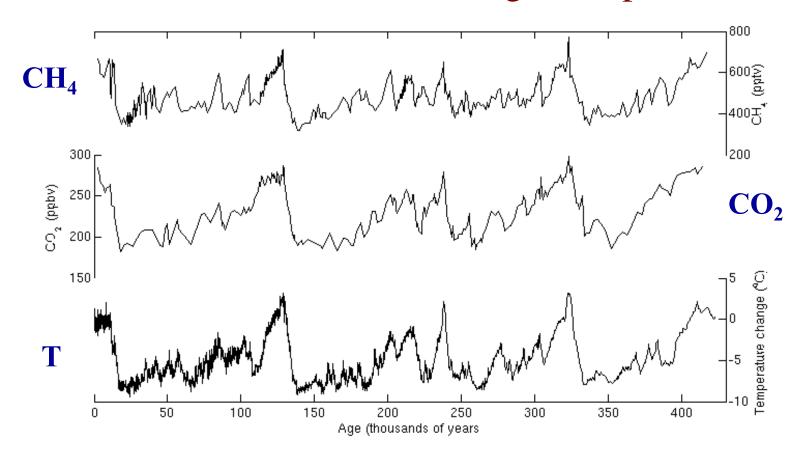
- All the comforts of the UW campus at –30 degrees C.
- These scientists will determine how cold it was in Greenland in the Last Ice Age.

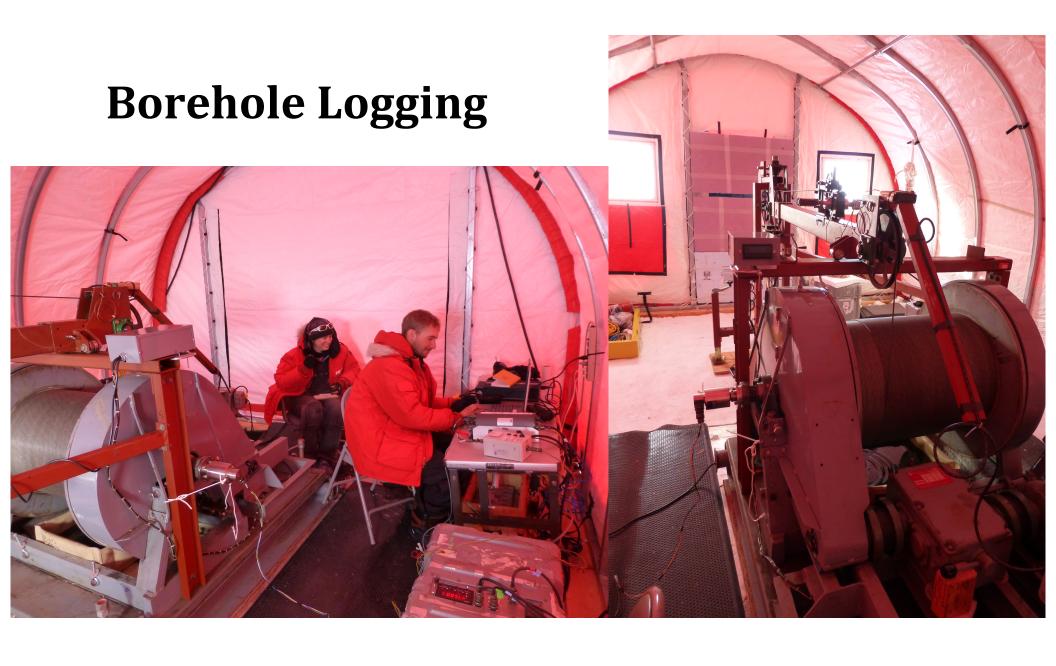
Ice holds samples of our past atmosphere





Greenhouse Gases and Ice-Age Temperature





Borehole Logging

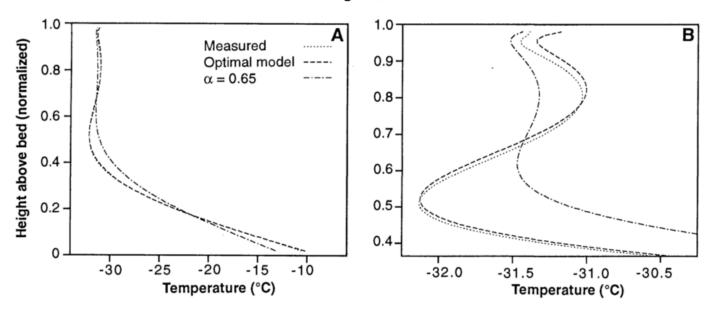
Lots of Glaciologically interesting variables

- Ice Fabric
- Dust content of the layers
- Acoustic / Electromagnetic properties
- Borehole size

Precise measurements of Ice Temperature

Large Arctic Temperature Change at the Wisconsin-Holocene Glacial Transition

Kurt M. Cuffey,* Gary D. Clow, Richard B. Alley, Minze Stuiver, Edwin D. Waddington, Richard W. Saltus

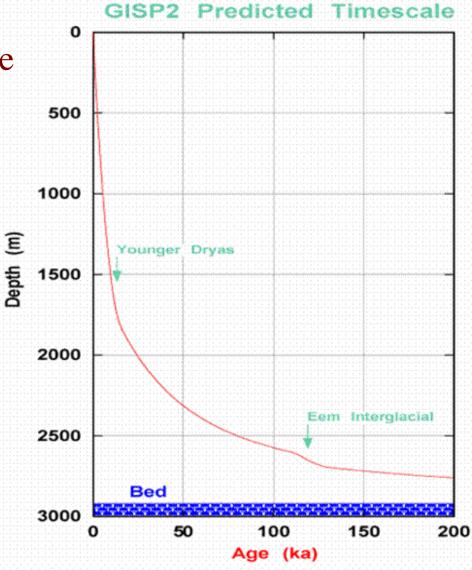


Dating the Ice Core

- Model-based time scales
- Layer counting
- Matching well-dated cores with cores elsewhere (gas-based or dust chronologies)

GISP2 depth-age

Predicted by an ice-flow model before the core was drilled, with a good guess at the history of net snowfall.



Layer counting - the concept

Seasonal timing of parameters being measured

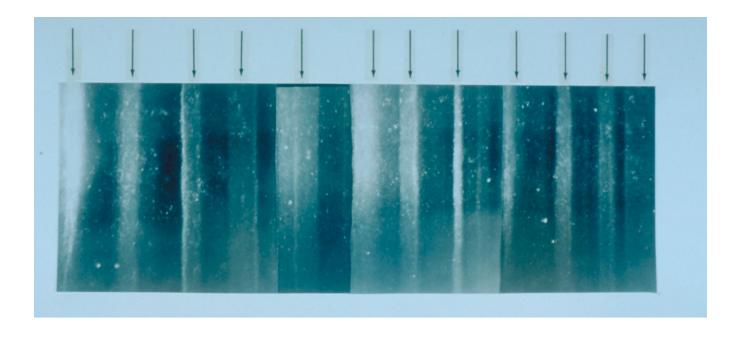
- e.g. δ^{18} O higher in summer than winter
- Dust varies seasonally with exposed bare ground (sources) and with wind speed (transport).
- Many others ...

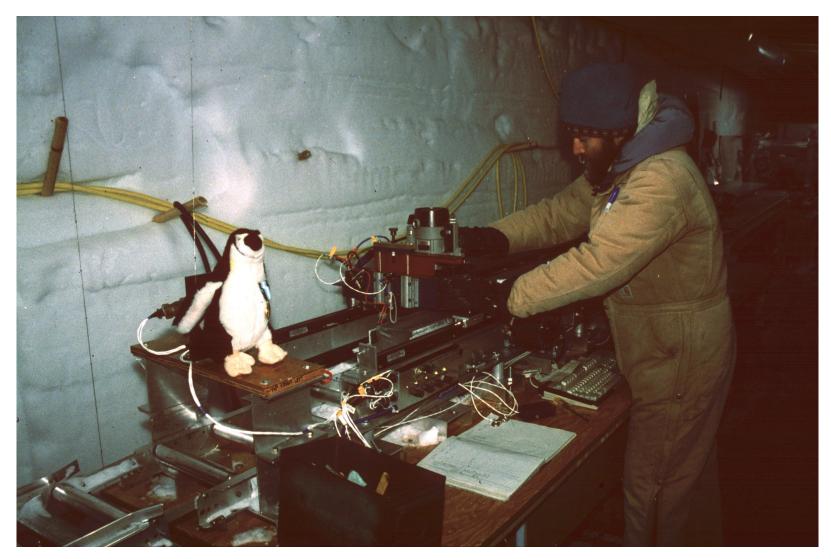
Validation/correction with stratigraphic markers

• e.g. volcanic events of known age (rocks dated at the volcano)

Layer counting - the practice

Summer layers, GISP2, Greenland ~20 kyr

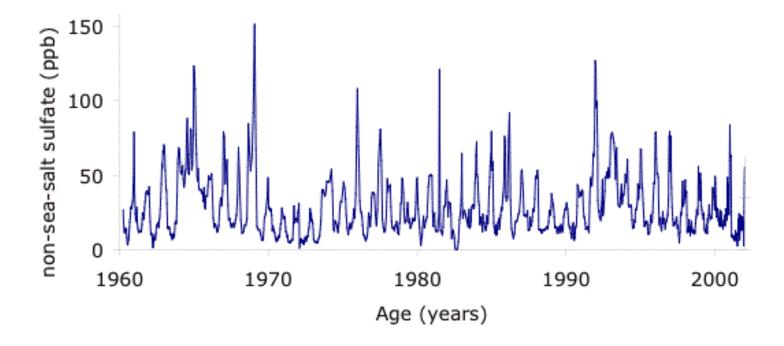




Electrical Conductivity Method (ECM)

Seasonal Cycles

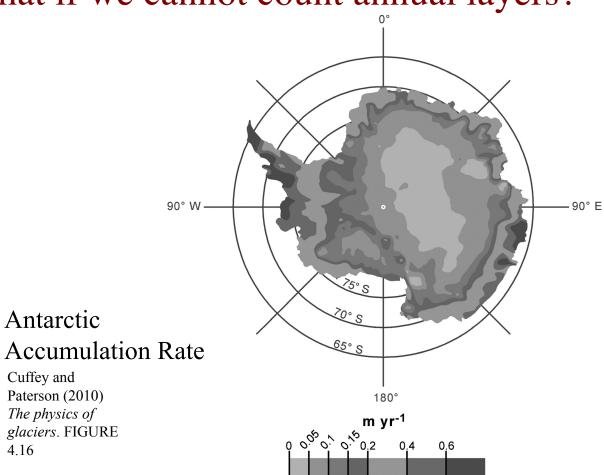
Spring plankton bloom in the Southern Ocean



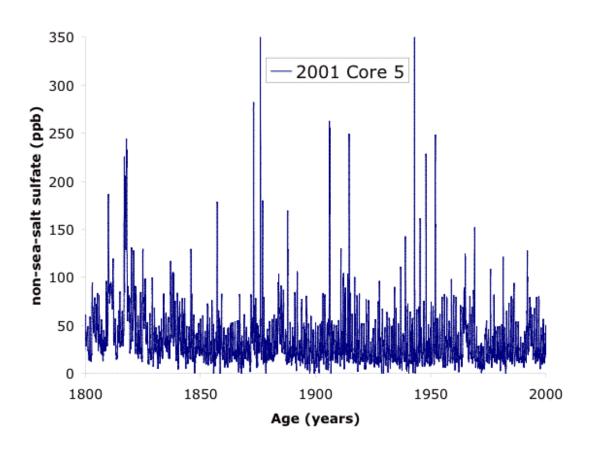
How high can we count?

- Annual layers in GISP2 ice core were counted back to ~50 ka
- accuracy estimated to be $\sim 1-2\%$
- This was an amazing achievement
- Current WAIS Divide ice core layer-counted to more than 30 ka

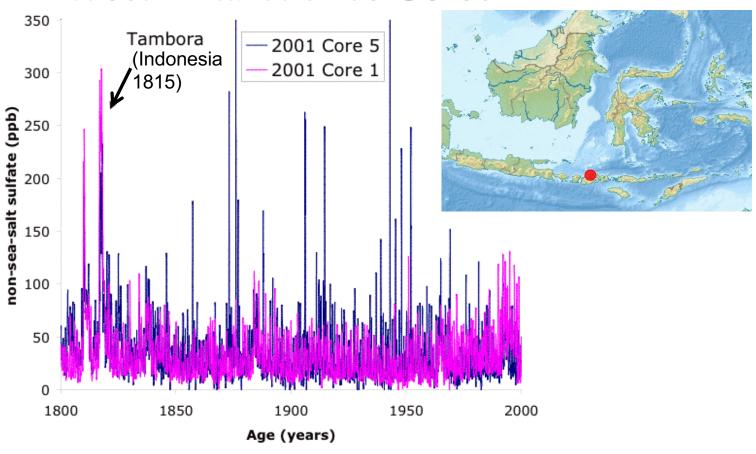
What if we cannot count annual layers?

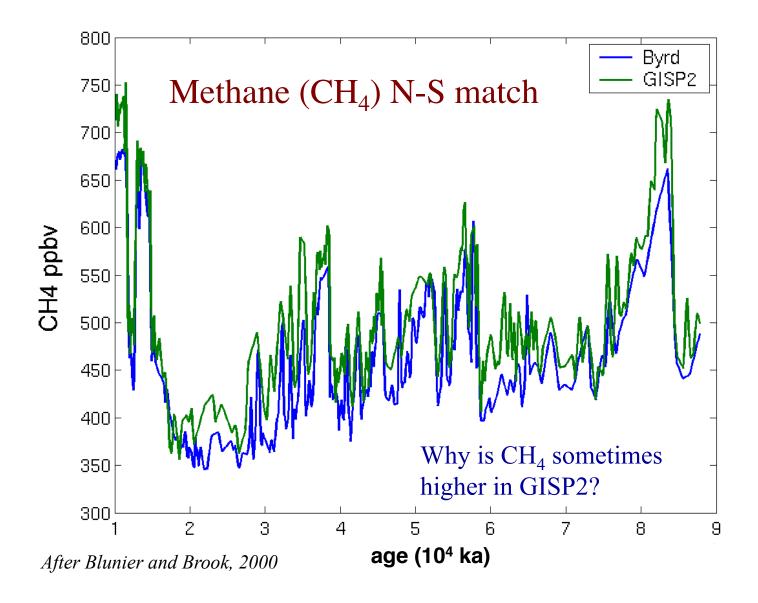


West Antarctic Ice Cores

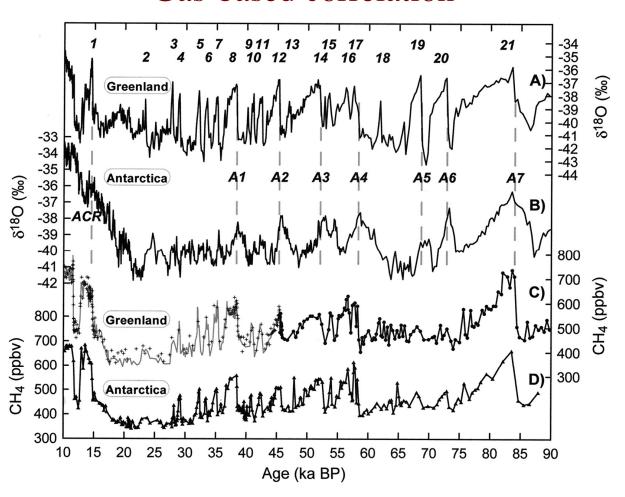


West Antarctic Ice Cores





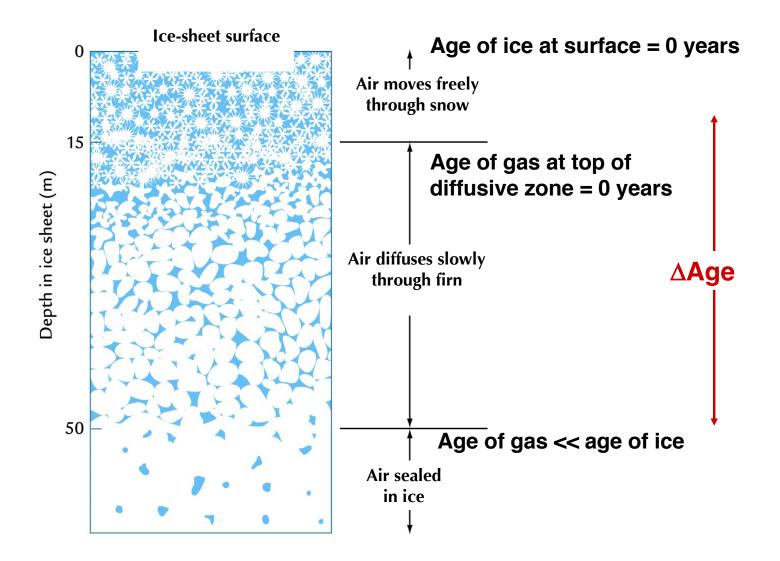
Gas-based correlation



Steps in gas-correlation dating

- From layer-counted ice core (A), calculate age of gas in the ice
- Match time series of globally mixed gases (CH₄, δ¹⁸O of O₂, CO₂) between cores A and B
- Determine age of ice in core B

Note that we must know \triangle age in both cores ...

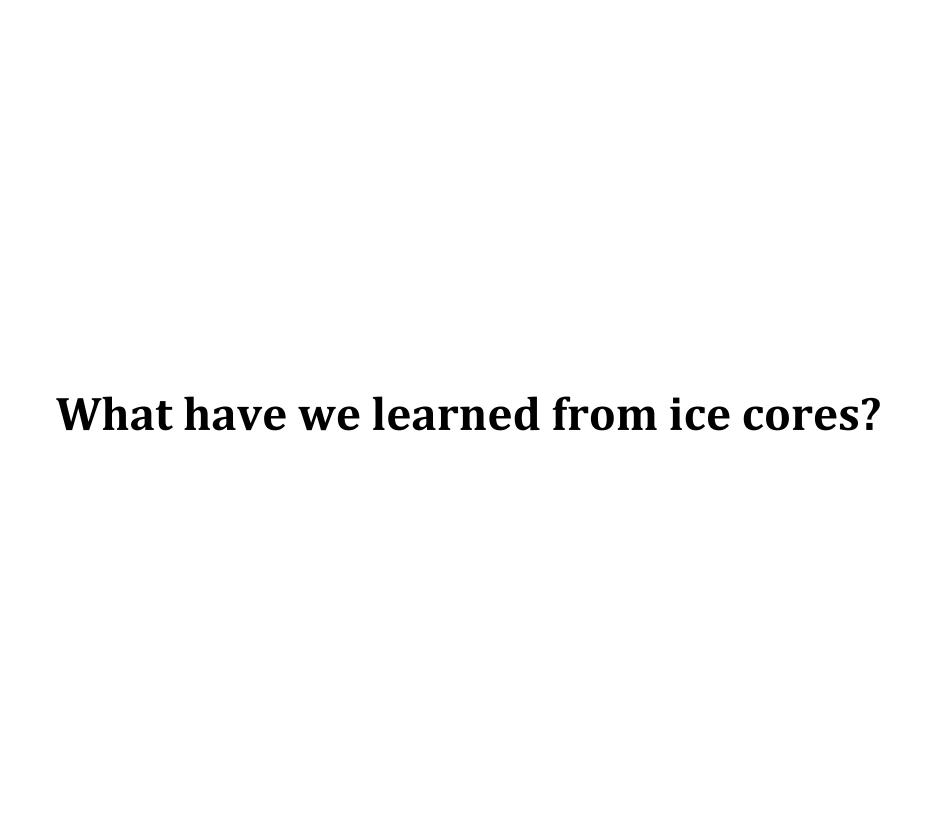


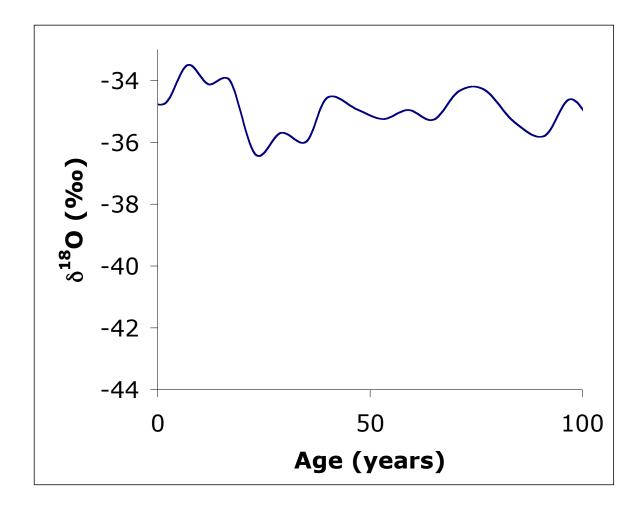
Calculating ∆age

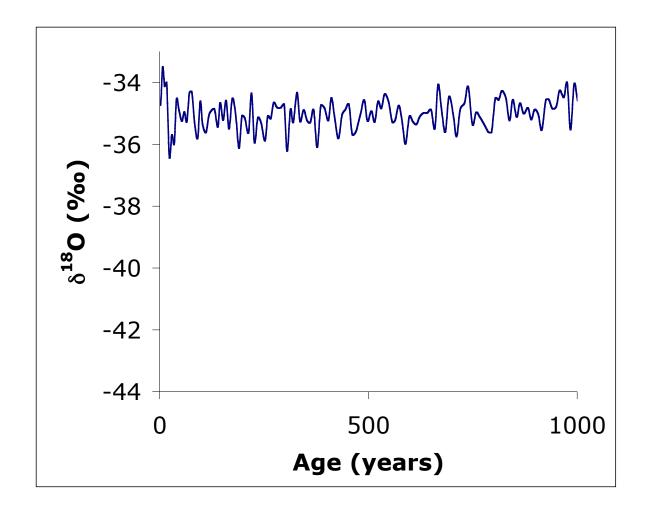
- lacktriangle Δ age is the age difference between the gas bubbles and the ice surrounding them.
- lacktriangle Δ age depends on the processes converting snow to firn and eventually ice, sealing off air bubbles
- These processes are dependent on **temperature** and **snow accumulation** rate; wind and other meteorological factors may also play an important role.

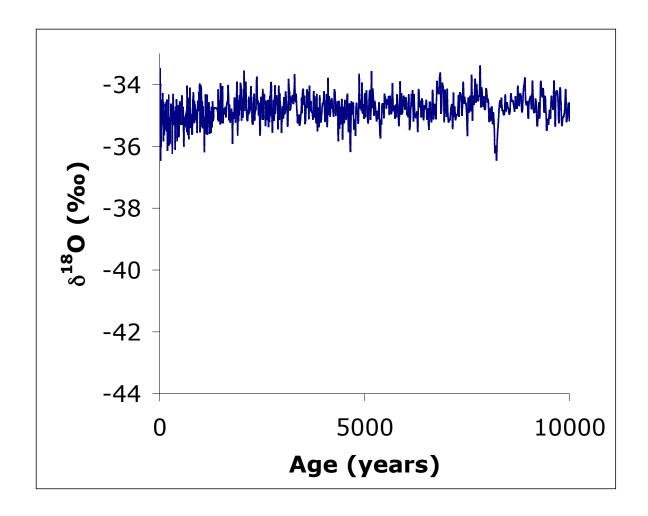
Uncertainties in Δ age

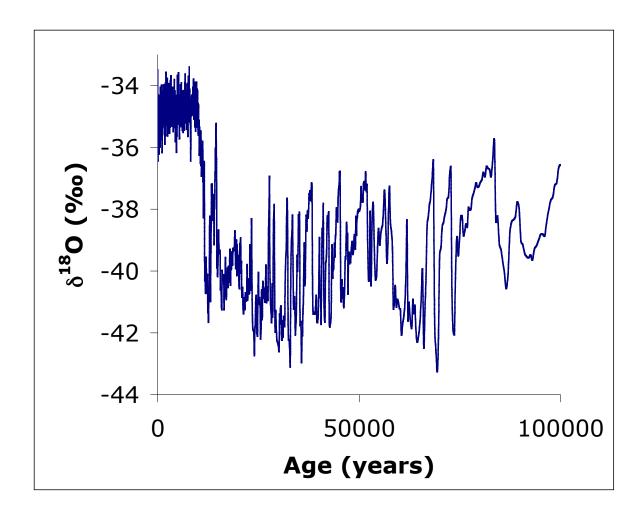
Item	How determined	How well known for sites with	
		low accumulation	high accumulation
Bubble close- off process	Empirical and theoretical models of firn densification (conversion of snow to ice)	poorly	well
Temperature history	Stable isotope profiles, borehole thermometry, gas isotopes (for selected time periods only)	moderately well, but variable	
Accumulation history	Inferred from geochemistry or isotopes	well	poorly

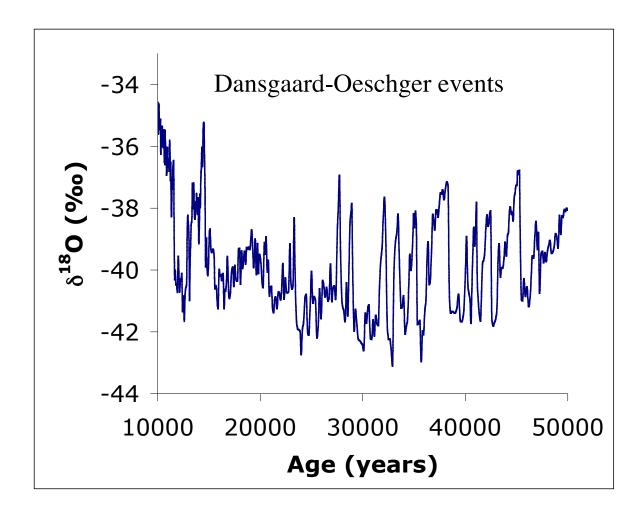




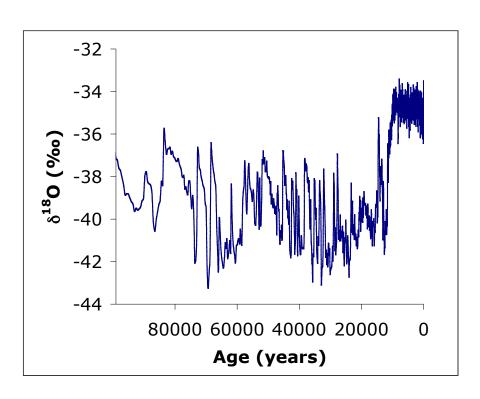


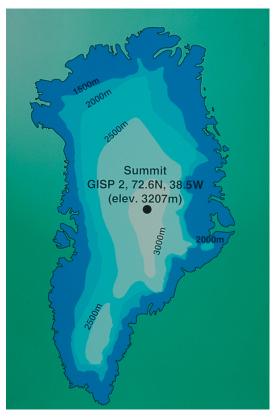




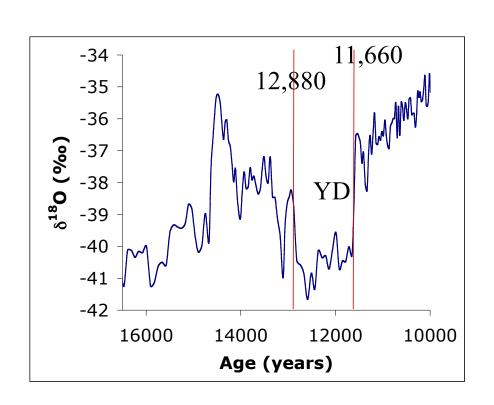


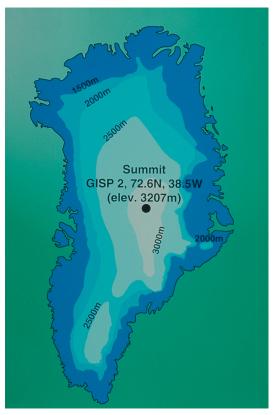
Central Greenland Ice Cores





Central Greenland Ice Cores





Temperature and Accumulation Rate

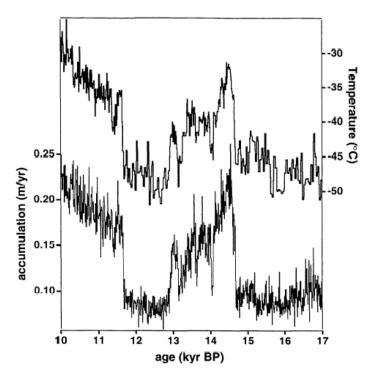
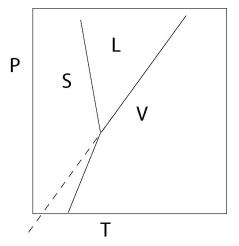


Figure 4. Comparison of the accumulation and δ records at the GISP2 site for the period between 18 and 10 kyr B. P. (adapted with permission from *Nature [Kapsner et al.*, 1995]; copyright Macmillan Magazines Limited), in using a calibration of 0.33%o/°C instead of 0.53%o/°C (see text).

Controlled by saturation vapor pressure?

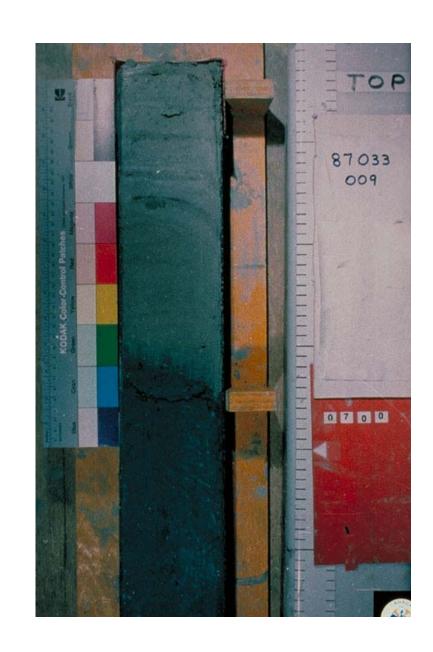
Remember the phase diagram for H₂0



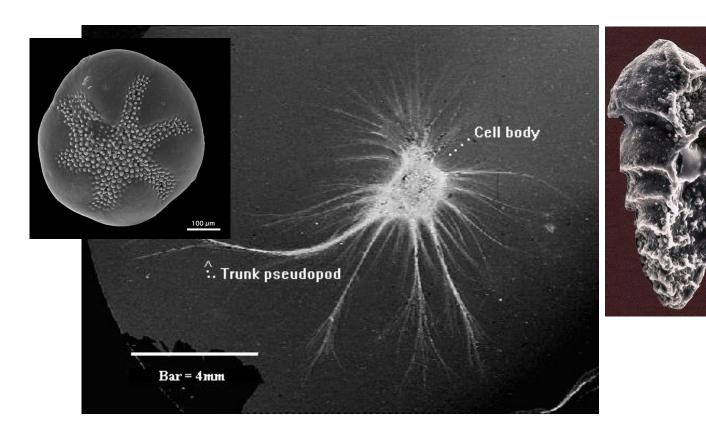
δ^{18} O in Ocean vs. Atmosphere

As the ice sheets build up during glaciations, they are preferentially made up of lighter isotopes.

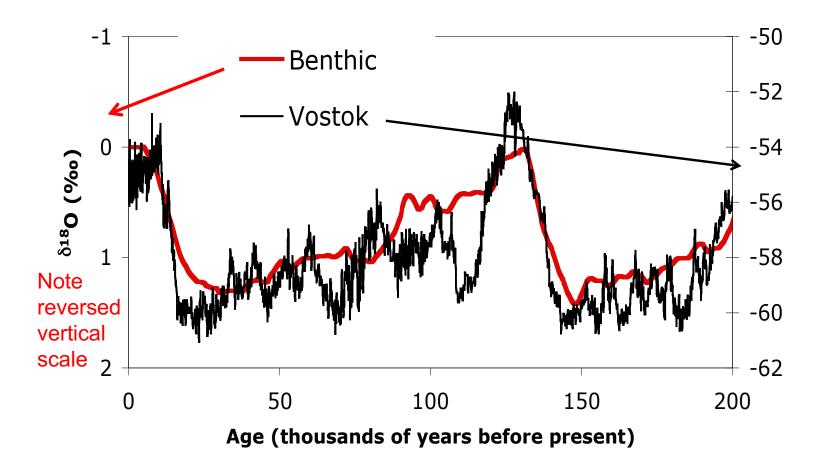
The consequence is that the ocean gets isotopically heavier as ice sheets grow.

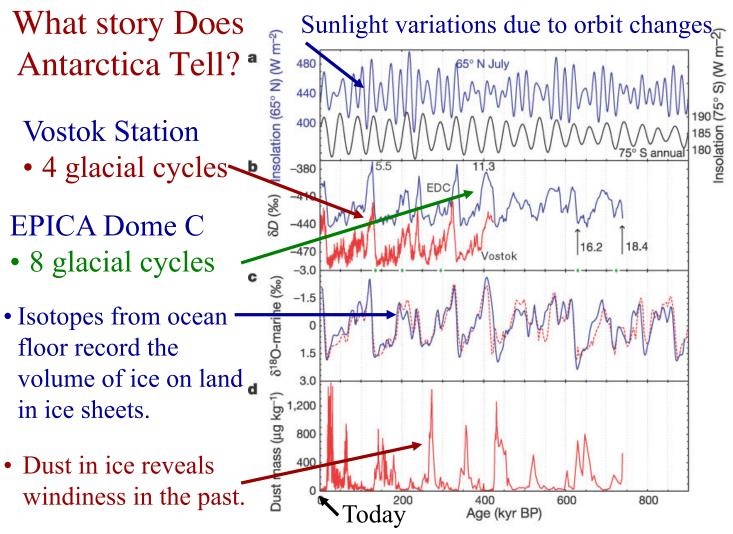


Planktonic and Benthic Foraminifera



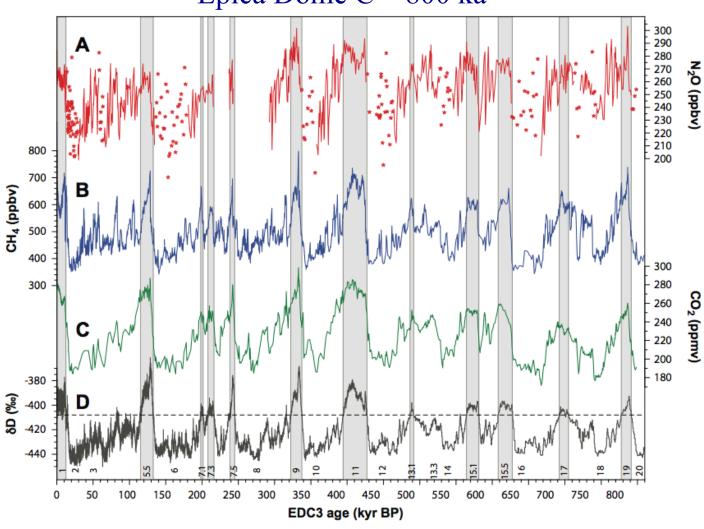
$$CaC^{16}O_3 + H_2^{18}O \leftarrow \frac{equilibrium}{} CaC^{16}O_2^{18}O + H_2^{16}O$$





Nature 429, 623-628 (10 June 2004) Eight glacial cycles from an Antarctic ice core

Epica Dome C – 800 ka



Rare isotopes in Air

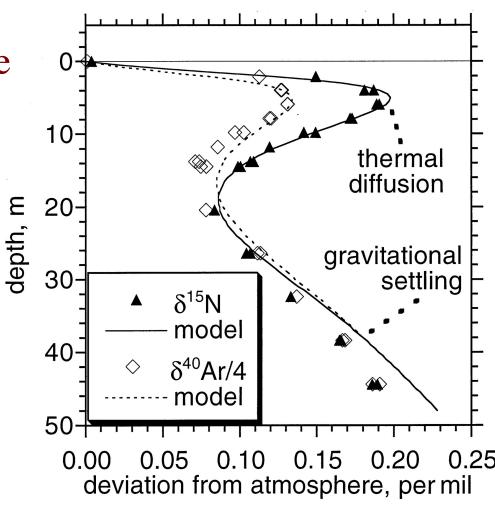
Air contains N₂ O₂, and Ar

- These gases contain atoms with rare but stable isotopes ¹⁵N (vs ¹⁴N), ³⁶Ar (vs ⁴⁰Ar), and ¹⁸O (vs ¹⁶O)
- Content of ¹⁵N and ⁴⁰Ar can be expressed in δ notation δ ¹⁵N and δ ⁴⁰Ar, just like δ ¹⁸O
- Gas with these rare isotopes behaves slightly differently to normal gas when temperature changes.
 In still air in pore spaces, gas with a heavy isotope tends to:
 - Settle to bottom
 - Move toward colder places

δ^{15} N and fast climate change

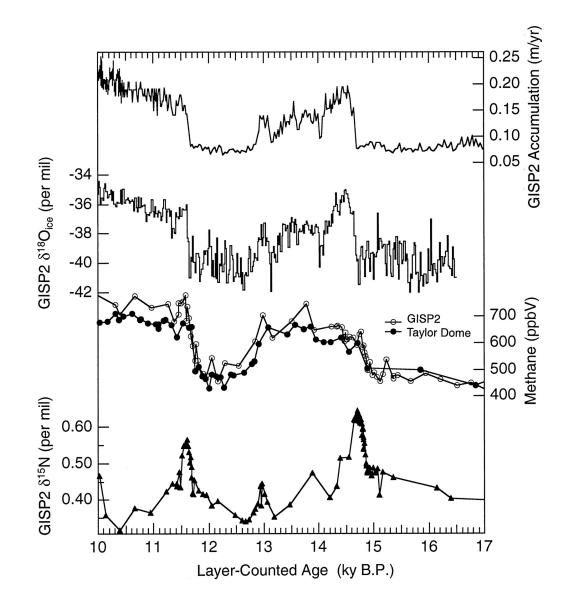
N₂ gas is largest component of atmosphere. In still air in pore spaces, gas with heavy isotope e.g. ¹⁵N or ⁴⁰Ar tends to

- Settle to bottom
- Move toward cold
- Why are δ^{15} N and δ^{40} Ar so positive at 8 meters depth?



How large were fast temperature changes?

Size of the $\delta^{15}N$ spikes is related to the size of the fast shift in temperature



Nature 484, 5 April 2012

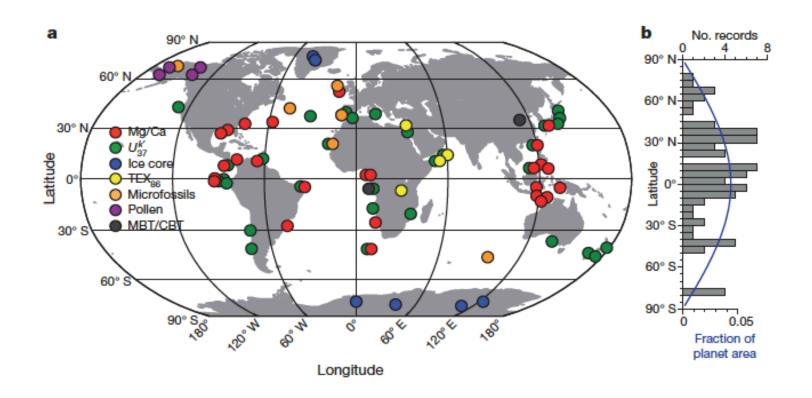
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doi:10.1038/nature10915

Global warming preceded by increasing carbon dioxide concentrations during the last deglaciation

Jeremy D. Shakun^{1,2}, Peter U. Clark³, Feng He⁴, Shaun A. Marcott³, Alan C. Mix³, Zhengyu Liu^{4,5,6}, Bette Otto-Bliesner⁷, Andreas Schmittner³ & Edouard Bard⁸

Sites with climate records – (temperature)



Shakun et al. Nature 484 5 April 2012.

