

Density of Sea Water ρ

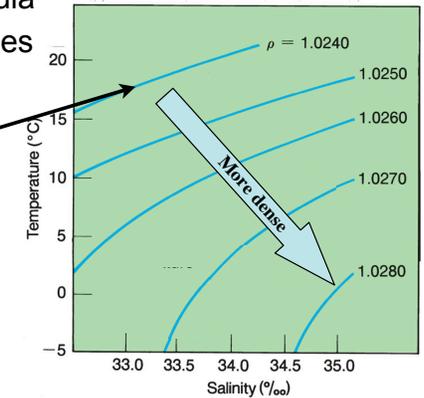


- Definition: mass of substance per unit volume
 - Grams per cm^3 (=cc, =ml)
- ρ of pure water at 4°C = 1.0 g/cm^3
- Salts make water more dense
 - Salinity = grams salts per kilogram water
 - = parts per thousand or ‰
 - $1 \text{ g/kg} = 0.1 \%$
 - In 35 g/kg seawater (at 4°C) density = 1.028
- Temperature also affects density
 - Warm water expands, density decreases
 - Cold water contracts, density increases

T-S Diagrams



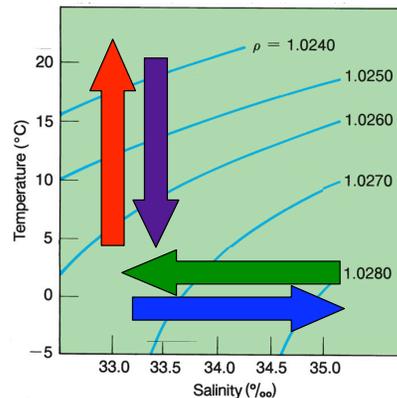
- Density is calculated from T & S
 - Uses a complex formula
 - Results printed in tables
 - Easier to use is a T-S diagram
 - Isopycnals = lines of constant density
 - Density increases from upper left to lower right



T-S Diagrams



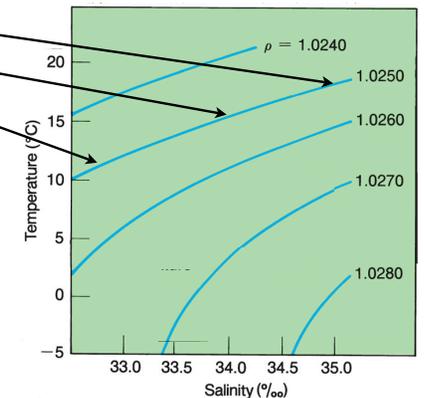
- T&S have opposite effects on ρ
 - $\uparrow T = \downarrow \rho$
 - but $\uparrow S = \uparrow \rho$
 - $\downarrow S = \downarrow \rho$
 - but $\downarrow T = \uparrow \rho$



T-S Diagrams



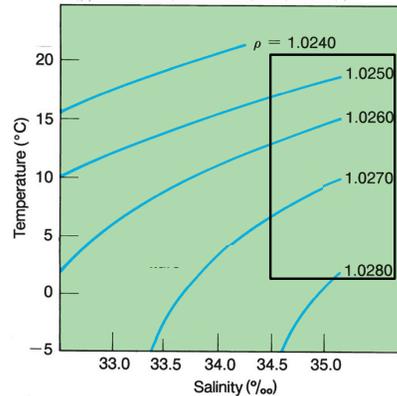
- Many combinations of T&S give the same ρ
 - $\rho = 1.025$
 - $T = 18, S = 35$
 - $T = 15, S = 33.9$
 - $T = 10, S = 32.5$
 - Changes in T&S counteract



T-S Diagrams



- Small changes in ρ very important
 - Most salinities $34.5 < S < 35.5$
 - Most temps $3^\circ < T < 20^\circ\text{C}$
 - Most densities $1.025 < \rho < 1.028$



Density of Sea Water

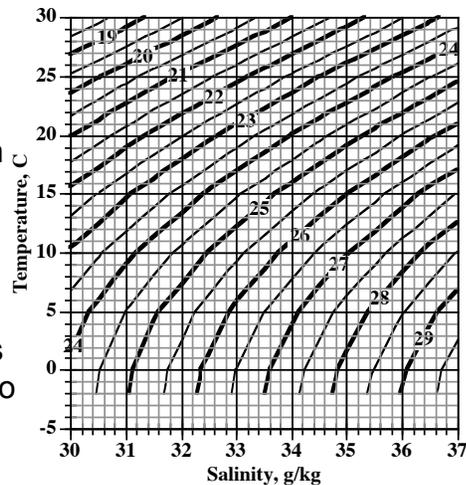


- Sigma-t (σ_t) is an abbreviation or shorthand for density.
 - $(\rho - 1) * 1000$.
 - $\rho = 1.025, \sigma_t = 25.0$; $\rho = 1.028, \sigma_t = 28.0$
- Used because small differences in density have important effects on water movement.

Determining Density



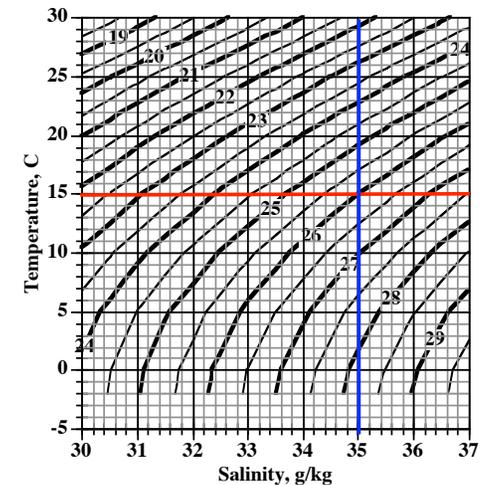
- T-S diagram a graphical display of σ_t values
 - Read directly from T & S
 - Simpler than solving the formulas
 - Today instruments are programmed to make calculations automatically



Determining Density



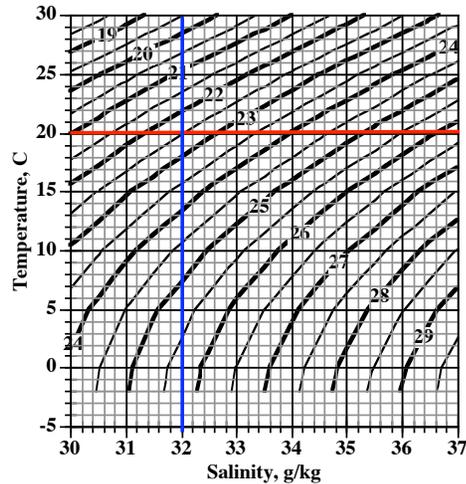
- Example #1
 - $T = 15^\circ\text{C}$
 - $S = 35\text{ g/kg}$
 - $\sigma_t = ?$
 - 26
 - $\rho = ?$
 - 1.026 g/cm^3



Determining Density



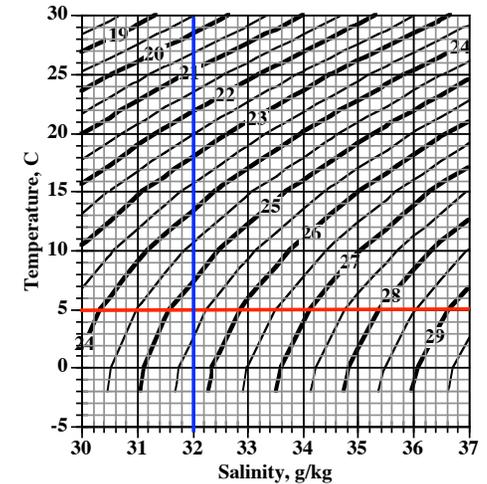
- Example #2
 - $T = 20^\circ \text{C}$
 - $S = 32 \text{ g/kg}$
 - $\sigma_t = ?$
 - 22.5
 - $\rho = ?$
 - 1.0225 g/cm^3



Determining Density



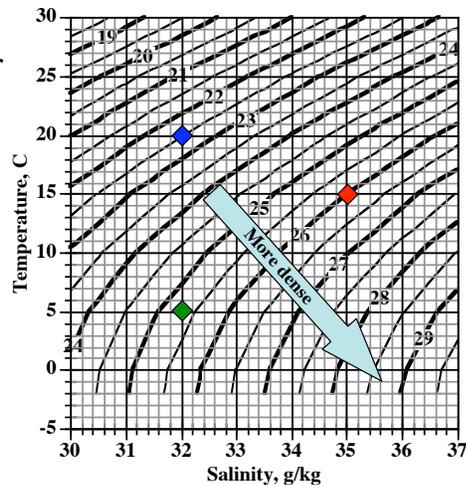
- Example #3
 - $T = 5^\circ \text{C}$
 - $S = 32 \text{ g/kg}$
 - $\sigma_t = ?$
 - 25.3
 - $\rho = ?$
 - 1.0253 g/cm^3



Determining Density



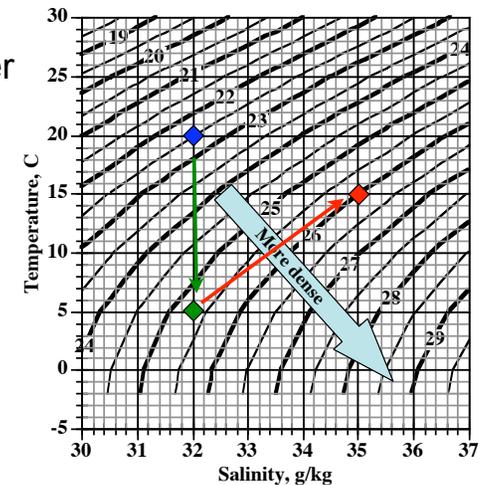
- Rank the examples in order of increasing density
 - #2 $\sigma_t = 22.5$
 - $\rho = 1.0225 \text{ g/cm}^3$
 - #3 $\sigma_t = 25.3$
 - $\rho = 1.0253 \text{ g/cm}^3$
 - #1 $\sigma_t = 26.0$
 - $\rho = 1.026 \text{ g/cm}^3$



Determining Density



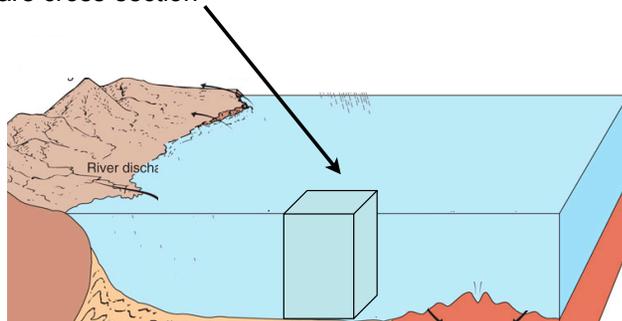
- Rank the examples in order of increasing density
 - #2 $\sigma_t = 22.5$
 - $\downarrow T @ \text{ same } S$
 - = $\uparrow \rho$
 - #3 $\sigma_t = 25.3$
 - $\uparrow S \text{ overcomes } \uparrow T = \uparrow \rho$
 - #1 $\sigma_t = 26.0$



Vertical Stratification



- Water column
 - Hypothetical vertical section of water from surface to bottom
 - Square cross-section

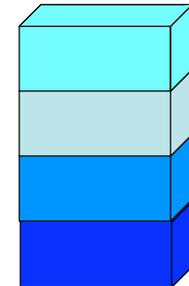


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Vertical Stratification



- Stratification is the vertical density layering of water column
 - Waters of different density at different depths
 - Buoyancy — less dense water tends to float atop more dense
- Density stratification
 - Warmer water floats & cooler water sinks (constant salinity)
 - Fresher water floats & saltier water sinks (constant temperature)

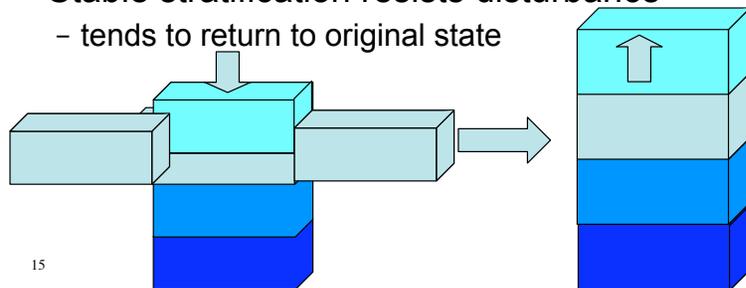


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Vertical Stability



- Stability of stratified water column
 - Depends on relative density of layers
 - Less dense water atop more dense water = stable
 - It will persist until disturbed
- Stable stratification resists disturbance
 - tends to return to original state

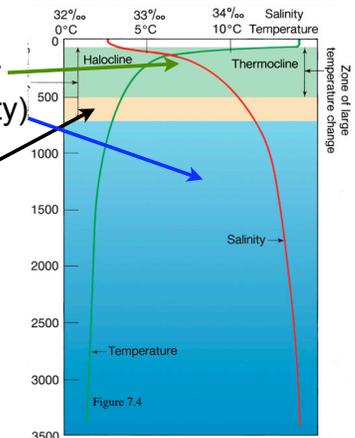


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Examples: Stability



- Thermal—Lake Washington
 - Surface warms in summer
 - Warm (lower-density) water lies atop cool (higher-density) water
 - Thermocline: boundary between layers of different temperature
 - Rapid change in temperature with depth



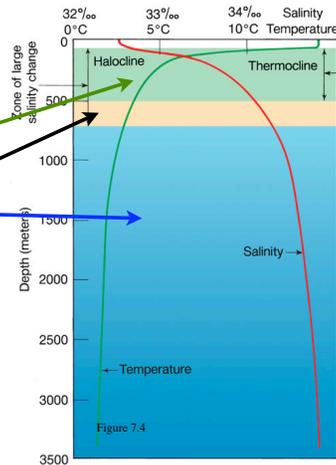
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Examples: Stability



• Haline—Puget Sound

- River runoff meets sea water at the river mouths
- Fresh or brackish (low-salinity) water lies atop higher-salinity water
- Halocline: boundary between layers of different salinity
 - Rapid change in salinity with depth



Examples: Stability

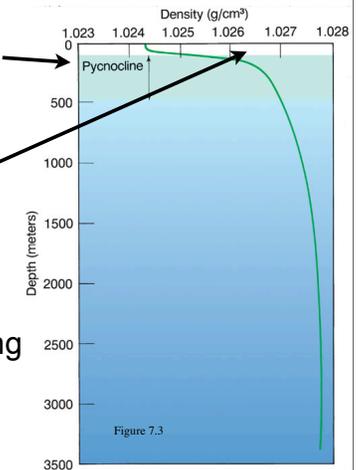


• Pycnocline

- Boundary between 2 layers of different density
- If there is a thermocline or halocline, there is also a pycnocline.

• Mixed layer

- Layer above the pycnocline
- Homogenized by wind mixing
- Nearly uniform properties over depth.



Vertical Stability

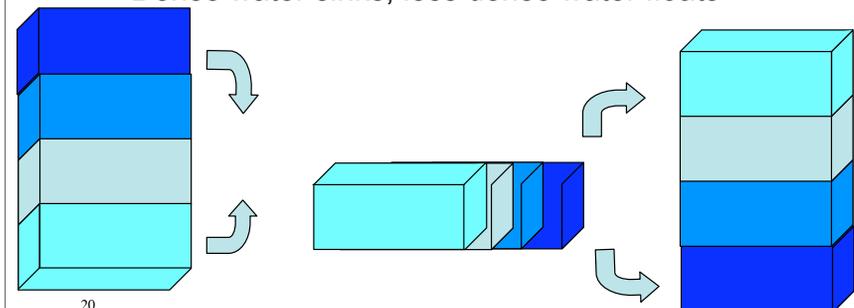


- In general, the oceans are stable
 - Greater density difference between layers = stronger stability
- If not, they would move until stable
 - Vertical instability occurs in certain situations in the oceans
- Neutral stability = unstratified
 - Density is same at all levels
- In general, T has greater effects than S
 - Some important exceptions
 - Puget Sound, Mediterranean

Instability



- Instability of stratified water column
 - More dense water atop less dense
 - Initiates rather than resists motion
 - Dense water sinks, less dense water floats



Examples: Instability



- Thermal—Polar seas in winter
 - Strong cooling
 - Dense water at the surface
 - Sinks below warmer water beneath
 - Major factor in global density-driven ocean currents
- Haline—Polar seas in winter
 - Sea ice freezing & “brine exclusion”
 - Sea ice is almost pure fresh water
 - Salt remains in sea water, raising salinity & density
 - Sinks below less-saline water beneath.

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Examples: Instability



- Haline—Mediterranean
 - Evaporation in desert climate
 - Creates high-salinity surface layer
 - Denser than the lower-salinity water beneath, and so it sinks.

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Stratifying processes



- What external natural processes enhance stratification?
 - Anything that changes density
 - Heating & cooling
 - Freshwater runoff
 - Evaporation & precipitation
 - Freezing & melting of sea ice

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Stabilizing processes



- What external natural processes enhance vertical stability?
 - Surface solar heating (T)
 - Freshwater runoff (S)
 - Rain (S)
 - Melting of sea ice (S)

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Destabilizing processes



- What external natural processes enhance vertical instability?
 - Surface cooling (T)
 - Sea ice formation (T & S)
 - Surface evaporation (S)
 - Sea-floor heating (T)

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Destratifying forces



- What external natural processes break down vertical stratification?
 - Forces that move water rather than exchanging heat or fresh water
 - Wind mixing
 - Fast, turbulent currents (esp. over or around bathymetric barriers)
 - Instability-induced convection

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The Real Ocean



- Vertical changes in both T & S
 - If both T & S increase, what happens to density & stability?
 - If both T & S decrease, what happens to density & stability?
- Must determine density from T & S to answer this question

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The Real Ocean



- Curious example—the Red Sea
 - Surface = 30°C, 42.5 g/kg
 - Heating & evaporation
 - Bottom = 36°C & 257 g/kg
 - Rift valley & hydrothermal vents
- Vertically stable or unstable?
 - Stable - S overcomes T
- Puget Sound in winter
 - Surface = 4°C, 20 g/kg
 - Deep = 8°C, 31 g/kg
 - Stable because low surface S overcomes low T

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